

UNIVERSIDADE DE SÃO PAULO
FACULDADE DE ODONTOLOGIA DE BAURU

KARLA DRUZIAN OLIVEIRA

**Impact of the incorporation of 3Y-TZP nanoparticles and
different sintering methods in bovine hydroxyapatite:
development, microstructural characterization, and
mechanical properties**

**Impacto da incorporação de nanopartículas de 3Y-TZP e
métodos distintos de sinterização em hidroxiapatita
bovina: desenvolvimento, caracterização microestrutural e
mecânica**

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mecânica**

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Orientador: Prof. Dr. José Henrique Rubo
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DEDICATÓRIA

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*“O mais importante é nunca parar de questionar.
A curiosidade tem uma razão para existir. Nunca
perca sua sagrada curiosidade.”*

Albert Einstein

RESUMO

O objetivo deste estudo foi caracterizar microestruturalmente e mecanicamente uma cerâmica experimental densa de hidroxiapatita bovina na sua forma pura (HA) e com adição de 10% de 3Y-TZP (HA+10%3 Y-TZP), sinterizada por diferentes metodologias, visando sua aplicação em restaurações e próteses dentárias fixas. Discos ($\varnothing 14 \times \pm 1.2\text{mm}$, ISO 6872) foram obtidos por prensagem uniaxial e isostática, seguidos de sinterização em curvas convencionais e em dois passos. Foram estabelecidos dez grupos experimentais, incluindo HA pura (1300C, 1450C, 1500C, 95%2S e 105%2S) e HA+10% 3Y-TZP (1300C, 1450C, 1500C, 95%2S e 105%2S). As amostras foram submetidas a análises de dureza Vickers (DV), resistência à flexão biaxial (RFB), difração de raios X (DRX) e microscopia eletrônica de varredura (MEV). Para DV, os dados foram submetidos a ANOVA, seguido de pós teste Tukey ($p<0.05$) e para RFB, Kruskal-Wallis seguido de Dwass-Steel-Critchlow-Fligner ($p<0.05$). Os grupos 1300C (409 ± 29.9) de HA pura e 105%2S (98.6 ± 8.72) de HA+10% 3Y-TZP apresentaram melhores resultados de DV. Em RFB, os resultados da mediana, primeiro e terceiro quartis que indicaram valores mais elevados foram, respectivamente, para os grupos de HA pura 1300C (112; 106, 120 MPa) e a 105%2S (103; 90.4, 118 MPa), estatisticamente semelhantes, e para o grupo de HA+10% 3Y-TZP a 105%2S (46.2; 43.9, 47.0). Na análise de DRX, foi identificada formação de fases secundárias em praticamente todos os grupos, com exceção de 1300C e 95%2S, ambos de HA pura. Os grupos com adição de nanopartículas mostraram formação de maior quantidade de poros ao longo das superfícies, conforme imagens em MEV. Assim, a modificação da metodologia convencional de sinterização (105%2S) foi importante para a otimização das propriedades mecânicas. No entanto, a hidroxiapatita pura, sem adição de nanopartículas, apresentou melhores resultados.

Palavras-chave: apatitas; cerâmica; nanotecnologia.

ABSTRACT

Impact of the incorporation of 3Y-TZP nanoparticles and different sintering methods in bovine hydroxyapatite: development, microstructural characterization, and mechanical properties

The purpose of this study was to characterize microstructurally and mechanically an experimental dense ceramic of bovine hydroxyapatite (HA) in its pure form and with the addition of 10% 3Y-TZP (HA+10%3 Y-TZP), sintered by different methodologies, aiming at its application in fixed dental restorations and prostheses. Discs ($\varnothing 14 \times \pm 1.2\text{mm}$, ISO 6872) were obtained by uniaxial and isostatic pressing, followed by sintering in conventional and two-step curves. Ten experimental groups were established, including pure HA (1300C, 1450C, 1500C, 95%2S and 105%2S) and HA+10% 3Y-TZP (1300C, 1450C, 1500C, 95%2S and 105%2S). The samples underwent Vickers Hardness (DV), biaxial flexural strength (BFS), X-ray diffraction (XRD), and scanning electron microscopy (SEM) analyses. For DV, the data were submitted to ANOVA, followed by Tukey's post-test ($p<0.05$), and for RFB, Kruskal-Wallis followed by Dwass-Steel-Critchlow-Fligner ($p<0.05$). The 1300C (409 ± 29.9) pure HA and 105%2S (98.6 ± 8.72) HA+10% 3Y-TZP groups showed better microhardness results. In RFB, the results of the median, first, and third quartiles indicating higher values were, respectively, for the pure HA groups at 1300C (112; 106, 120 MPa) and 105%2S (103; 90.4, 118 MPa), statistically similar, and for the HA+10% 3Y-TZP group at 105%2S (46.2; 43.9, 47.0). Secondary phase formation was identified in practically all groups, except for 1300C and 95%2S, both of pure HA, in DRX analysis. Groups with nanoparticle addition showed a greater amount of pore formation along the surfaces, as observed in SEM images. Thus, modification of the conventional sintering methodology (105%2S) was important for optimizing mechanical properties. However, pure hydroxyapatite, without nanoparticle addition, showed better results.

Keywords: apatites; ceramics; nanotechnology.

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1 INTRODUÇÃO

O aumento da demanda por estética está impulsionando o desenvolvimento de novas tecnologias como alternativas para reabilitações, onde a cerâmica emerge como a base para materiais restauradores indiretos¹. Isso se deve às suas características de biocompatibilidade, alta estabilidade química e apelo estético². A biocompatibilidade é uma qualidade essencial dos materiais cerâmicos, garantindo que desempenhem suas funções sem causar danos ao indivíduo beneficiário³⁻⁵.

As cerâmicas começaram a ganhar mais destaque a partir da década de 1970 e podem ser divididas em bioinertes e bioativas. Um exemplo de cerâmica bioativa é a hidroxiapatita, um composto de fosfato de cálcio encontrado nos tecidos ósseos, proteínas e dentes, que possibilita ampla gama de aplicações na área médica e odontológica⁶⁻¹¹, graças às suas propriedades de osteocondução, biocompatibilidade, bioatividade e ausência de toxicidade¹²⁻¹⁶.

A hidroxiapatita (HA) pode ser obtida para a fabricação de cerâmicas densas de várias maneiras e a partir de diversas fontes biológicas¹⁷⁻¹⁹. Uma dessas abordagens é a extração a partir de ossos de origem bovina, que se destaca como uma metodologia promissora devido à sua capacidade de reduzir custos e impactos ambientais, aproveitando resíduos da indústria pecuária como matéria-prima^{20,21}. Durante o processo de calcinação, ocorre a eliminação de patógenos e componentes orgânicos, mitigando o risco de transmissão de doenças, como a Encefalopatia Espongiforme Bovina²²⁻²⁷.

Na Odontologia, pode ser utilizada para substituição e regeneração óssea guiada, na configuração densa ou porosa²⁸⁻³⁰. No entanto, apresenta resistência mecânica reduzida, a qual limita seu uso em aplicações que envolvem suporte de carga^{31,32}. Portanto, é contraindicada em situações em que são expostas a alta incidência de forças^{33,34} pois apresenta como fator limitante a baixa resistência à flexão e baixa tenacidade à fratura³⁵, em comparação com outros materiais restauradores.

Visando reduzir a tensão de trincas, impedindo a propagação, otimizando a durabilidade e comportamento mecânico do material, a adição de partículas e nanopartículas têm sido empregadas, proporcionando reforço utilizando processos

como crack bridging, crack deflection e crack bridging sliding³⁶⁻³⁹. Alumina, titânio, biovidro, mulita e zircônia são alguns exemplos de agentes que podem ser combinados com a HA para esse fim⁴⁰⁻⁴⁸.

A zircônia apresenta propriedades superiores de resistência ao desgaste, flexão e tenacidade à fratura em comparação às vitrocerâmicas^{49,50}. Além disso, ela pode ocorrer em diferentes formas cristalinas em faixas de temperatura distintas: (1) monoclinica, estável em temperatura ambiente até 1170°C, (2) tetragonal, de 1170°C a 2370°C, e (3) cúbica, a partir de 2370°C. A adição de estabilizantes permite a manutenção da fase tetragonal em temperatura ambiente, uma vez que essa fase alotrópica apresenta propriedades mecânicas superiores às outras.

Entre os óxidos estabilizantes disponíveis, o óxido de ítrio tem demonstrado resultados excelentes⁵¹⁻⁵³. A melhoria nas propriedades mecânicas decorre do mecanismo de tenacificação por transformação, que induz compressão nas trincas. Quando sob estresse, os grãos ao redor da ponta da trinca passam por uma transformação de fase da tetragonal para a monoclinica, resultando em um aumento de aproximadamente 3–5% no volume e reduzindo a propagação da trinca, o que melhora a tenacidade à fratura do material⁵²⁻⁵⁵.

