## FERNANDA MENDES DE REZENDE

Chemical analysis of color change and transcriptional profile in flavonoids biosynthesis during flower development of *Tibouchina pulchra* (Cham.) Cogn.

Análises químicas da mudança de cor e perfil transcricional da biossíntese de flavonoides durante o desenvolvimento floral de *Tibouchina pulchra* (Cham.) Cogn.



São Paulo

2018

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> Tese apresentada ao Instituto de Biociências da Universidade de São Paulo, para a obtenção de Título de Doutor em Botânica, na Área de Recursos Econômicos vegetais.

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São Paulo

2018

Rezende, Fernanda Mendes Análises químicas da mudança de cor e perfil transcricional da biossíntese de flavonoides durante o desenvolvimento floral de *Tibouchina pulchra* (Cham.) Cogn. 165 páginas.

Tese (Doutorado) – Instituto de Biociências da Universidade de São Paulo. Departamento de Botânica.

Palavra-Chave: 1. antocianinas; 2. Ácidos fenólicos; 3. CHS, 4. FLS, 5. ANS.

Universidade de São Paulo. Instituto de Biociências. Departamento de Botânica.

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Dedico à minha família e a todos que de alguma forma contribuíram para a realização deste trabalho.

## Agradecimentos

Ao fim de quatro anos tenho muito o que agradecer. Foram anos de aprendizado, dedicação e mudanças. Muitas pessoas participaram deste processo.

Primeiramente gostaria de agradecer à Claudia, minha orientadora, por toda paciência, incentivo e suporte em minha formação. Desde a iniciação científica até aqui passaram-se dez anos, meu amadurecimento científico deve-se em grande parte ao seu apoio. Obrigada.

À Magda, minha co-orientadora, agradeço por me inserir no mundo da biologia molecular e por todo aprendizado acadêmico compartilhado. Também sou muito grata pelo grande apoio à ida ao exterior. Não poderia deixar de mencionar que as pausas para um cigarro foram sempre produtivas, seja para falar da vida ou para ter insights sobre o trabalho.

Aos professores da Fitoquímica por todos os anos em que aprendi muito com vocês, tanto numa conversa informal, quanto em nossos seminários. Ao Marcelo, agradeço por toda a ajuda nas identificações químicas e paciência diante dos meus questionamentos.

Ao Eduardo Leal da Sistemática Vegetal que me auxiliou com os nomes de espécies da grande tabela do capítulo 1. Ao Erismaldo, jardineiro do IB, que muito me ajudou para que as mudas de *Tibouchina* ficassem bem.

Ao Mads que me recebeu em seu laboratório na Dinamarca e ao Thomas, além de todos os técnicos dos laboratórios e amigos que fiz por lá. Em especial Irene, Christou e Jorge.

Aos meus colegas da Fitoquímica, agradeço pelos anos em que discutíamos resultados entre pausas de almoço e café. Aqui vai um agradecimento especial à Priscila, Alice, Fernanda, Dalila e Kátia que me ajudaram nas coletas ou regando plantas quando eu estava ausente. Falando em almoço, não posso deixar de agradecer às amigas que almoçam comigo, quase que diariamente, nos últimos anos: Aline, Kátia e Mourisa. Apesar de hoje em dia estar ausente, agradeço à Carol, pois muito trocamos nos anos de convivência.

Às técnicas do laboratório, agradeço por tantas vezes que passamos almoços falando sobre cromatografias e afins. Obrigada por serem sempre prestativas e por todas as risadas compartilhadas.

Ao pessoal do lab de Biologia Celular e Biologia Molecular de plantas, agradeço pelas trocas de experiências e bons momentos. Agradecimento especial à Dani, Paula e Bruno que me ajudaram muito nos procedimentos no laboratório.

À minha família agradeço por serem os pilares que sustentam meus sonhos. Obrigada mãe, pai, amo vocês! Não posso deixar de fora aquela família que a gente escolhe, nossos amigos. Agradeço aos meus por me levarem para me descontrair, afinal não é só de um doutorado que se vive.

Por fim, aos órgãos financiadores: CAPES, FAPESP (2013/10413-0) e Instituto de Biociências.

Gosto das cores, das flores, das estrelas, do verde das árvores, gosto de observar. A beleza da vida se esconde por ali, e por mais uma infinidade de lugares, basta saber, e principalmente, basta querer enxergar.

Clarice Lispector

Rezende, F.M. Análises químicas da mudança de cor e perfil transcricional da biossíntese de flavonoides durante o desenvolvimento floral de *Tibouchina pulchra* (Cham.) Cogn. [tese]. São Paulo: Instituto de Biociências, Universidade de São Paulo; 2018.

#### Resumo

A mudança de cor em flores é um fenômeno generalizado entre as angiospermas, mas pouco compreendido do ponto de vista genético e químico. Este estudo teve como objetivo investigar esse fenômeno em *Tibouchina pulcha*, uma espécie nativa brasileira, cujas flores mudam do branco para rosa escuro. Para alcançar o objetivo, os perfis dos compostos e de expressão de genes da via de biossíntese de flavonoides foram estudados. Utilizando técnicas hifenadas (UPLC-HRMS and NMR), trinta compostos foram quantificados e identificados, sendo: dezesseis descritos pela promeira vez em *T. pulchra*. Além idsso, um composto inédito foi identificado. Cinco genes que codificam para enzimas chaves da via foram parcialmente clonados, sequenciados, e os níveis de mRNA foram analisados (RT-qPCR) durante o desenvolvimento das flores. Coletivamente, os dados obtidos demonstram que a mudança de cor em flores de *T. pulchra* é regulada transcricionalmente. Além disso, quantificações de metais sugerem que o ion Fe<sup>3+</sup> esteja relacionado a saturação da cor no estágio rosa escuro.

Palavras-chave: antocianina, ácidos fenólicos, CHS, FLS, ANS

Rezende, F.M. Chemical analysis of color change and transcriptional profile in flavonoids biosynthesis during flower development of *Tibouchina pulchra* (Cham.) Cogn. [thesis]. São Paulo: Instituto de Biociências, Universidade de São Paulo; 2018.

#### Abstract

Floral color change is a widespread phenomenon in angiosperms, but poorly understood from the genetic and chemical point of view. This thesis aimed to investigate this phenomenon in *Tibouchina pulcha*, a native species whose flowers change from white to dark pink. To reach this goal, flavonoid biosynthetic gene expression and compounds were profiled. By using hyphenated techiniques (UPLC-HRMS and NMR), thirty phenolic compounds were quantified and identified, being sixteen described for the first time in *T. pulchra*. Moreover, an inedite compound was also identified. Five key genes of the flavonoid biosynthetic pathway were partially cloned, sequenced, and the mRNA levels were analyzed (RT-qPCR) along flower development. Collectively, the obtained results demonstrated that the flower color change in *T. pulchra* is regulated at transcriptional level. Additionally, a metal quantification suggests that Fe<sup>3+</sup> ion might influence the saturation of the color at dark pink stage.

Key words: anthocyanin, phenolic acids, CHS, FLS, ANS

## **List of Abreviations**

[M+H]: molecular ion 2-ODD: 2-oxoglutarate/Fe(Ii)-dependent dioxygenase 4CL: 4-coumarate:CoA-ligase aa: amino acids AACT: anthocyanidin acyl transferase ACT: acyl transferase amu: atomic mass unit ANS: anthocyanidin synthase AT: Arabdopsis thaliana B5GT: betanidin-5-O-glucosyltransferase bbCID: broadband collision Induced dissociation bHLH: basic helix-loop-helix BLAST: basic local alignment search tool Bp: base pair Brara: Brassica rapa C3H: coumarate 3-hydroxylase C4H: cinnamate 4-Hydroxylase CCD: carotenoid cleavage dioxygenase CD: carotene desaturase CD<sub>3</sub>OD: deuterated methanol cD5G2'GlcUT: cyclo-DOPA-5-glucose-2'-Oglucuronyl- transferase cD5GT: cyclo-DOPA-5-O-glucosyltransferase CDNA: complementary DNA cDOPA: cyclo: DOPA CHI: chalcone isomerase CHS: chalcone synthase CI: carotene isomerase Co-A: Coenzyma A Ct: treshold cycle CTAB: cetyl trimethylammonium bromide d: dublet DAD: diode array detector dd: double dublet DEPC: dietilpirocarbonato DFR: dihydroflavanol 4-reductase DMAPP: dimethylallyl diphosphate DMSO-d6: deuterated dimethyl sulfoxide DNA: desoxyribonucleic acid

DNAse: desoxyribonuclease Dntp: riphosphate desoribonucleotides DO: diphenol oxidase (DO) DOD: DOPA dioxygenase DOPA: dihydrophenylalanine DW: dried weight DXP: 1-deoxy-D-xylulose-5-phosphate DXR: DXP reductoisomerase DXS: deoxy-D-xylulose synthase DXS: DXP synthase E4P: erythrose 4-phosphate EDTA: ethylenediaminetetraacetic acid ESI: electrospray Ionization Eucgr: Eucalyptus grandis F3'5': flavonoid 3'5'-hydroxylase F3'H: flavonoid 3'-hydroxylase F3H: flavonoid 3-hydroxylase FDP: farnesyl diphosphate synthase FGT flavonoid glucosyltransferase FLS: flavonol synthase FPP: farnesyl diphosphate FS: flavone synthase GA3P: glyceraldehyde 3-P GDS: GPP synthase reaction GGDS: geranylgeranyl diphosphate synthase GGPP: geranylgeranyl diphosphate GPP: geranyl diphosphate HCI: hydrocloric acid HMBC: heteronuclear multiple bond correlation HPLC: high performance liquid chromatography HRMS: high-resolution mass spectrometry HSQC: heteronuclear single quantum coherence IFS: isoflavone synthase IPP: isopentenyl diphosphate IPPI: isopentenyl diphosphate isomerase IPTG: isopropil β-D-1-tiogalactopiranosida IspH: 4-hydroxy-3-methylbut-2-enyl diphosphate reductase J: coupling constant

LB: Luria Bertani growth medium LCYB: Lycopene B-Cyclase LCYE: Lycopene B-Cyclase m: multiplet MAL: malonate pathway Me: methyl group Medtr: Mendicago trunculata MeOH: methanol MEP: 2-C-methyl-D-erythritol 4- phosphate MEP: methylerythritol phosphate pathway MEV: mevalonate pathway miRNA: micro RNA mRNA: messenger RNA MS: mass spectrometry MYB: myeloblastosis n: nucleotide NMR: nuclear magnetic resonance NS: neoxanthin synthase OMT: O-methyl transferase P450: Cytochrome P450 Monooxygenase PAD: photodiode array detector PAL: phenylalanine ammonia lyase PCR: polymerase chain reaction PD: phytoene desaturase PEP: phosphoenolpyruvate ppm: parts per million PS: phytoene synthase QTOF: quadrupole time-of-flight

RNA: ribonucleic acid RNAi: RNA interference RT: retention time RT-Qpcr: reverse transcriptase quantitative polymerase chain reaction s: singlet S1: stage one of T. pulchra flower S2: stage two of T. pulchra flower S3: stage three of T. pulchra flower S4: stage four of T. pulchra flower SKM: shikimate pathway sl: large singlet SOC: super optimal broth Soly: Solanum lycopersicum tasiRNA: trans-acting short-interfering RNA TOH: tyrosine hydroxylase Tp: Trifolium pratense UPLC: ultra performance liquid chromatography UV/Vis: ultraviolet/visible WDRs: tryptophan-aspartic acid dipeptide repeat X-Gal: 5-bromo-4-chloro-3-indoxyl-β-Dgalactopyranoside ZE: zeaxanthin epoxidase  $\beta$ -CO:  $\beta$ -carotene oxygenase B-RH: B-Ring Hydroxylase δ: chemical shift E-RH: E-ring hydroxylase

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#### 1. INTRODUCTION

Flower colors are important in the interaction between plants and their pollinators. Since Darwin evolutionary theory, biologists have studied the role of floral variation in plant reproductive fitness and its influence in pollination. Flower colors define pollination syndromes, which are associated with pollinator groups such as bees, moths, butterflies, bats, birds, beetles, flies, and others (Fenster *et al.* 2004; Yan *et al.* 2016). The coevolution of pollinators and angiosperm drives the floral diversity (*e.g.* flower morphology, color, scent, nectar quantity and nectar quality) (Van der Niet *et al.* 2014), and this resulted in reproductive isolation and plant speciation (Kay & Sargent 2009).

Interestingly, pollinators have different color vision spectra and, due to the diverse chemical nature of molecules, the differential accumulation of specific metabolites leads to changes in pollinator preferences, which may not appear evident to the human eye (Grotewold 2006). For example, bee-pollinated flowers are often blue or yellow but rarely red, which may appear black to bees (Grotewold 2016; Papiorek *et al.* 2016).

Furthermore, flower coloring is an important aesthetic factor for the ornamental plant market since it directly influences the commercial value of flowers (Zhao & Tao 2015). Gregor Mendel, who established the fundamental laws of genetics, was interested in plant breeding and succeeded in obtaining a new variety of *Fuchsia* (Onagraceae, Myrtales) with a novel color, which rendered him a prize for his agronomic research (Freire-Maia 1995). The advent of genetic engineering tools brings new insights to basic research and breeding, aiming to obtain different coloring patterns (*i.e.* more lasting or vibrant), or completely different colors from those of the wild type genotype. This is the case of the first genetically modified commercial flower species, *Dianthus* (Caryophyllaceae, Caryophyllales), which display a wide variation of colors ranging from white to red (Tanaka *et al.* 2005).