Cerâmicas policristalinas são materiais cerâmicos inorgânicos não metálicos que não possuem fase vítreia. Constituídas por grãos separados por contornos de grão, esses materiais exibem uma disposição aleatória dos cristais. A zircônia policristalina tetragonal estabilizada por ítria (Y-TZP) pode ser classificada em gerações. A primeira geração, com a adição de 3% mol de óxido de ítrio (3Y-TZP), é reconhecida por suas excelentes propriedades mecânicas e alta opacidade, devido à birrefringência natural da fase tetragonal. Esses materiais foram adotados como alternativas às cerâmicas metálicas⁵⁵. Portanto, a combinação desses dois materiais, HA e 3Y-TZP, apresenta potencial significativo, uma vez que a bioatividade e biocompatibilidade da HA, combinadas à tenacidade à fratura da 3Y-TZP, oferecem características adequadas para bioaplicações⁵⁶⁻⁶¹.

A sinterização desempenha um papel crucial na densificação do material, afetando diretamente suas propriedades mecânicas finais, incluindo o tamanho dos grãos^{40,62}. Para aprimorar essas propriedades, novos métodos foram desenvolvidos, visando alcançar uma densificação completa da amostra e a manutenção de grãos menores.

Uma dessas técnicas, introduzida no início da década de 1990⁶³, é a sinterização em duas etapas, na qual o aquecimento das amostras é realizado em dois estágios. Essa abordagem, que faz uso de fornos convencionais, promove a formação de grãos uniformes, evitando seu crescimento acelerado e resultando em uma HA totalmente densa, com propriedades mecânicas aprimoradas⁶⁴⁻⁶⁷.

Na primeira etapa da sinterização, o material é aquecido até uma temperatura mais alta (T1) por um curto período, geralmente cerca de 1 minuto. Esse tempo é suficiente para atingir uma densidade intermediária, correspondente a pelo menos 75% da densidade teórica, seguido de um resfriamento rápido. Na segunda etapa, a temperatura é reduzida para um nível mais baixo (T2) e mantida por um período mais longo, permitindo a densificação total do material. Esse processo fornece a energia necessária para manter a difusão do contorno do grão sem promover a migração do mesmo^{66, 69}. Como resultado, ocorre uma estabilização da microestrutura inicialmente desenvolvida, impedindo o crescimento dos grãos ao mesmo tempo em que preserva os mecanismos de difusão envolvidos no processo de densificação.

A sinterização em duas etapas, com a primeira temperatura mais elevada, pode ser empregada na fabricação de materiais cerâmicos com propriedades aprimoradas, resultando em melhorias significativas nas propriedades mecânicas. Essa abordagem tem sido associada a maior resistência, inclusive ao envelhecimento hidrotérmico, quando comparada à sinterização convencional de cerâmicas 3Y-TZP⁷⁴. Além disso, alguns estudos têm demonstrado melhorias em algumas características da HA, como dureza e tenacidade à fratura, por meio dessa metodologia⁷⁵.

2 ARTIGO

The article presented in this Dissertation was written according to the *Dental Materials* instructions and guidelines for article submission.

Impact of the incorporation of 3Y-TZP nanoparticles and different sintering methods in bovine hydroxyapatite: development, microstructural characterization, and mechanical properties

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ABSTRACT

Objectives. Characterize microstructurally and mechanically an experimental dense ceramic bovine hydroxyapatite (HA) in its pure form and with the addition of 3Y-TZP (HA+10%3 Y-TZP), sintered by different methodologies, aiming at its application in fixed dental restorations and prostheses.

Methods. Discs ($\varnothing 14 \times \pm 1.2\text{mm}$, ISO 6872) were obtained by uniaxial and isostatic pressing, followed by sintering in conventional and two-step curves. Ten experimental groups were established, including pure HA (1300C, 1450C, 1500C, 95%2S and 105%2S) and HA+10% 3Y-TZP (1300C, 1450C, 1500C, 95%2S and 105%2S). The samples underwent Vickers Hardness (DV), biaxial flexural strength (BFS), X-ray diffraction (XRD), and scanning electron microscopy (SEM) analyses. For DV, the data were submitted to ANOVA, followed by Tukey's post-test ($p<0.05$), and for RFB, Kruskal-Wallis followed by Dwass-Steel-Critchlow-Fligner ($p<0.05$).

Results. The 1300C (409 ± 29.9) pure HA and 105%2S (98.6 ± 8.72) HA+10% 3Y-TZP groups showed better microhardness results. In RFB, the results of the median, first, and third quartiles indicating higher values were, respectively, for the pure HA groups at 1300C (112; 106, 120 MPa) and 105%2S (103; 90.4, 118 MPa), statistically similar, and for the HA+10% 3Y-TZP group at 105%2S (46.2; 43.9, 47.0). Secondary phase formation was identified in practically all groups, except for 1300C and 95%2S, both of pure HA, in DRX analysis. Groups with nanoparticle addition showed a greater amount of pore formation along the surfaces, as observed in SEM images.

Significance. Thus, modification of the conventional sintering methodology (105%2S) was important for optimizing mechanical properties. However, pure hydroxyapatite, without nanoparticle addition, showed better results.

Keywords: apatites; ceramics; nanotechnology.

1. INTRODUCTION

The growing interest in aesthetics drives the search for new technologies in rehabilitation, with ceramics standing out as a restorative material due to its biocompatibility, chemical stability, and aesthetic appearance [1,2]. Biocompatibility, essential for biomedical materials, ensures that no harm is caused to the organism [3-5]. Ceramics, classified as bioinert and bioactive, have gained popularity since the 1970s, with hydroxyapatite (HA) being an example of a bioactive ceramic with broad applications in medicine and dentistry due to its characteristics [6-16].

The obtaining of HA for the fabrication of dense ceramics can be carried out in various ways, including the extraction from bovine bones, a promising approach in terms of cost and sustainability [17-27]. Although HA is employed in dentistry for bone regeneration, its limited mechanical strength makes it unsuitable for load-bearing applications [28-35]. To enhance its mechanical properties and durability, it has been associated with particles and nanoparticles, such as zirconia [36-55].

Tetragonal zirconia stabilized with yttria (3Y-TZP) is widely used due to its excellent mechanical properties, serving as an alternative to metallic ceramics restorations [49-55]. The combination of HA and 3Y-TZP shows potential for biomedical applications, harnessing the bioactivity of HA and the fracture toughness of 3Y-TZP [56-61].

Two-step sintering is a methodology involving the heating of samples in two stages. Introduced in the early 1990s, it was developed to enhance the mechanical properties of ceramic materials such as HA and 3Y-TZP, providing complete densification and grain size control [62-67]. In the first stage of sintering, the material is heated to a higher temperature (T1) with a dwell time of 1 minute, which is sufficient to achieve an intermediate density (at least 75% of the theoretical density), followed by rapid cooling. In the second stage, the lower temperature (T2) is kept for a longer period until total densification provides energy to maintain grain boundary diffusion without grain migration [66, 69].

The purpose of this study was to characterize microstructurally and mechanically the effects of different sintering parameters on experimental bovine HA ceramics, both pure and with the addition of 10% 3Y-TZP nanoparticles. The goal was

to improve mechanical properties while restraining the decomposition of HA, by comparing two-step sintering methodology with conventional methods for both pure HA and HA with a significant proportion of added nanoparticles. The null hypothesis was that the two-step sintering process would provide no difference at microstructural and mechanical properties to sintering conventional method for dense bovine hydroxyapatite ceramics with 10% 3Y-TZP addition and pure HA.

2. MATERIAL AND METHODS

2.1. Sample preparation

In this study, the dense polycrystalline ceramic of HA + 10% 3Y-TZP was fabricated at the School of Engineering of São Carlos (Composite and Tribology Laboratory) and at the Integrated Research Center of the Bauru School of Dentistry (Biomaterials Laboratory II).

After approval by the Ethics Committee for Animal Use (CEUA/FOB/USP) under registration number 006/2023, Nelore bovine metatarsals were used as the source for obtaining HA. The bone tissue underwent thermochemical processes for complete removal of organic matter. This included immersion in hydrogen peroxide (H_2O_2) at 100 volumes for five minutes, heated to 100°C, followed by washing with deionized water. Subsequently, the material was placed in an oven at 900°C for calcination for 24 hours, with a heating rate of 5°C/min [16].

The remaining material underwent a reduction process using mortar and pestle during manual grinding. The bones were subjected to grinding in a rotary mill at a speed of 104 rpm for 48 hours and in a vibratory mill for 72 hours. In a polyethylene flask, with a height of 85 mm and a volume of 300 cm³, 40% volume (500 g) of 3-YTZP spheres with a diameter of 3 mm, 0.05% volume of para-aminobenzoic acid ($C_7H_7NO_2$), 30% volume of HA, and 69.95% volume of isopropyl alcohol (C_3H_8O) were added. As a result, a nanometric (40 nm) and uniform powder is obtained, which reduces the porosity of the ceramic, sintering time and temperature, and increases reactivity between particles, without altering the stoichiometry, morphology, and purity of the HA [16,68].