According to the European Commission, in 2012, the European Union assumed 42.6% of the total world flower and ornamental plant production, followed by China, USA and Japan with 15.5%, 11.1% and 9.5%, respectively. Brazil is the ninth in the ranking with 1.71% of the world market (Neves & Pinto 2015). In Brazil, there is increasing investment and attention to flower and ornamental plant production market. Since 1970, the floriculture has a commercial

and professional character, promoted mainly by Japanese and Dutch immigrants. The creation of the Brazilian Institute of Floriculture (IBRAFLOR) in 1994, was an important enterprise to integrate and politically represent the Brazilian's interests at national and international levels. In 2010, a strategic agenda was created focused on encouraging and supporting the development and growth of the flower and ornamental plant domestic market (AEFP 2011). Since 2012, the increasing of this sector, with 6.17% per year, exposes the success of these strategies (Neves & Pinto 2015). Brazilian flora provides an invaluable diversity of native ornamental species to be exploited highlighting the economic potential of this market for the country's development.

The color of the flowers is one of the best-studied floral traits in terms of its genetic and chemical bases, as well as its ecological significance. Regardless the chlorophyll, there are three types of pigments present in vegetal tissues and all confers color to flowers: carotenoids, betalains and flavonoids (specially anthocyanins). These secondary metabolites are originated from products of the primary metabolism by the plastidial shikimate (SKM) and methylerythritol phosphate (MEP) pathways, and the cytosolic malonate (MAL) pathway. Carotenoids are produced by MEP, SKM originates betalains, while intermediates of SKM and MAL condensate to produce flavonoids. In this chapter, we review the biosynthesis of these compounds and the current knowledge about spatial and temporal flower color change, an unusual, but widespread phenomenon in angiosperms.

#### 1.1. Carotenoids

Carotenoids, also known as tetraterpenoids (C<sub>40</sub>), belong to terpenoid class of secondary metabolites. They are lipid soluble substances produced by all photosynthetic organisms (plants, algae and bacteria), likewise some non-photosynthetic bacteria and fungi (Hirschberg *et al.* 1997; Li *et al.* 2001). In green tissues, carotenoids act as components of the light harvesting machinery and photo-protectants from oxidative damage, being essential players for photosynthetic activity (Bartley & Scolnik 1995; Ronen *et al.* 1999; Davies 2004). Yet, in many flowers and fruits, after chloroplasts to chromoplast transition, carotenoids act as pollinator and seed dispersal attractants (Ronen *et al.* 1999; Niyogi *et al.* 2001; Lao *et al.* 2011).

Since  $19^{th}$  century, carotenoids, together with other natural pigments, attracted the attention of organic chemists.  $\beta$ -carotene was the first carotenoid isolated in 1817 and, in 1937, xanthophylls were identified as the pigments involved in autumnal leaf color (Isler 1971).

Furthermore, carotenoids are essential components of human diets, providing precursors for the biosynthesis of vitamin A (Krinsky & Johnson 2005). They also serve as antioxidants, reducing age related degeneration of the eye and lowering the risk of cancer and cardiovascular diseases (Sandmann *et al.* 2006; Johnson & Krinsky 2009; Ford & Erdman 2012).

In general, yellow and orange tones shown by tissue/organ plants are conferred by carotenoids (Tanaka 2010), especially in flowers. Carotenoids can also coexist with anthocyanins resulting in brown and bronze tones despite of their contribution for red colors (Forkmann 1991; Nisar *et al.* 2015).

The building blocks of all terpenoids are isopentenyl diphosphate (IPP) and its isomer dimethylallyl diphosphate (DMAPP) (Chappell *et al.* 1995), being synthetized by two distinct pathways: mevalonate (MEV) and MEP, cytosolic and plastidial localized, respectively (Lichtenthaler 1999; Eisenreich *et al.* 2001; Rodriguez-Concepcion & Boronat 2002). However, the exchange of intermediates between the pathways is well documented (Dudavera *et al.* 2006).

MEV pathway supplies most of the C<sub>5</sub> units for the biosynthesis of C<sub>15</sub> (sesquiterpenes), C<sub>30</sub> (triterpenes and sterols), and larger compounds (dolichols), substances whose later biosynthetic steps take place in cytosol. However, plastidial localized terpenoids, such as C<sub>5</sub> (isoprene), C<sub>10</sub> (monoterpenes), C<sub>20</sub> (diterpenes), C<sub>30</sub> (triterpenes) and C<sub>40</sub> carotenoids are synthesized via MEP (Lichtenthaler et al. 1997; Milborrow & Lee 1998; Matusova et al. 2005; Rodriguez-Concepcion 2010; Buchanan et al. 2015). The MEP pathway (Figure 1.1) initiates with glyceraldehyde 3-phosphate (GA3P) and pyruvic acid that by the action of the DEOXY-D-XYLULOSE SYNTHASE (DXS) form deoxy-D-xylulose 5-phosphate (DXP). Subsequently intramolecular rearrangement and reduction of DXP by the DXP REDUCTOISOMERASE (DXR) leads MEP formation. After some steps IPP and DMAPP are formed, and undergo sequential condensation reactions to form C<sub>20</sub> geranylgeranyl diphosphate (GGPP), precursor of carotenoid biosynthesis. The condensation of two molecules of GGPP, tail-to-tail, results in phytoene, which is further converted in lycopene by a series of desaturations and isomerizations by the action of PHYTOENE DESATURASE (PD), CAROTENE DESATURASE (CD) and CAROTENE ISOMERASE (CI). Lycopene represents a branch point in carotenoid pathway that is enzyme-mediated cyclized by LYCOPENE CYCLASES (LCY $\epsilon$ , LCY $\beta$ ), to produce  $\alpha$ -carotene or  $\beta$ -carotene (Nisar *et al.* 2015). These two compounds are precursors of the oxygenated carotenoids, xanthophylls (e.g. lutein, violaxanthin and zeaxanthin), which are produced by the action of  $\beta$ -RING HYDROXYLASE ( $\beta$ -RH) and  $\epsilon$ -RING HYDROXYLASE ( $\epsilon$ -RH) (Finkelstein 2013; Nisar *et al.* 2015).

The enzyme encoding genes involved in carotenogenesis have been identified in many plant species, while its regulation has been addressed by several studies (Fraser & Bramley 2004; Tanaka *et al.* 2008; Li & Yuan, 2013; Liu *et al.* 2015a). Some steps are regulated by light and chloroplast redox status (Jahns & Holzwarth 2012; Llorente *et al.* 2016). In Solanaceae crop species (*e.g.* tomato fruit) and *Arabidopsis thaliana* (L.) Heynh model plant, transcription factors of carotenoid biosynthetic genes were well characterized (*e.g.* PHYTOCHROME INTERACTING FACTOR 1a, RIPENING INHIBITOR, etc) (Toledo-Ortiz *et al.* 2010; Martel *et al.* 2011), demonstrating that the accumulation of carotenoids is transcriptionally regulated. Red-colored flowers due to the accumulation of carotenoids are rare among angiosperms and even scarcer are the studies about their biosynthetic regulation (Ohmiya 2011).



Figure 1.1. Simplified schematic representation of plant pigment biosynthetic pathways. Primary metabolism, carotenoid, flavonoid and betalain pathways are highlighted in green, orange, red, and purple, respectively. In blue boxes are the names of the main intermediate for each pathway: SKM (shikimate), MEP (methyl-D-erythritol 4-P), and MAL (malonate). Abbreviations indicate the following metabolites and enzymes: 1-deoxy-D-xylulose-5-P (DXP), 4-HYDROXY-3-METHYLBUT-2-ENYL DIPHOSPHATE REDUCTASE (IspH), ACYL TRANSFERASE (ACT), ANTHOCYANIDIN ACYL TRANSFERASE (AACT), ANTHOCYANIDIN SYNTHASE (ANS), AROGENATE DEHYDRATASE (AD), AROGENATE DEHYDROGENASE (ADH), CAFFEIC ACID O-METHYLTRANSFERASE (COMT), CAROTENE ISOMERASE (CI), CHALCONE SYNTHASE (CHS), CHALCONE ISOMERASE (CHI), CHALCONE REDUCTASE (CHR), CHORISMATE MUTASE (CM), CINNAMATE 4-HYDROXYLASE (C4H), Coenzyma A (CoA), COUMARATE 3-HYDROXYLASE (C3H), cvclo-DOPA (cDOPA), DIHYDROFLAVANOL 4-REDUCTASE (DFR), dihydrophenylalanine (DOPA), dimethylallyl diphosphate (DMAPP), DIPHENOL OXIDASE (DO), DOPA DIOXYGENASE (DOD), DXP REDUCTOISOMERASE (DXR), DXP SYNTHASE (DXS), erythrose 4-phosphate (E4P), farnesyl diphosphate (FPP), FARNESYL DIPHOSPHATE SYNTHASE (FDS), FLAVANONE 3'5'-HYDROXYLASE (F3'5'H), FLAVANONE 3'-HYDROXYLASE (F3'H), FLAVANONE 3-HYDROXYLASE (F3H), FLAVONE SYNTHASE (FS), FLAVONOID GLYCOSYL TRANSFERASE (FGT), FLAVONOL SYNTHASE (FLS), geranyl diphosphate (GPP), geranylgeranyl diphosphate (GGPP), GERANYLGERANYL DIPHOSPHATE SYNTHASE (GGDS), glyceraldehyde 3- phosphate (GA3P), GPP SYNTHASE (GDS), Coenzyme A (HSCoA) ISOFLAVONE SYNTHASE (IFS), ISOPENTENYL DIPHOSPHATE ISOMERASE (IPPI), isopentenyl

diphosphate (**IPP**), LYCOPENE  $\beta$ -CYCLASE (**LCY\beta**), LYCOPENE  $\epsilon$ -CYCLASE (**LCY\epsilon**), NEOXANTHIN SYNTHASE (**NS**), *O*-METHYLTRANSFERASE (OMT), PHENYLALANINE AMMONIA LYASE (**PAL**), PHENYLPYRUVATE AMINOTRANSFERASE (**PPAT**) phosphoenolpyruvate (**PEP**), PHYTOENE DESATURASE (**PD**), PHYTOENE SYNTHASE (**PS**), PREPHENATE AMINOTRANSFERASE (**PAT**) PREPHENATE DEHYDROGENASE (**PDH**), SHIKIMATE DEHYDROGENASE (**SDH**), TYROSINE AMMONIA LYASE (**TAL**), TYROSINE OXYDASE HYDROXYLASE (**TOH**), UDP-GLUCOSYLTRANSFERASE (**UGT**), ZEAXANTHIN EPOXIDASE (**ZE**), B-CAROTENE OXYGENASE (**β-CO**),  $\beta$ -RING HYDROXYLASE (**β-RH**),  $\epsilon$ -RING HYDROXYLASE (**ε-RH**). Adapted from Andersen & Markham (2005), Dewick (2009), Khan & Giridhar (2015), and Nisar *et al* (2015).

#### 1.2. Betalains

Betalains are pigments possessing a nitrogenous core structure. They confer colors from red to violet (betacyanins) or yellow to orange (betaxanthins) and are exclusively present in members of some families of Caryophyllales order. This well-defined clade comprises 38 families being 17 betalain accumulators (Strack *et al.* 2003; Grotewold 2006; APG 2016). Betalains are considered chemosystematics markers of the core Caryophyllales (Slimen *et al.* 2017), where approximately 75 distinct compounds were identified (Khan & Giridhar 2015). Interestingly, within core Caryophyllales, only two families (*i.e.* Caryophyllaceae and Molluginaceae) are known to produce anthocyanin as the noncore Caryophyllales (Brockington *et al.* 2011).

As well as many other naturally occurring compounds, betalains present chemical instability related to several factors, such as light, temperature, oxygen, pH value, water content, and storage conditions (Gonçalves *et al.* 2012; Fernández-Lopez *et al.* 2013; Molina *et al.* 2014; Otálora *et al.* 2016).

Betalains are derived from the amino acid L-tyrosine, an aromatic amino acid synthetized by the SKM plastidial pathway (Figure 1.1). The primary substrates for this pathway are the phosphoenolpyruvate (PEP- a glycolytic intermediate) and the erythrose 4-phosphate (E4Pfrom the pentose phosphate cycle). Chorismic and prephenic acids are key intermediates that lead to the synthesis of the aromatic amino acids: phenylalanine, tryptophan, and tyrosine. These amino acids are involved in aromatic alkaloids and phenolic compounds pathways. In monocots, tyrosine is the precursor for phenolic compounds synthesis, while in eudicots, the universal precursor of numerous phenolic compounds (*e.g.* phenylpropanoids, lignans, lignins, stilbenes and flavonoids) is the phenylalanine (Maeda & Dudavera 2012).

Betalain synthesis starts with dihydrophenylalanine (DOPA) formation catalyzed by the TYROSINE OXYDASE HYDROXYLASE (TOH). DOPA is an important precursor not only of betalains but also of other secondary metabolites in plants (Tanaka 2008). The cleavage of DOPA mediated by DOPA 4,5-DIOXYGENASE (DOD) followed by non-enzymatical rearrangements result in the formation of betalamic acid, the core structure of betalains (Khan & Giridhar 2015). Betalamic acid is the chromophore molecule of both betacyanins (violet) and betaxanthins (yellow) (Tanaka 2008). Its condensation with cyclo-DOPA (cDOPA) and glucosyl derivatives, amino acids, and amine groups leads to the formation of these two categories of betalains: betacyanins and betaxanthins (Khan & Giridhar 2015) (Figure 1.1).