Nanoparticles of 3Y-TZP (Tosoh® Corporation Tokyo, Japan) were weighed at a concentration of 10% relative to the volume of HA and added to this mixture, which was then returned to the vibratory mill for 24 hours for homogenization. The flask was emptied, and the mixture was dried with a hot air blower at 80 °C and subsequently granulated into smaller particles on stainless steel meshes (#100 ≤ 150 µm; #150 mesh ≤ 106 µm; #200 mesh≤75 µm) [16,68].

Following ISO 6872 [70], specimens were produced in the form of discs (Table 1). The pure HA and the mixture (HA + 10% 3Y-TZP) in powder form were weighed at 0.7 g and inserted into a cylindrical matrix, followed by uniaxial pressing at 100 MPa for 1 minute. Subsequently, using elastomeric balloons under vacuum, they were pressed by an isostatic press at 200 MPa for 1 minute [47].

Table 1 - Density and proportion of components used in specimens' preparation

Component	Density (g/cm3)	Proportion (%)	Function
HA	3.14	30%	Ceramic powder
Para-aminobenzoic acid	1.37	0.05%	Deflocculant
Isopropyl alcohol	0.78	60,95%	Solvent
3-YTZP nanoparticles	6	10%	Reinforcement

2.2. Sintering

The sintering process was carried out in a Lindberg/Blue M chamber furnace (Asheville, NC, USA) under atmospheric pressure. After sintering, the specimens reached final dimensions of 14 ± 2 mm in diameter and 1.2 ± 0.2 mm in height, following ISO 6872:2015 standards [70]. The study design is illustrated in Figure 1.

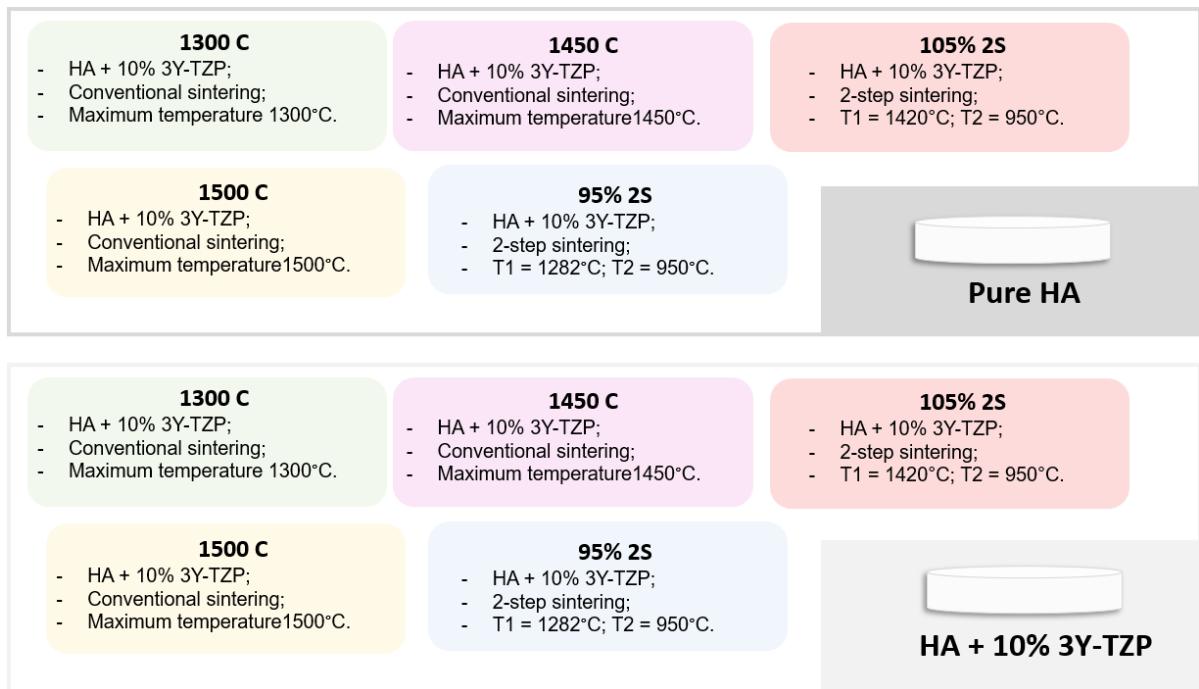


Figure 1. Study design.

In the conventional method, temperature was gradually increased from room temperature to 160°C at a heating rate of 2.7°C/min; from 160°C to 600°C at 4°C/min; from 600°C to 1100°C at 5°C/min; and finally, from 1100°C to the determined temperature for each group (1300°C, 1450°C, or 1500°C), at 6°C/min. The specimens were held at this maximum temperature for 120 minutes, followed by furnace cooling to room temperature (Fig. 3).

In the two-stage sintering process, the temperature for the first sintering stage (T1) was determined based on the temperature at which the material reaches critical density without grain growth [63, 67]. The sintering process begins with a gradual temperature increase from room temperature to 650°C at a heating rate of 4°C/min, then up to 900°C at 5°C/min, reaching 95% (1282°C) or 105% (1420°C) of 1350°C at 10°C/min, with a 1-minute plateau at this temperature (T1), followed by cooling to 950°C (T2) at 10°C/min, with a 20-hour plateau, and then slow furnace cooling to room temperature (Figure 2).

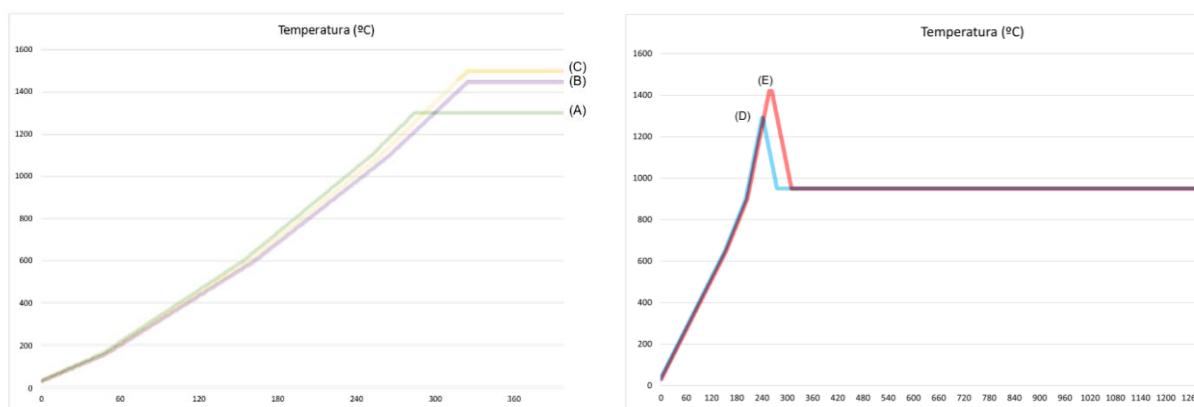


Figure 2. Sintering curves for the groups: 1300C (A); 1450C (B); 1500C (C); 95%2S (D); and 105%2S (E).

2.3. Mechanical properties

2.3.1. Vickers Microhardness

A Wilson Tukon 1202 machine from Buehler, Illinois, USA, equipped with a Vickers indenter was utilized. Five specimens from each group were evaluated, with three indentations performed on each disk. The microhardness tester was programmed to apply a load of 500 gf for 15 seconds per indentation. The Buehler Omnimet MHT software was employed for data analysis, and an average hardness value was calculated based on the obtained data.

2.3.2. Biaxial Flexural Strength (BFS)

The biaxial flexural strength (BFS) test was conducted using the device (OD36, Odeme Dental Research, Luzerna, Brazil) on a universal testing machine (Sintec 5G, MTS). Ten disk-shaped specimens from each group were analyzed following the recommendations of ISO 6872:2015 [69].

The specimens, cleaned with distilled water and isopropyl alcohol, were centrally positioned and supported by three steel spheres with a diameter of (4.5 ± 2) mm, placed 120° apart on a circular support with a diameter of (11 ± 1) mm. A load was applied through a flat pin with a diameter of (1.4 ± 0.2) mm at the center of the specimen at a speed of (1 ± 0.5) mm/min until fracture occurred. The obtained values

were collected in newtons (N). To calculate the biaxial flexural strength (MPa), the following eq.(1) was used:

$$\sigma = -0.23870 P(X-Y)/b^2 \quad (1)$$

σ is the maximum strength (MPa), P is the total fracture load (N), and b is the sample thickness (mm).

X and Y are values that were calculated from the eq.(2)(3):

$$X = (1 + v)\ln(r_2/r_3)^2 + [(1 - v)/2](r_2/r_3)^2 \quad (2)$$

$$Y = (1 + v) [1 + \ln(r_1/r_3)^2 + [(1 - v)(r_1/r_3)^2] \quad (3)$$

V corresponds to the value of the Poisson's ratio (if the value is unknown, 0.25 was used), r_1 is the radius of the support circle (mm), r_2 is the radius of the load application area, and r_3 is equivalent to the radius of the specimen.