Betaxanthins were structurally identified in *Beta vulgaris* L. (Amaranthaceae), *Portulaca grandiflora* Hook. (Portulacaceae), *Mirabilis jalapa* L. (Nyctaginaceae), *Glottiphyllum longum* (Haw.) N.E.Br. (Aizoaceae), *Rivina humilis* L. (Phytolacaceae), *Amaranthus tricolor* L. (Amaranthaceae), among other species. Moreover, betacyanins were found in *Celosia argentea* L. (Amaranthaceae), *Iresine herbstii* Hook. (Amaranthaceae), *Bougainvilea glabra* Choisy (Nyctaginaceae), and *Gomphrena globose* L. (Amaranthaceae) (Khan & Giridhar 2015).

Betalains and anthocyanins seem to be mutually exclusive pigments because there are no evidences of the occurrence of these two pigments in the same plant. This two classes of compounds play equal biological function, since both confer similar color to flowers and fruits, and accumulate in response to stress signals. The functional and biochemical similarities among betalains and anthocyanins suggest similar biosynthetic regulatory mechanisms (Stafford 1994; Brockington et al. 2011). Studies with Phytolacca americana Z. (Phytolacaceae) (Takahashi et al. 2009) demonstrated that DOD encoding gene is transcriptional regulated. The identification of MYELOBLASTOSIS (MYB) and BASIC HELIX-LOOP-HELIX (bHLH) transcription factor responsive elements in *P. americana* DOD promoter region suggest that these proteins regulate betalain biosynthesis (Takahashi et al. 2009). Interestingly, MYB and bHLH, together with TRYPTOPHAN-ASPARTIC ACID DIPEPTIDE REPEAT proteins (WDRs) are transcription factors that interact among each other to form the MBW complex, which has been extensively demonstrated to regulate flavonoid biosynthesis (Hichri et al. 2011). Recently, the first MYBlike regulatory gene for betalain pathway, BvMYB1, has been identified in Beta vulgaris, however, unlike anthocyanin-related MYBs, the product of this gene does not interact with bHLH members of the MBW complex. Phylogenetic analyses indicated that BvMYB1 and anthocyanin MYBs of Antirrhinum majus L. (Plantaginaceae, Lamilaes) (AmROSEA) and A. thaliana (AtPAP1) (Solanaceae, Solanales) probably derive from a common ancestor gene (Hatlestad et al. 2014).

Recent studies on comparative genetics of betalain producing taxa revealed another possible explanation for the mutual exclusion between anthocyanins and betalains. The promoter region of *DIHYDROFLAVONOL 4-REDUCTASE* (*DFR*) and *ANTHOCYANIDIN SYNTHASE* (*ANS*) are different in anthocyanin and betalain producing species. Both genes are expressed in betalain producing taxa, but they do not lead to anthocyanin production (Shimada et al. 2005, 2007). It has been proposed that, even when the putative MYB and bHLH transcriptional factors for anthocyanin biosynthesis are present and bind to the *DFR* and *ANS* promoters of the Caryophyllales, other factors involved in the activation of these genes might be missing in anthocyanin non-producing plants (Broun 2005; Ramsay *et al.* 2005; Shimada *et al.* 2007). The other possible explanation is the presence of a repressor-binding site in Caryophyllales promoters (Shimada *et al.* 2007). Despite of advances in this understanding, the evolution of this pathway and the mechanism behind the mutual exclusivity of betalains and anthocyanins has not been resolved (Sakuta 2014).

#### 1.3. Anthocyanins

Among angiosperms, flavonoids, specially anthocyanins, are the predominant floral pigments (Winkel-Shirley 2001; Tanaka *et al.* 2010). These pigments are responsible for the red/orange to violet/blue coloration of fruits and flowers according to the vacuole pH (Katsumoto *et al.* 2007). Similar to carotenoids, anthocyanins act as photoprotectants, due to their UV absorption spectra and high antioxidant activity. In general, in non-reproductive tissues, anthocyanins are accumulated in response to biotic and abiotic stresses, such as pathogens, high/low temperatures, high light, UV-B radiation, drought, nutrient deficiencies, high ozone concentration, etc. (Chalker-Scott 1999; Feild *et al.* 2001; Neill *et al.* 2002; Oberbaueri & Starr 2002; Steyn *et al.* 2002; Takos *et al.* 2006; Gould *et al.* 2008; Rezende & Furlan 2009).

Anthocyanins are synthesized by the combination of precursors from the SKM and MAL pathways, being the terminal branch of the highly conserved flavonoid pathway (Holton & Cornish 1995). Flavonoids constitute a relatively diverse family of aromatic molecules derived from L-phenylalanine (Winkel-Shirley 2001). Flavonoid synthesis is initiated by the action of the CHALCONE SYNTHASE (CHS) that catalyzes the condensation of *p*-coumaroyl-CoA (SKM) with three molecules of malonyl-CoA (MAL- fatty acid pathway), resulting in the formation of a tetrahydroxylalcohol. This compound is cyclized producing chalcone, precursor of all classes of

flavonoids including flavonols, flavones, flavandiols, dihydroflavonols, proanthocyanidins, isoflavonoids and anthocyanins (Dewick 2009).

Flavones are formed from the C-ring dehydrogenation of flavanones, a class of flavonoids formed by chalcone isomerization mediated by the enzyme CHALCONE ISOMERASE (CHI), which forms a third ring, characteristic in the  $C_6C_3C_6$  flavonoid basic structure (Dewick 2009).

Dihydroflavonol is the branching point for the synthesis of flavonols and anthocyanins by FLAVONOL SYNTHASE (FLS) and DFR, respectively (Cooper-Driver 2001; Saito & Yamazaki 2002; Dewick 2009; Buchanan *et al.* 2015). Anthocyanin synthesis further requires the activity of ANS, FLAVONOID GLYCOSYL TRANSFERASE (FGT), ANTHOCYANIDIN ACYL TRANSFERASE (ACT) and O-METHYL TRANSFERASE (OMT). Anthocyanidins are the aglycones and chromophores of anthocyanins (glycosylated compounds), probably the best-known and most studied group of flavonoids.

In nature, anthocyanins provide a wide range of colors. Out of the twenty-three described anthocyanidins, six are particularly abundant: cyanidin, delphinidin, malvidin, pelargonidin, peonidin, and petunidin. The difference in color tones is determined by the substitution patterns in the core structure of the B-ring (*e.g.* hydroxyl and methoxyl) and in the A and C-ring (*e.g.* glycosylations and acylations) (Castañeda-Ovando *et al.* 2009, Trouwillas *et al.* 2016).