2.4. Microstructural properties

2.4.1. X-Ray Diffraction (XRD)

For the identification of crystalline phases in the specimens under different sintering parameters, an X-ray diffractometer (PC Rigaku D/MAX 2500) with Cu radiation (kal) $\lambda=1.54060 \text{ \AA}$ was utilized, operating at 20 mA and 40 kV. For samples of HA + 10% 3Y-TZP, the Step Scan mode was employed with steps of 0.02° and a speed of 1 second per step, and an angular scan between 20° and 70° . For the samples with Pure HA, Continuous Scan mode was used with an angular scan between 10° and 70° . Rietveld refinement was performed using X'pert HighScore Plus software (Malvern Panalytical, Malvern, United Kingdom).

2.4.2. Scanning Electron Microscopy (SEM)

Five specimens were cleaned using an ultrasonic bath for 15 minutes immersed in distilled water, and then submerged for 15 minutes in 92.8% ethanol (Rioquímica, São José do Rio Preto, SP, Brazil). Subsequently, they were coated with

gold (Denton Vacuum, Moorestown, NJ, USA), and the analyses of the groups were performed using a Scanning Electron Microscope (ZEISS, Supra40, Jena, Germany). Measurements were taken on the fracture surfaces at magnifications of 500x, 1.000x, and 2.000x.

2.5. Statistical analysis

The results of Vickers Microhardness and Biaxial Flexural Strength were organized and subjected to the Shapiro-Wilk normality test. In the case of parametric distribution, the data were subjected to ANOVA followed by Tukey's post hoc test. For non-parametric distribution, they were subjected to Kruskal-Wallis followed by Dwass-Steel-Critchlow-Fligner post hoc test. In all cases, a significance level of 5% ($p<0.05$) was adopted. The datasets were analyzed using the statistical software Jamovi (Version 2.3.21).

3. RESULTS

3.1. Vickers Microhardness

The results from specimens composed of pure HA revealed no statistically significant difference between the 1300C (409 ± 29.9) and 105% 2S (403 ± 9.29) groups, which demonstrated superior performance compared to the others. However, groups characterized by conventional curves and higher maximum temperatures exhibited lower values, with the 1500C group (151 ± 47.1) showing the lowest result.

For groups containing HA + 10% 3Y-TZP, the microhardness values did not show statistically significant differences; however, it was observed that higher results were obtained in groups subjected to higher temperatures in both curves, namely 105% 2S (98.6 ± 8.72) and 1500C (94.1 ± 4.92). The microhardness data are presented in Table 2.

Table 2 - Vickers Microhardness divided by nanoparticle quantity and sintering methodology

	1300 C	1450 C	1500C	95% 2S	105% 2S
HA Pura	409± 29.9 Aa	198± 7.48 Ca	151± 47.1 Da	336± 14.6 Ba	403± 9.29 Aa
HA+10% 3Y-TZP	58.7± 3.66 Ab	82.5±3.20 Ab	94.1± 4.92 Ab	75.4± 3.14 Ab	98.6± 8.72 Ab

Different uppercase letters represent statistical differences compared within rows. Different lowercase letters represent statistical differences compared within columns ($p < 0.05$).

3.2. Biaxial Flexural Strength (BFS)

For pure HA, the 1300C (112; 106, 120 MPa) and 105% 2S (103; 90.4, 118 MPa) groups showed the best performance in terms of biaxial flexural strength, with no statistically significant difference between them. On the other hand, the 1500C group (26.4; 24.6, 28.0 MPa) exhibited the lowest result, differing significantly from the others ($p<0.05$).

Groups containing 10% addition of 3Y-TZP showed better results with two-stage sintering. The group with the highest value was the 105% 2S (46.2; 43.9, 47.0 MPa), followed by the 95% 2S group (38.5; 35.9, 40.4 MPa). Conventional sintering curves yielded lower results, with no statistically significant differences between them. The biaxial flexural strength data are presented in Table 3.

Table 3 - Biaxial Flexural Strength (MPa) of the different groups under analysis

	1300 C	1450 C	1500C	95% 2S	105% 2S
HA Pura	112(106-120) Aa	42(39.9-43.6) Ca	26.4(24.6-28.0) Da	57.5(53.0-61.3) Ba	103(90.4-118) Aa
HA+10% 3Y-TZP	34.9(33.3-36.5) Cb	32.4(31.3-32.6) Cb	33.5(31.2-34.4) Ca	38.5(35.9-40.4) Bb	46.2(43.9-47.0) Ab

Different uppercase letters represent statistical differences when comparing rows. Different lowercase letters represent statistical differences when comparing columns ($p<0.05$).

3.3. X-Ray Diffraction (XRD)

From the XRD analysis, in the groups of pure HA at 1300C and 95% 2S, it is possible to observe crystalline peaks corresponding to Hydroxyapatite (ICSD 26204). In the 105% 2S group, there is a significant amount of Hydroxyapatite (ICSD

26204) associated with a small portion of CaO (ICSD 14922). In the 1450C and 1500C groups, the predominance of HA peaks was not maintained, and the formation of secondary phases was observed. In the conventional curve with a maximum temperature of 1450°C, besides HA, peaks consistent with the crystallographic pattern of β-TCP (ICSD 6191) were identified (Table 4). However, in the conventional curve with a maximum temperature of 1500°C, complete HA degradation and the presence of only α-TCP (ICSD 923) and β-TCP (ICSD 6191) were observed (Figure 3).

Table 4 - Composition of the groups after Rietveld analysis

	1300 C	1450 C	1500C	95% 2S	105% 2S
HA Pura	100% HA	42.3% HA 57.7%B-TCP	54.9%β-TCP 45.1%α-TCP	100%HA	99.1%HA 0.9%CaO
HA + 10% 3Y-TZP	79.9% B-TCP 18.9%CaZrO ₃ 1.2% ZrO ₂	80.5%B-TCP 17.2%CaZrO ₃ 2.3%YSZ	78.3%β-TCP 21.7%CaZrO ₃	76.6%β-TCP 20.6%CaZrO ₃ 2.8%ZrO ₂	77.7%β-TCP 21.2%CaZrO ₃ 1.1%ZrO ₂

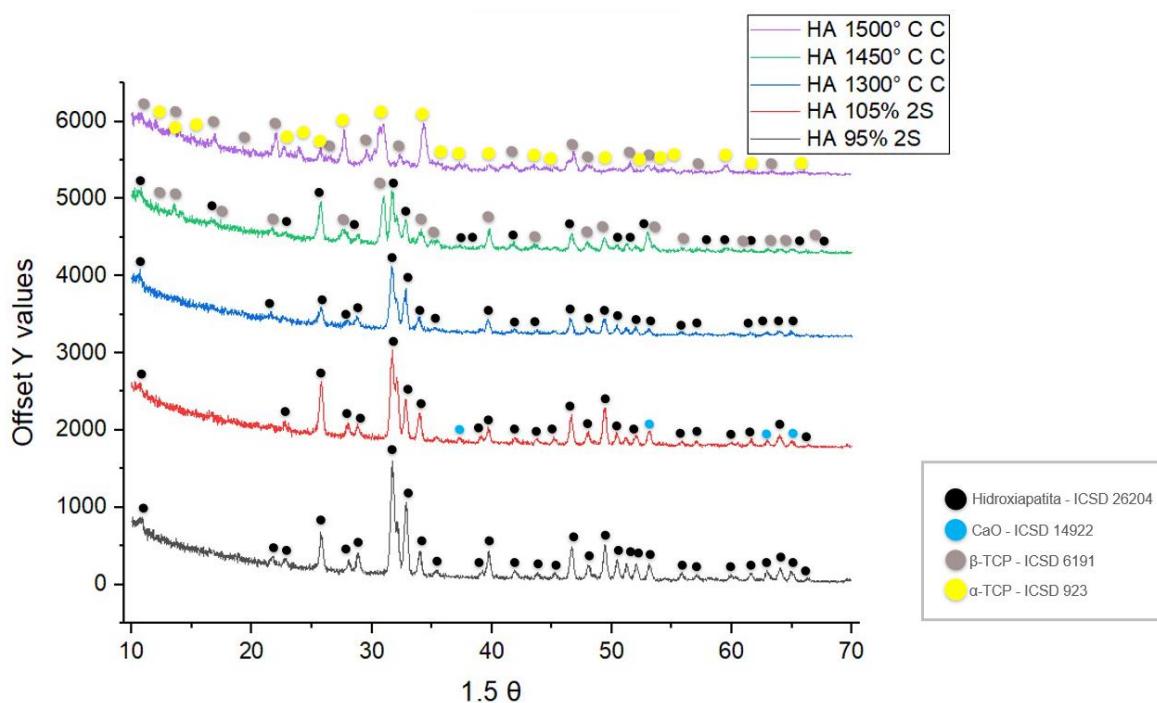


Figure 3 - X-Ray Diffraction (XRD) of samples containing pure HA.

In the HA + 10% 3Y-TZP groups, decomposition of HA, formation of secondary phases, and reaction between the components can be observed in all

groups (Table 4). The powder sample presents crystalline peaks corresponding to Hydroxyapatite (ICSD 26204) and YSZ (ICSD 75309). After sintering, regardless of the curve used, all groups showed a crystalline pattern of β -TCP (ICSD 6191) and CaZrO₃ (ICSD 97463). In the case of ZrO₂ (ICSD 165037), among the sintered groups, the corresponding crystalline peaks were not identified only in the 1500C group (Figure 4).

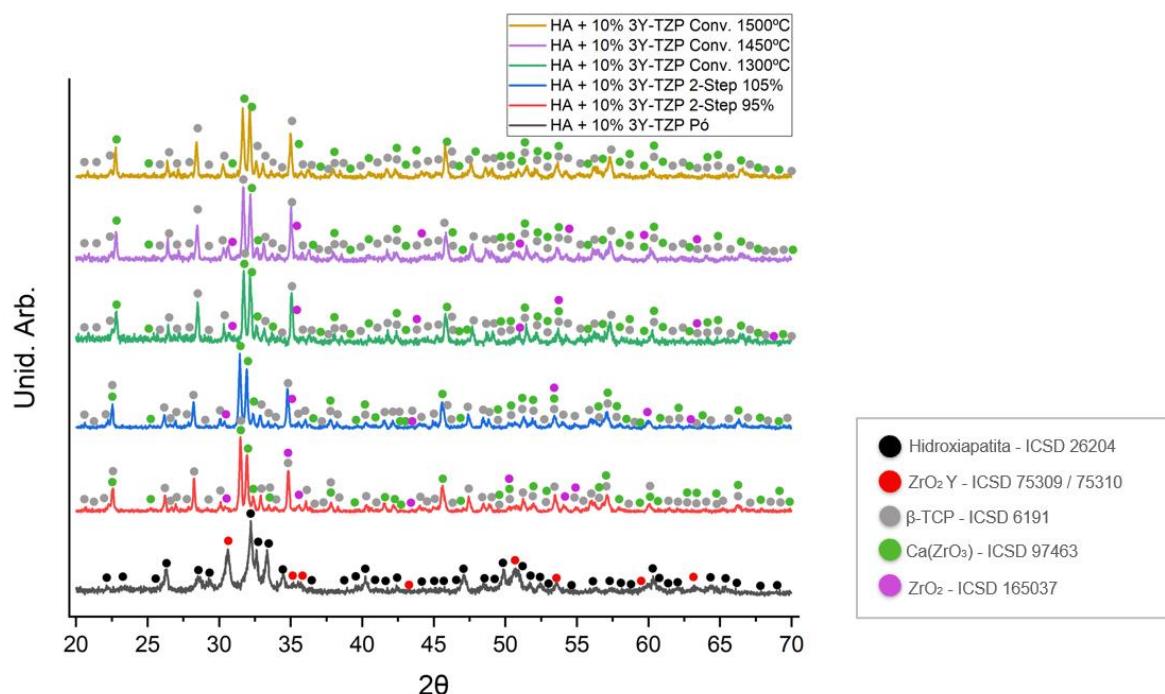


Figure 4. X-Ray Diffraction (XRD) of samples containing HA + 10%3YTZP.

3.4. Scanning Electron Microscopy (SEM)

The SEM images of the fractured surface of the pure HA samples show a similarity between the 1300C, 95%2S, and 105%2S groups. It is also possible to observe apparently greater cohesion between the grains of the 1300C and 150%2S groups. The surface of the 1450C and 1500C was characterized by greater grain disorganization and a apparently higher number of pores (Figure 5).

The addition of 3Y-TZP nanoparticles to the HA matrix resulted in increased porosity and the presence of more defects, as well as influencing densification. Therefore, the SEM analysis indicated that the HA+10%3Y-TZP group exhibited a less uniform composite (Figure 6).

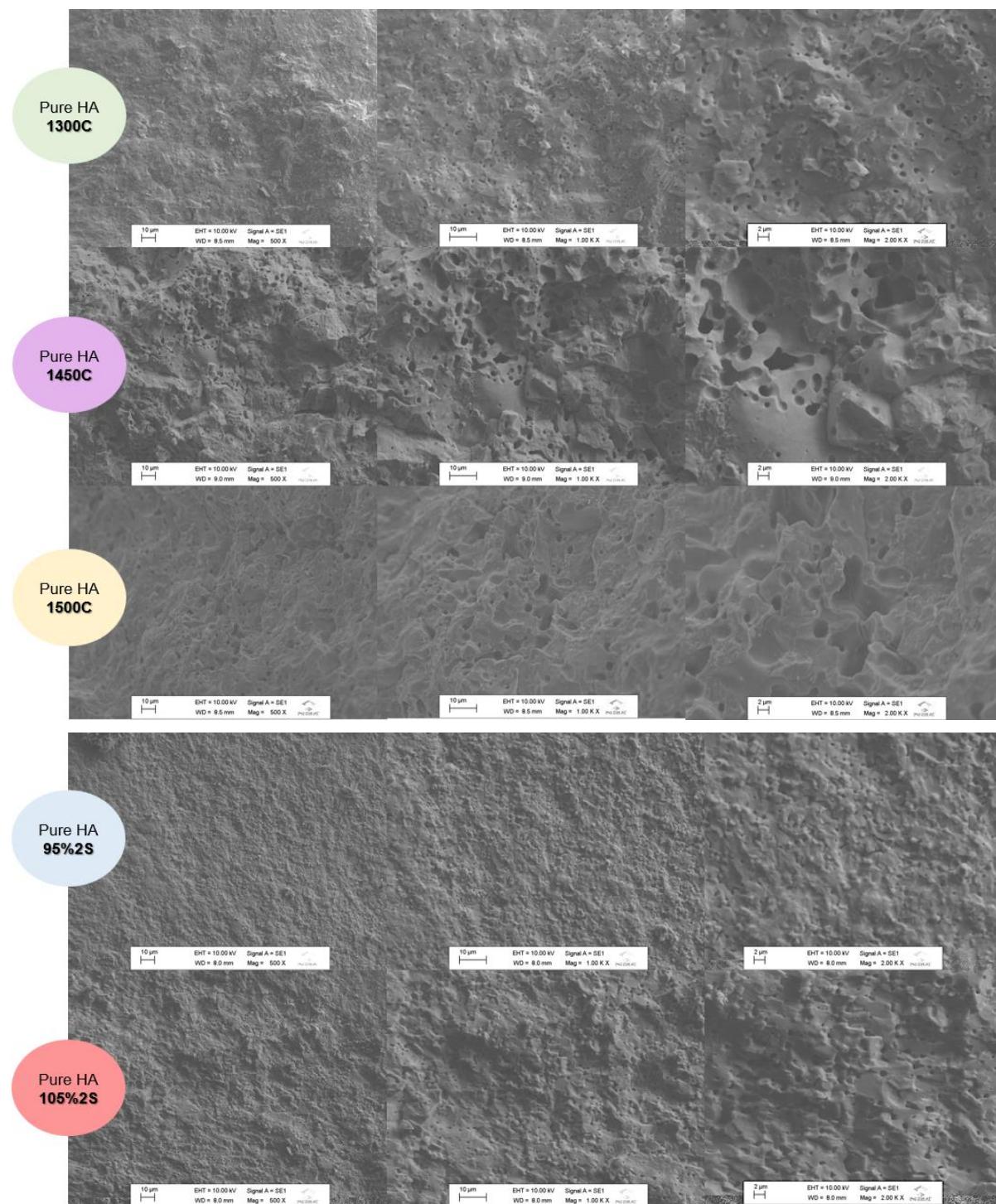


Figure 5. Micrographs of the fractured surface (bulk) of sintered specimens Pure HA: 1300C (A), 1450C (B), 1500C (C), 95%2S (D), and 105%2S (E).