When considering their structures, studies on anthocyanins have demonstrated that they can exist in various colored and colorless forms; the predominant form is largely determined by pH. At acidic pH (< pH 2.5), the dominant form is the colored flavylium cation form, however, about pH 3 most anthocyanins undergo nucleophilic attack by water with subsequent loss of a proton resulting in the colorless or pale-yellow hemiacetal (or hemiketal) form. The hemiacetal form can then undergo ring-opening tautomerization to the *cis*-hydroxychalcone or (*Z*)-hydroxychalcone, which can isomerize thermally or photochemically to the *trans*-chalcone or (*E*)-chalcone, both of which are also colorless or pale-yellow (Silva *et al.* 2016). Additionally, anthocyanins with two or more *ortho*-hydroxyl groups in the B-ring form complexes with divalent and trivalent metal cations, such as  $AI^{3+}$ ,  $Fe^{2+}/Fe^{3+}$ , and  $Mg^{2+}$ , which results in color changes (Buchweitz *et al.* 2001; Yoshida *et al.* 2009; Soumille *et al.*, 2013; Sigurdson *et al.* 2014).

Two categories of genes are involved in the biosynthesis of anthocyanins (Martin *et al.* 1991; Jackson *et al.* 1992): structural enzyme encoding genes, such as *PHENILALANINE AMONIA LIASE (PAL), CINNAMATE 4-HYDROXYLASE (C4H), CHS, FLAVANONE 3-HYDROXYLASE* 

(*F3H*), *DFR*, *ANS*, and *FGT*; and regulatory transcription factor encoding genes, basically MYB, bHLH, and WDR proteins, which coregulate the expression of the structural genes (Uematsu *et al.* 2014). Maize (*Zea mays* L., Poaceae, Poales), snapdragon (*A. majus*) and petunia (*Petunia hybrid* Vilm., Solanaceae, Solanales) are the major model species for the study of flavonoid biosynthesis, from which several structural and regulatory genes have been characterized (Holton & Cornish 1995; Mol *et al.* 1998; Whinkel-Shirley 2001; Dhar *et al.* 2014). Moreover, anthocyanin biosynthesis and regulation have also been studied in other plants, such as *Solanum tuberosum* L. (potato) and *S. melongena* L. (eggplant), Solanaceae (Solanales), and *A. thaliana*, etc. (Jeon *et al.* 1996; Holton & Cornish 1997; Quattrocchio *et al.* 1999; Noda *et al.* 2000; Jung *et al.* 2005; Katsumoto *et al.* 2007; Chaudhary & Mukhopadhyay 2012).

Over the past 150 years, flavonoids have featured in major scientific breakthroughs, from Mendel's elucidation of genetic bases, where seed coat color was one of the major characters followed in his experiments with peas (*Pisum sativum* L., Fabaceae, Fabales), to McClintock's discovery of transposable elements, which move in and out of flavonoid biosynthetic genes expressed in colored maize kernels (Whinkel-Shirley 2001). Flavonoid pathway also helped to clarify the phenomenon of transgene co-suppression. Aiming to intensify the violet pigmentation, an extra copy of *CHS* was introduced in *P. hybrida* resulting in totally white flowers. Expression profile analysis showed that in these transgenic plants neither the transgene nor the endogenous copy of the gene were expressed (Napoli *et al.* 1990).

Few studies regarding RNAi-based regulation of flavonoid biosynthetic pathway can be found, but certainly it will be a research focus of interest in the future. In *A. thaliana*, it was reported the regulation of MYB anthocyanin regulators by miRNAs and tasiRNA (Rajagopalan *et al.* 2006; Hsieh *et al.* 2009; Gou *et al.* 2011).

Recently, additional levels of regulation have been reported to modulate the production of anthocyanins. These include the post-translational modifications of anthocyanin transcription factors (Xie *et al.* 2012; Ye *et al.* 2012; Maier *et al.* 2013; Patra *et al.* 2013), chromatin remodeling (Hernandez *et al.* 2007), and the identification of repressor proteins that limit expression of the anthocyanin biosynthetic genes (Aharoni *et al.* 2001; Dubos *et al.* 2008; Matsui *et al.* 2008; Yuan *et al.* 2013).

Despite vacuolar pH, metal complexation and biosynthetic regulations, copigmentation influence the color of plant organs (Tanaka 2010). Copigmentation can be defined as the formation of noncovalent complexes involving an anthocyanin or anthocyanin derived pigment

and a copigment (in the presence or absence of metal ions), and the subsequent changes in optical properties of the pigment. There are over ten-thousand compounds that could potentially serve as copigments but, in any case, they should have (i) sufficiently extended  $\pi$ conjugated systems, which are supposed to favor  $\pi - \pi$  stacking interactions and (ii) hydrogen bond donor/acceptor groups such as OH and C=O groups (Trouwillas et al. 2016). The major natural copigments are hydrolyzable tannins, flavonoids, and phenolic acids. Some flavonoid classes are copigments (flavonols, flavones, flavanols, and even dihydroflavonols), which appear to be the most efficient due to extension of their  $\pi$ -conjugation over their entire tricyclic core structure (rings A, B, and C), being the flavones and flavonols (e.g. quercetin, kaempferol, isoquercitrin, and rutin) the most efficient among them. Hydroxycinnamic acids and derivatives such as caffeic, p-coumaric, ferulic, sinapic, chlorogenic, and caftaric acids are commonly described as relatively efficient copigments as well (Trouwillas et al. 2016). Some non-phenolic copigments have also been described, including alkaloids, amino acids, organic acids, nucleotides, and polysaccharides (He et al. 2012), but their efficiency is usually lower than polyphenols. It remains unclear whether these compounds can significantly contribute to color expression provided by anthocyanins (Trouwillas et al. 2016).

#### 1.4. Spatial distribution of flower pigments

Petal tissue and leaf blade structure are similar, both can be divided into four parts: upper epidermis, palisade tissue, sponge tissue, and lower epidermis. Petal pigments are mainly distributed in the upper epidermal cells, but they can also be found in the palisade tissue and the lower epidermis. For example, in pale blue grape hyacinth (*Muscari latifolium* J.Kirk, Asparagaceae, Asparagales) pigments were found in the palisade tissue (Qi *et al.* 2013), while in *Tulipa* (Liliaceae, Liliales) cv. murasakizuisho (Shoji *et al.* 2007), *Ipomoea tricolor* Cav. (Convolvulaceae, Solanales) cv. heavenly blue (Yoshida *et al.* 2009) and *Meconopsis grandis* Prain. (Papaveraceae, Ranunculales) (Yoshida *et al.* 2006), pigments occur in the petal lower epidermis. Typically, no pigment is deposited in the sponge tissue, but the thickness and density of this layer is related to the brightness of flower color. As thicker and denser the sponge tissue is, the brighter is the color (An, 1989).