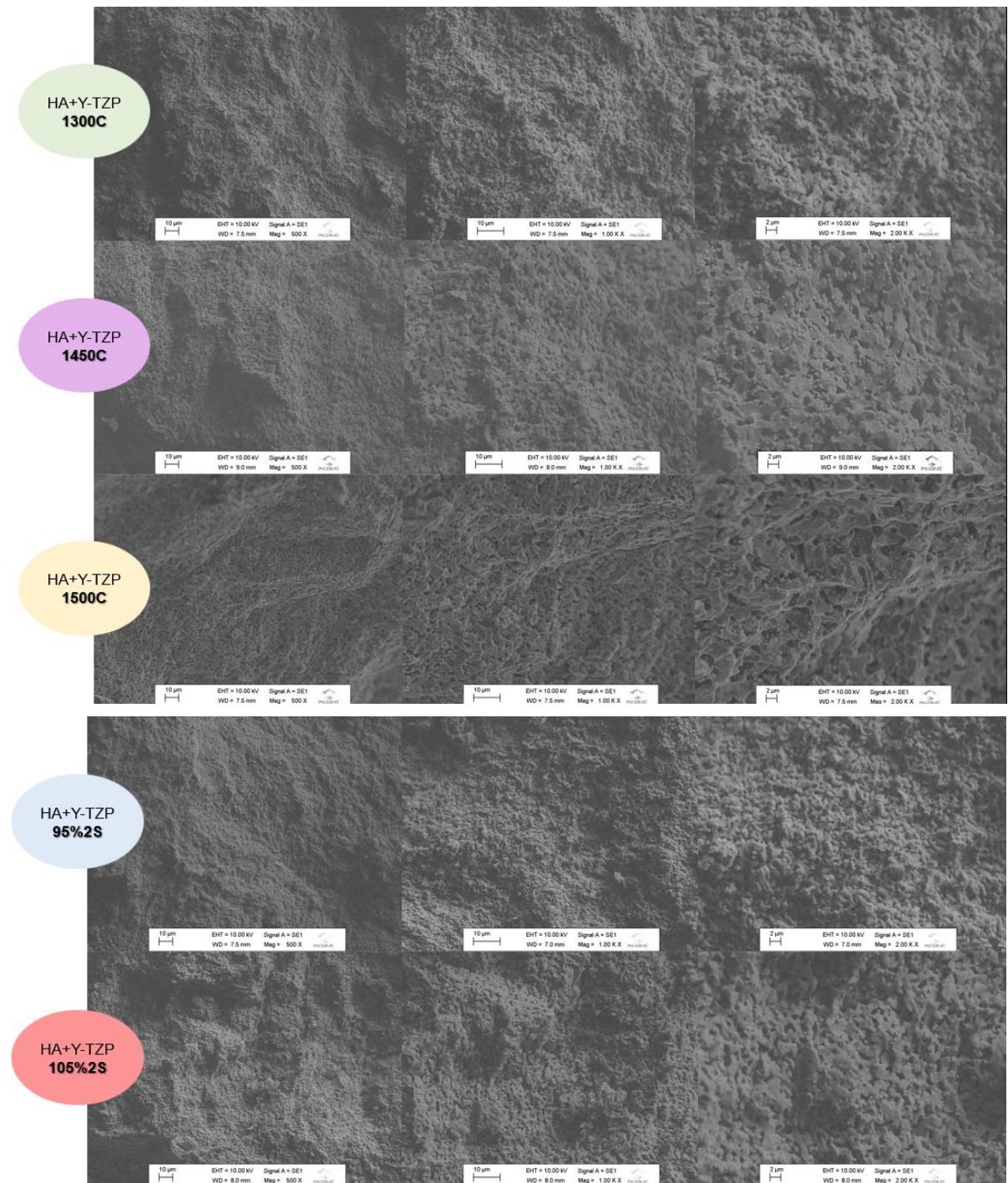


Figure 6. Micrographs of the fractured surface (bulk) of sintered specimens of HA + 10% 3Y-TZP: 1300C (A), 1450C (B), 1500C (C), 95%2S (D), and 105%2S (E).

4. DISCUSSION

The null hypothesis of the study was rejected, as the two-step sintering process resulted in distinct microstructural and mechanical properties compared to conventional curves for the investigated groups. Notably, the 150%2S groups, both for Pure HA and HA+10%3Y-TZP, showed promising results compared to other groups with the same composition tested at different temperatures in both conventional and two-step curves.

The values of biaxial flexural strength (BFS) for pure HA indicated that the 105%2S group, along with the 1300C group, recorded values exceeding 100 MPa and showed no statistically significant difference, consistent with the microhardness analysis results. This finding, favoring the two-step sintering parameter, aligns with findings from other studies [66,67], but contrasts with the conclusions of Minim et al. (2023) [68]. The presence of binders, such as polyvinyl butyral (PVB), and lower temperatures in the two-step methodology may account for the discrepancies observed between the studies.

For the HA+10%3Y-TZP composite, the 105%2S group stood out with higher RFB values, statistically differing from all other groups. Conversely, the conventional curves yielded lower results, with no statistically significant difference among them. When considering these results alongside the microhardness values, which demonstrate better outcomes at higher temperatures in both sintering methods, it is evident that higher temperatures led to an improvement in mechanical properties in samples with 3Y-TZP addition. This aligns with the literature, which suggests that adding Y-TZP to an HA matrix requires the use of high temperatures to produce dense samples, as conventional sintering of commercial 3Y-TZP powder, on the nanoscale, can only achieve high densities at temperatures above 1300°C [71]. In this study, the lowest results for the groups with 10% zirconia nanoparticle addition were observed among the conventional curves, with a Vickers microhardness of 58.7 ± 3.66 for the 1300C group and an RFB of 32.4; 31.3, 32.6 MPa for the 1450C group.

However, Hydroxyapatite (HA) proves to be unstable at temperatures above 1300°C, leading to its decomposition and the formation of α- and β-TCP [$\text{Ca}_3(\text{PO}_4)_2$], calcium oxide (CaO), and water. Furthermore, the nanoparticulation of HA improves the sintering process of the sample and reduces the temperature required for densification to around 1100°C [72]. Therefore, it is justified that for pure HA, the 1500C

group, with a conventional methodology and maximum temperature of 1500°C, exhibited the lowest combined result of microhardness (151 ± 47.1) and biaxial flexural strength (26.4; 24.6, 28.0 MPa).

From the XRD analysis, it is observed that the formation of secondary phases occurs not only with increasing temperature but also with the addition of nanoparticles in both groups. However, in the group composed of pure HA 105%2S, despite the high temperature, only 0.9% of CaO formation was observed, maintaining a significant amount (99.1%) of crystalline peaks consistent with the pattern of Hydroxyapatite (26204-ICSD). This indicates that the decomposition of HA was lower than at high temperatures with conventional curves (1450C and 1500C).

Samples with the addition of 3Y-TZP exhibited lower results compared to pure HA, a finding consistent with the SEM images presented, which show a higher presence of pores along the surfaces, reducing the material's resistance. The incorporation of ZrO₂ into HA, as observed by Nayak *et al.* (2008), indeed influences porosity and densification [73]. A previous study indicated that ZrO₂ nanoparticles, obtained through a hydrothermal method, did not contribute to the increase in the mechanical strength of HA, with the HA + 15% ZrO₂ composite exhibiting SEM images with a considerable number of pores compared to other tested groups [48]. According to an investigation involving YTZP/HA in proportions of 95/5, 90/10, 85/15, and 80/20, pore development can be attributed to the evolution of water vapor during the reaction between zirconia and hydroxyapatite [3].

In previous studies, the addition of 5% and 8% TiO₂ nanoparticles to the HA matrix [47] demonstrated the absence of secondary phase formation in a conventional sintering curve with a maximum temperature of 1300°C. However, another study using ZnO nanoparticles in the HA matrix indicated that samples with 1% ZnO nanoparticle addition did not show secondary phase formation, while those with 5% exhibited more pronounced formation [26].

When 10% of 3Y-TZP was added, the formation of secondary phases was observed in all sintering parameters. Our findings are consistent with those of Towler *et al.* (2000) [30], who identified inherent disadvantages in the combination of HA + Y-TZP (300 and 50 nm, respectively). With the increase in ZrO₂ content beyond 4% by weight, even at relatively low temperatures, the decomposition of HA was intensified,

resulting in an increase in the amount of β -TCP formed with additions of 6 and 10% by weight of ZrO_2 in the HA matrix.

The addition of zirconia phases not only necessitates higher sintering temperatures but also reduces the decomposition temperature of HA to the range of 1000-1050°C due to the interaction between these materials. At 1150°C, the decomposition of HA results in the formation of other phases, including calcium-stabilized cubic zirconia and calcium zirconate ($CaZrO_3$).

In the presence of zirconia (10% by volume), nearly all HA decomposes into some TCP phase at 1400°C, reaching only 78.8% relative density at 1500°C/1h [72]. Consequently, the accelerated decomposition of HA in the presence of zirconia restricts densification and impairs the mechanical properties of the composites, negating the advantage of combining a bioactive ceramic of low resistance (HA) with a bioinert ceramic of high resistance (Y-TZP) [56]. Further studies are required to reduce the reactivity between these materials and assess long-term effects, using 3Y-TZP percentages below 4% with the two-step sintering methodology under hydrothermal aging.

5. CONCLUSION

The samples of pure HA exhibited superior mechanical properties compared to those containing 10% of 3Y-TZP nanoparticles. However, the two-step sintering parameter with the higher temperature (105%2S) yielded the best results among the groups with nanoparticle addition (HA+10%3Y-TZP). Moreover, among the pure HA groups, it was statistically similar to conventional sintering (1300C), which achieved the best results. Therefore, the modification of the conventional sintering methodology was crucial for optimizing the studied material.