The shape of petal epidermal cells also impacts on flower color. Conical cells enhance pigment light absorption, by increasing the amount of incident light on epithelial cells, thereby leading to darker flower color and enhanced color saturation. Flat cells reflect more light,

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leading to lighter flower color. Protruding papillae in epidermal cells generate a velvet shine on the petals (Zhao & Thao 2015). The deposition subcellular localization depends on the nature of the pigment. In general, carotenoids are deposited in the plastids, while flavonoids and betalains are deposited into the vacuoles.

The wide variation in flower color and color patterning is indicative of the diversity in the regulatory mechanisms of pigment biosynthetic pathways (Saito *et al.* 2006). Studies of gene expression regulation with flavonoid producing species, such as petunia and snapdragon, have clarified the genetic control of different coloring patterns, such as pigmented venation (stripes) (Schwinn *et al.* 2006; Shang *et al.* 2011), picotee (segments) (Saito *et al.* 2006), as well as distinct pigmentation/intensity among the corolla (Jackson *et al.* 1992; Schwinn *et al.* 2006; Albert *et al.* 2011; Martins *et al.* 2013).

The mechanism for vein associated pigmentation (venation) patterning is the result of anthocyanin accumulation in the epidermal cells overlaying the vasculature, in accordance to the expression domain of the MYB (transcript gradient from vasculature) and bHLH (epidermal) transcriptional factors, which control the expression of the structural genes (Shang *et al.* 2011; Davies *et al.* 2012). Some petunia cultivars have corollas with white margins and red/purple cores; or the opposite, pigmented margins and white central regions. In these cultivars, color is determined by differential accumulation of anthocyanins (petunidin, cyanidin and peonidin) and flavonols (quercetin and kaempferol) among the corolla regions. Gene expression studies showed that white margins have low levels of *CHS* mRNA, while flowers with white center accumulate a high amount of *FLS* transcripts. Thus, the white color may either be the result of the lack of flavonoid production or the deviation of the metabolic flux towards the production of the white-colored flavonol (Saito *et al.* 2006).

*Parakeelya mirabilis* Hershk. (Montiaceae, Caryophyllales) petals display violet, red, and yellow regions separated by a colorless white zone. These colored sectors accumulate different amounts of betacyanins and betaxanthins, suggesting that their biosynthesis is regulated both, spatially and temporally, by a still unknown regulatory mechanism (Chung *et al.* 2015).

#### **1.5.** Temporal distribution of pigments in flowers

Color change is not a simple degeneration phenomenon, instead it includes the gain or loss of pigments (*e.g.* such as anthocyanins, carotenoids, flavonols and betalains), metal complexation or changes in pH (Weiss 1995).

In *Viola cornuta* L. (cv Yesterday, Today and Tomorrow- Violaceae, Malpighiales), a model system for flower color change studies, the flowers change from white to purple. Fazard *et al.* (2002) studied the physiological and biochemical mechanisms involved in flower color change by the investigation of environmental signals that would influence the accumulation of anthocyanins. They verified that this change is related to the presence of pollen in the stigma and to the incidence of light, and suggested that anthocyanin biosynthesis would be regulated at the transcriptional level. This agrees with previous studies where it was observed that in some species, pollination triggers biosynthesis of anthocyanins, produced in less than 24h (Weiss 1991). Additionally, the transcriptional induction of anthocyanin biosynthetic enzyme encoding genes by light has been described in several systems (Koes *et al.* 1989; Procissi *et al.* 1997; Shin *et al.* 2007; Liu *et al.* 2015b; Kim *et al.* 2017).

*Brunfelsia calycina* Benth. (Solanaceae, Solanales), is a model for anthocyanin degradation mechanism, the petals turn from dark-purple to completely white within three days after anthesis, due to the reduction of anthocyanin concentration and increase of phenolic acid contents during flower development (Vaknin *et al.* 2005; Bar-Akiva *et al.* 2010). Zipor *et al.* (2014) showed a novel BASIC VACUOLAR PEROXIDASE as the responsible for the *in planta* degradation of anthocyanins in *B. calycina*.

Regarding carotenoid pathway, in marigold petals (*Tagetes* spp., Asteraceae, Asterales), Moehs *et al.* (2001) reported DXS and PS as the enzymes responsible for the color development from pale-yellow to deep yellow. In *Sandersonia aurantiaca* Hook. (Colchicaceae, Liliales), petals of pale-yellow flowered cultivars showed a lower expression of *PD* than those of yellow flowered cultivars, and the expression level of this gene showed to be proportional to the carotenoid accumulation level (Nielsen *et al.* 2003). Kishimoto & Ohmiya (2006) reported that the white colored chrysanthemum petals (Asterales) are the result of the up-regulation of the *CAROTENOID CLEAVAGE DIOXYGENASE* (*CCD*) gene, which degrades carotenoids into colorless compounds. *CCD* genes are also responsible for the differences in carotenoid levels between yellow and white tissues in melon, peach, potato, and saffron (Campbell *et al.* 2010; Brandi *et al.* 2011; Ibdah *et al.* 2006; Rubio *et al.* 2008).

Flower color change during anthesis is a natural phenomenon, widespread in angiosperms, and already reported in at least 33 orders, 78 families, and 253 genera; but it is an uncommon strategy regarding the few number of species into each genus (Weiss 1995; Weiss & Lamont 1997). It has been reported that color change is often triggered in response

to pollination. This may involve the whole flower, only the center, the corolla tube, the nectary/hypanthium, the nectar guide/banner petal spot, selected petals, the petal appendages, the androecium or the gynoecium (Weiss 1995). Otherwise, maintaining post-color flower change is an enhanced of floral display advantage, as it increases the attractiveness of individual plants and, consequently, the approximation frequency of pollinators since they become showy at long distances (Weiss 1995; Niesenbaum *et al.* 1999; Oberrath & Böhning-Gaese 1999; Sun *et al.* 2005). From the ecological point of view, flower color change can reflect the reward for pollination. For example, some flowers (*e.g. Weigela middendorffiana* C. Koch, *W. japonica* Thunb. – Caprifoliaceae, Dipsacales - and *Pedicularis monbeigiana* Bonati – Orobanchaceae, Lamiales) have a large amount of nectar, as well as high pollen viability and stigma receptivity, but all these characteristics decline after the floral color change (Ida & Kudo 2003, 2010; Sun *et al.* 2005; Zhang *et al.* 2012). In other cases, including *Leucojum vernum* L. (Amaryllidaceae, Asparagales), *Nicotiana rustica* L. (Solanaceae, Solanales), and *Pulmonaria officinalis* L. (Boraginaceae, Boraginales), floral color change 'advertises' nectar rewards and successfully attracts pollinators (Lunau 1996).