REFERÊNCIAS

1. ZHANG, Y.; KELLY JR. Dental Ceramics for Restoration and Metal Veneering. **Dent Clin North Am.**, v. 61, n. 4, p. 797-819, 2017.
2. MEIRA, C. R. **Processamento de hidroxiapatita bovina associada com prototipagem rápida visando implantes ósseos.** 2014. 159 f. Dissertação (Doutorado) – Escola de Engenharia de São Carlos, Universidade de São Paulo, São Carlos, 2014.
3. FERREIRA, C.R.D. *et al.* Study of microstructural, mechanical, and biomedical properties of zirconia/hydroxyapatite ceramic composites. **Ceram. Int.**, v. 48, n. 9, p. 12376-12386, 2022.
4. PIRES, A. L. R.; BIERHALZ, A. C.; MORAES, A.M. Biomaterials: types, applications, and market. **Quím. Nova**, v. 38, n. 7, p. 957–97, Ago. 2015.
5. KUMAR, V. A. *et al.* Effect of ZrO₂ content on the mechanical properties and microstructure of HA_p/ZrO₂ nanocomposites. **Ceram. Int.**, v. 44, p. 10345–10351, Jun. 2018.
6. ANDREIUOLO, R.; GONÇALVES, S. A.; DIAS, K. R. H. C. A zircônia na Odontologia Restauradora. **Rev. Bras. Odontol.**, v. 68, p. 49, 2011.
7. AN, L. *et al.* Controlled additive-free hydrothermal synthesis and characterization of uniform hydroxyapatite nanobelts. **Ceram. Int.**, v. 42, p. 3104–3112, 2016.
8. EANES ED. Crystal-growth of mineral phases in skeletal tissues. **Prog Cryst Growth Ch.** 1980;3(1):3–15.
9. OGUCHI, H. *et al.* Long-Term Histological-Evaluation of Hydroxyapatite Ceramics in Humans. **Biomaterials**, v. 16, n. 1, p. 33-8, 1995.
10. SADAT-SHOJAI, M. *et al.* Synthesis methods for nanosized hydroxyapatite with diverse structures. **Acta Biomater.** v. 9, n. 8, p. 7591-7621, Aug. 2013.
11. KAWACHI, E. Y. *et al.* Bioceramics: Tendencies and perspectives of an interdisciplinary area. **Quim Nova**, v. 23, n. 4, p. 518-22, 2000.
12. CIOBANU, G.; HARJA, M. Cerium-doped hydroxyapatite/collagen coatings on titanium for bone implants. **Ceram. Int.**, v. 45, p. 2852–2857, Feb. 2019.
13. ELIAZ N, METOKI N. Calcium phosphate bioceramics: a review of their history, structure, properties, coating technologies and biomedical applications. **Materials (Basel)**. v. 10, n. 4, p. 334. 2017.
14. BORDEA, I.R. *et al.* Nanohydroxyapatite use in dentistry: a systematic review. **Drug Metab Rev.**, v. 52, n. 2, p. 319-32, maio. 2020.

15. GHIASI, B. *et al.* Hydroxyapatite as a biomaterial - a gift that keeps on giving. **Drug Dev Ind Pharm.**, v. 46, n. 7, p. 1035-1062, jul. 2020. DOI: 10.1080/03639045.2020.1776321.
16. FERRAIRO, B. M. *et al.* Production of bovine hydroxyapatite nanoparticles as a promising biomaterial via mechanochemical and sonochemical methods. **Mater. Chem. Phys.**, v. 295, fev. 2023. ISSN 0254-0584. Disponível em: <https://doi.org/10.1016/j.matchemphys.2022.127046>.
17. AKRAM, M. *et al.* Extracting hydroxyapatite and its precursors from natural resources. **J Mater Sci.**, v. 49, p.1461-75, 2014.
18. MOHD PU'AD N. A. S. *et al.* Syntheses of hydroxyapatite from natural sources. **Heliyon**, v. 5, n. 5, e01588, 2019.
19. AGBEBOH, N.I. *et al.* Environmentally sustainable processes for the synthesis of hydroxyapatite. **Heliyon**, v. 6, n. 4, e03765, 2020.
20. SZABO-JÚNIOR, A. M. **Educação ambiental e gestão de resíduos**. 3. ed. São Paulo: Rideel, 2010.
21. JABBOUR, A. B. L. S.; JABBOUR, C. J. C. **Gestão ambiental nas organizações: fundamentos e tendências**. São Paulo: Atlas, 2013.
22. LUPI, O. Prionic disease: evaluation of the risks involved in using products of bovine origin. **An bras Dermatol**, v. 78, n. 1, p. 7-18, jan./fev. 2003.
23. SOGAL, A. TOFE, A. J. Risk assessment of bovine spongiform encephalopathy transmission through bone graft material derived from bovine bone used for dental applications. **J Periodontol.** v. 70, n. 9, p.1053-1063. 1999. doi:10.1902/jop.1999.70.9.1053
24. RACE, R. E. *et al.* Scrapie infectivity and proteinase K-resistant prion protein in sheep placenta, brain, spleen and lymph nodes: Implications for transmission and antemortem diagnosis. **J Infect Dis**, v. 178, p. 949-53, 1998.
25. SCHWENKE, K. A. *et al.* Kinetics of the reduction of CJD prion seeding activity by steam sterilization support the use of validated 134 °C programs. **J Hosp Infect.** v.132, p. 125-132. Feb. 2023: doi: 10.1016/j.jhin.2022.08.014.
26. PIRES, L. A. *et al.* Effects of ZnO/TiO₂ nanoparticle and TiO₂ nanotube additions to dense polycrystalline hydroxyapatite bioceramic from bovine bones. **Dent Mater**, v. 36, 2020, e38-e46. Disponível em: <https://doi.org/10.1016/j.dental.2019.11.006>.
27. CHEN, L. *et al.* The role of surface charge on the uptake and biocompatibility of hydroxyapatite nanoparticles with osteoblast cells. **Nanotechnology**, v. 22, n. 10, 2011.
28. PRAKASAM, M. *et al.* Fabrication, Properties and Applications of Dense Hydroxyapatite: **J Funct Biomater.**, v. 6, p. 1099-1140, dez. 2015.

29. MBARKI, M. et al. Hydroxyapatite bioceramic with large porosity. **Mater Sci Eng C Mater Biol Appl**, v. 76, p. 985-990, jul. 2017. DOI: 10.1016/j.msec.2017.03.097.
30. TOWLER, M. R.; GIBSON, I. R.; BEST, S. M. Novel processing of hydroxyapatite-zirconia composites using nano-sized particles. **J. Mater. Sci. Lett.** v.19, p. 2209 – 2211, 2000.
31. CANILLAS, M. et al. Processing of hydroxyapatite obtained by combustion synthesis. **Bol. Soc. Espanola Ceram. Vidr**, v. 56, p. 237–242, 2017.
32. CATAURO, M. et al. Morphological and thermal characterization of zirconia/hydroxyapatite composites prepared via sol-gel for biomedical applications. **Ceram. Int**, v. 45, p. 2835–2845, 2019.
33. BOUSLAMA, N.; BEN AYED F.; BOUAZIZ, J. Effect of fluorapatite additive on densification and mechanical properties of tricalcium phosphate. **J Mech Behav Biomed Mater**, v. 3, n. 1, p. 2-13, 2010.
34. SEBDANI, M. M.; FATHI, M. H. Preparation and characterization of hydroxyapatiteforsterite-bioactive glass nanocomposite coatings for biomedical applications. **Ceram. Int.**, v. 38, n. 2, p.1325-30, 2012.
35. AYOUB, G. et al. Composite nanostructured hydroxyapatite/yttrium stabilized zirconia dental inserts—The processing and application as dentin substitutes. **Ceram. Int.**, v. 44, p. 18200–18208, 2018.
36. EVANS, A. G. Perspective on the Development of High-Toughness Ceramics. **J. Am. Ceram.v** 73, n.2, p.187-206, 1990.
37. FORTULAN, C. A. **Compósito alumina-zircônia: obtenção através de conformação coloidal e caracterização microestrutural** [Relatório de Pós-doutorado]. São Carlos: Universidade Federal de São Carlos; 1999.
38. FANG, Z.; FENG, Q. L.; TAN, R. W. In-situ grown hydroxyapatite whiskers reinforced porous HA bioceramic. **Ceram. Int.**, v. 39, n. 8, p. 8847-52, 2013.
39. SAJJAD; M. et al. Study of the effect of the concentration, size and surface chemistry of zirconia and silica nanoparticle fillers within an epoxy resin on the bulk properties of the resulting nanocomposites. **Polym. Int.**, v.6, n.2, p.274-285, Feb. 2012.
40. RAMIRES, P. A. et al. The influence of titania/hydroxyapatite composite coatings on in vitro osteoblasts behaviour. **Biomaterials**, v. 22, n. 12, p. 1467-74, 2001.
41. SILVA, V. V.; LAMEIRAS, F. S.; DOMINGUES, R. Z. Microstructural and mechanical study of zirconia-hydroxyapatite (ZH) composite ceramics for biomedical applications. **Compos Sci Technol**, v. 61, n. 2, p. 301-10, 2001.

42. KALITA, S. J. Effects of MgO-CaO-P₂O₅-Na₂O-based additives on mechanical and biological properties of hydroxyapatite. **J Biomed Mater Res.** Part A, v. 71, n. 1, p. 35-44, 2004.
43. RAVARIAN, R. *et al.* Synthesis, characterization and bioactivity investigation of bioglass/hydroxyapatite composite. **Ceram. Int.**, v. 36, n. 1, p. 291-7, 2010.
44. YELTEN, A.; YILMAZ, S.; OKTAR, F. N. Sol-gel derived alumina-hydroxyapatite-tricalcium phosphate porous composite powders. **Ceram. Int.**, v. 38, n. 4, p. 2659-65, 2012.
45. NING, C. Q.; ZHOU, Y. On the microstructure of biocomposites sintered from Ti, HA and bioactive glass. **Biomaterials**, v. 25, n. 17, p. 3379-87, 2004.
46. NATH, S. *et al.* Nanoindentation response of novel hydroxyapatite-mullite composites. **Mater. Sci. Eng.** v. 513-14, p. 197-201, 2009.
47. DE AZEVEDO-SILVA, L. J. *et al.* Novel 8%-TiO₂-nanoparticle-reinforced dense polycrystalline bovine hydroxyapatite bioceramic. **Int. J. Ceram. Eng. Sci.**, v. 4, n. 3, p. 158-169, 2022.
48. PADOVINI, D. S. S. *et al.* Hydroxyapatite/ZrO₂@ SiO₂ bioceramic composite: Producing a promising biomaterial from natural sources. **MRS Commun.**, v. 13, n. 4, p. 657-663, 2023.
49. DARMAWAN, B. A. *et al.* Two-step sintering of partially stabilized zirconia for applications in ceramic crowns. **Materials**, v. 13, n. 8, p. 1857, 2020.
50. RAMESH, S.; MURALITHRAN, G. Phase stability and microstructural development of y-tzp-doped hydroxyapatite. **Biomed Eng**, v. 13, n. 02, p. 66-71, 2001.
51. EL-GHANY, O.S.A.; SHERIEF, A.H. Zirconia based ceramics, some clinical and biological aspects. **Futur. Dent. J.** v. 2, n. 2, p. 55-64, 2016.
52. RAMOS, C. M. *et al.* Application of Micro-Raman Spectroscopy to the Study of Yttria-Stabilized Tetragonal Zirconia Polycrystal (Y-TZP) Phase Transformation. **Appl. Spectrosc.** v. 69, n. 7, p. 810-4, 2015.
53. GRACIS, S. *et al.* A New Classification System for All-Ceramic and Ceramic-like Restorative Materials. **Int. J. Prosthodont.**, v. 28, n. 3, p. 227-235, 2015.
54. LOPES, A. C. O. **Microstructural, mechanical, and optical characterization of a ZTA (zirconia-toughened alumina) composite after aging.** 2019. 81 p. Dissertação (Mestrado) – Faculdade de Odontologia de Bauru, Universidade de São Paulo, Bauru, 2014.
55. BENALCÁZAR-JALKH, E. B. *et al.* A Narrative Review on Polycrystalline Ceramics for Dental Applications and Proposed Update of a Classification System. **Materials**, v. 16, n. 24, p. 7541, 2023.

56. CASTKOVA, K. *et al.* Synthesis of Ca, Y-zirconia/hydroxyapatite nanoparticles and composites. **J. Eur. Ceram. Soc.** v. 36, p. 2903–2912, 2016.
57. CURRAN, D. J. *et al.* Mechanical properties of hydroxyapatite–zirconia compacts sintered by two different sintering methods, **J. Mater. Sci. Mater. Med.**, v. 21, p. 1109–1120, 2010.
58. INUZUKA, M. *et al.* Hydroxyapatite-doped zirconia for preparation of biomedical composites ceramics. **Solid State Ionics**, v. 172, p. 509–513, 2004.
59. KAWASHIMA, N. *et al.* The surface characteristics of the sintered body of hydroxyapatite–zirconia composite particles. **Colloids Surf. B Biointerfaces**, v. 10, p. 23–27, 1997.
60. MATSUNO, T. *et al.* Microstructure and mechanical properties of sintered body of zirconia coated hydroxyapatite particles. **J. Mater. Sci. Lett.**, v. 19, p. 573–576, 2000.
61. SILVA, V. V.; DOMINGUES, R. Z. Hydroxyapatite–zirconia composites prepared by precipitation method, **J. Mater. Sci. Mater. Med.**, v. 8, p. 907–910, 1997.
62. PROKOPENKO, O.; SEVOSTIANOV, I. Dependence of the mechanical properties of sintered hydroxyapatite on the sintering temperature. **Mater. Sci. Eng. A**, v. 431, n. 1–2, p. 218–227, Sep. 2006.
63. CHU, M.-Y. *et al.* Precoarsening to Improve Microstructure and Sintering of Powder Compacts. **J. Am. Ceram. Soc.**, v. 74, p. 2902–2911, 1991.
64. CHEN, I. W.; WANG, X. H. Sintering dense nanocrystalline ceramics without final-stage grain growth. **Nature**, v. 404, n. 6774, p. 168–171, 2000.
65. LOURENÇO, M. A. *et al.* Model of two-step sintering conditions for yttria-substituted zirconia powders. **Mat. Chem. Phys.**, v. 126, n. 1–2, p. 262–271, May. 2011.
66. WANG, X. H.; CHEN, P. L.; CHEN, I. W. Two-Step Sintering of Ceramics with Constant Grain-Size, I. Y₂O₃. **J. Am. Ceram. Soc.** v. 89, p. 431 – 437. 2006.
67. LIN, K.; CHEN, L.; CHANG, J. Fabrication of Dense Hydroxyapatite Nanobioceramics with Enhanced Mechanical Properties via Two-Step Sintering Process. International. **J. Appl. Ceram. Technol.** v. 9, n. 3, p. 479–485, May. 2012.
68. MINIM, P. R. *et al.* The combined effects of binder addition and different sintering methods on the mechanical properties of bovine hydroxyapatite. **J. Mech. Behav. Biomed. Mater.** v. 144, 2023. <https://doi.org/10.1016/j.jmbbm.2023.105993>.
69. SUTHARSINI, U.; THANIHAIACHELVAN, M.; RAMESHSINGH, R. Two-Step Sintering of Ceramics. In: **Sintering of Functional Materials**. London: IntechOpen; 2017.

70. ISO - International Organization for Standardization. ISO/BS 6872:2015: dentistry: ceramic materials. London: ISO, 2015. 28p.
71. PALMEIRA, A. A. *et al.* Densification and grain growth of nano- and micro-sized Y-TZP powders. **Ceram. Int.** v. 42, p. 2662-2669, 2016.
72. BRZEZIŃSKA-MIECZNIK, J. *et al.* Hydroxyapatite of natural origin - zirconia composites, preparation and reactions within the system. **Process. Appl. Ceram.**, v. 10, p. 219-225, 2016.
73. NAYAK, Y. *et al.* Pressureless sintering of dense hydroxyapatite-zirconia composites. **J. Mater. Sci. Mater. Med.**, v. 19, n. 6, p. 2437-44, 2008. doi: 10.1007/s10856-008-3371-z.
74. SUTHARSINI, U. *et al.* Effect of sintering holding time on low-temperature degradation of yttria stabilised zirconia ceramics. **Mater. Res. Innov.**, v. 18, p. S6-408–S6-411, 2014. DOI: 10.1179/1432891714Z.000000000988.
75. MAZAHERI, M. *et al.* Effect of a novel sintering process on mechanical properties of hydroxyapatite ceramics. **J. Alloys Compd.** v. 471, p. 180–184, 2009. DOI: 10.1016/j.jallcom.2008.03.066.

Anexo

Anexo A – Declaração de uso exclusivo de artigo em dissertação.



DECLARAÇÃO DE USO EXCLUSIVO DE ARTIGO EM DISSERTAÇÃO/TESE

Declaramos estarmos cientes de que o trabalho "Impact of the incorporation of 3Y-TZP nanoparticles and different sintering methods in bovine hydroxyapatite: development, microstructural characterization, and mechanical properties" será apresentado na Dissertação/Tese do(a) aluno(a) KARLA DRUZIAN OLIVEIRA e que não foi e nem será utilizado em outra dissertação/tese dos Programas de Pós-Graduação da FOB-USP.

Bauru, 16 de fevereiro de 2024.

Karla Druzian Oliveira
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Anexo

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