Bellow, Table 1.1 describes species with whole flower color change. Weiss reviewed in 1995 by using Cronquist classification, here we updated species names, families, and orders following the APG IV, Taxonomic Name Resoltion (http://tnrs.iplantcollaborative.org), and Leal, E. (personal communication). After Weiss (1995), only 17 new species were described regarding floral color change. Whole flower color change was reported for 48 families, representing 23 orders and 138 species in APG IV classification (Figure 1.2). However, only for a few species the mechanism of color change has been studied: *Viola* cornuta (white to purple), *Brunfelsia calycina* (purple to white), *Nicotiana mutabilis* Stehmann & Semir (white to pink, Solanaceae, Solanales) and *Paeonia ostii* Hong & Zhang (white to pink, Paeoniaceae, Saxifragales) (Farzad 2003; Vaknin *et al.* 2005; Macnish *et al.* 2010; Gao *et al.* 2016).

Within the species where the corolla changes the color during flower development, only for few of them, the molecular mechanism underneath this process has been explored (Farzad *et al.* 2002; Macnish *et al.* 2010; Gao *et al.* 2016; Ruxton & Schaefer 2016). Color change has been described for at least eight Brazilian native species, and in any case the mechanism that determines it has been addressed. The flower color development is predominantly mediated by anthocyanins (Table 1.1), whose biosynthesis is regulated by physical (temperature, light, water, pollinators, ion-beam radiation, and gamma rays) and chemical (pH, mineral nutrients,
and hormones) factors. The identification and characterization of genes encoding key enzymes or transcription factors involved in plant anthocyanin biosynthesis are powerful tools to develop metabolic engineering approaches aiming to manipulate flower color (Zhao & Tao 2015).

As discussed above, the color change of the flowers can be determined by the transcriptional or post-transcriptional control of the enzymes of flavonoid pathway, the presence of accessory pigments, the association of pigments with metals, or even a combination of these factors.





Taxon <sup>1</sup>		Taxon <sup>1</sup>	Color change <sup>2</sup> Diamont invol	Diamont involved <sup>3</sup>	Nativo babitat4	Poforonco
Order	Family	Species	Color change-	Pigment involved <sup>9</sup>	Native habitat	Reference
Apiales						
	Pittosp	ooraceae				
		Billardiera ringens (Harv.) E.M.Benn.	Yellow-orange to red	(+) A	Australia	Erickson <i>et al</i> . 1973
		Pittosporum rhombifolium A. Cunn. ex Hook.	White to orange-yellow	(+) C	Australia	Bailey 1949
		Pittosporum undulatum Vent.	White to yellow	(+) C*	Australia	Weiss 1995
Aspara	gales					
	Aspara	igaceae				
		Beschorneria yuccoides K.Koch	Green to pink	(+) A*	Chiapas, Mexico	Weiss 1995
	Orchid	aceae				
		Cyrtorchis arcuata (Lindl.) Schltr.	White to yellow	(+) C	E. Africa	Blundell 1987
		Epidendrum ciliare L.	White to pale yellow	(+) C	Tropical America	Weiss 1995
		Epidendrum paniculatum Ruiz & Pav.	White <del>-</del> to yellow	(+) C	Tropical America	Weiss 1995
		Epidendrum stamfordianum Bateman	White to pale yellow	(+) C	Tropical America	Weiss 1995
		Malaxis seychellarum (Kraenzl.) Summerh.	Green to brown	(+) A, (+) C	Seychelles	Friedmann 1986
		Vanda sp.	Lavender to white	(-) A*	Old World	Burg & Dulkman 1967
		Ypsilopus longifolius (Kraenzl.) Summerh.	White to yellow-green	(+) C	E. Africa	Blundell 1987
Asteral	es					
	Campa	anulaceae				
		Lobelia excelsa Bonpl.	Orange to red	(+)A	Chile	Weiss 1995
Brassica	ales					
	Brassic	aceae				
		Erysimum bicolor (Hornem.) DC.	White to purple	(+) A	Canary Islands	Weiss 1995
		Erysimum scoparium (Brouss. ex Willd.) Wettst.	White to purple	(+) A	Canary Islands	Weiss 1995
		Erysimum mutabile Boiss. & Heldr.	White to lavender	(+) A	Canary Islands	Bramwell & Bramwell 1984
		Iberis linifolia L.	White to lavender	(+) A	Spain	Weiss 1995
		Lobularia maritima (L.) Desv.	White to purple	(+) A*	Medit.	Weiss 1995

**Table 1.1.** Angiosperms that display color change along floral development. Adapted and updated from Weiss 1995.

Table 1.1. (Continued)

		Taxon <sup>1</sup>	Color shange <sup>2</sup>	Diamont involved <sup>3</sup>	Nativo babitat <sup>4</sup>	Poforonco
Order	Family	Species	- Color change-	Pigment involved <sup>®</sup>	Native habitat	Reference
Caryop	hyllales					
	Aizoaceae	2				
	Са	arpobrotus edulis (L.) N.E.Br.	Pale yellow to pink	(+) B*	S. Africa	Weiss 1995
	Cactaceae					
	Oj	<i>puntia phaeacantha</i> Engelm.	Yellow to orange	(+) B	Arizona	Weiss 1995
	Polygonac	ceae				
	Pc	olygonum emondi Meisn.	Red to whitish	(-) B	Himalayas	Weiss 1995
Dipsac	ales					
	Caprifoliad	ceae				
	Di	iervilla lonicera Mill.	Yellow to orange	(+)A, (+) C*	E. North America	Schoen 1977
	Lo	onicera hildebrandiana Collett & Hemsl.	White to yellow-orange	(+)A, (+) C	Burma	Weiss 1995
	Lo	onicera japonica Thunb.	White to yellow	(+)A, (+) C	Japan	Weiss 1995
	Lo La	onicera tatarica var. morrowii (A. Gray) Q. E. Yang, andrein, Borosova & J. Osborne	White to yellow	(+)A, (+) C	Japan	Niering & Olmstead 1979
	Lo	onicera periclymenum L.	White to yellow	(+)A, (+) C	Europe and Medit.	Weiss 1995
	Lo	onicera tatarica L.	White to yellow	(+)A, (+) C	Europe	Niering & Olmstead 1979
	Pa	atrinia villosa Juss.	Yellow to white	(+)A, (-) C	Old World	Weiss 1995
	W	<i>leigela decora</i> (Nakai) Nakai	Pale yellow to pink	(+)A, (+) C	Japan	Weiss 1995
Ericales	5					
	Ericaceae					
	Br	rachyloma preissii Sond.	White to red	(+) A	Australia	Erickson <i>et al</i> . 1973
	Polemonia	aceae				
	Са	obaea scandens Cav.	Whitish to purple	(+) A	Tropical America	Proctor & Proctor 1978
Fabales	S					
	Fabaceae					
	Ar	<i>rgyrolobium fischeri</i> Taub.	Bright yellow to red	(+) A	E. Africa	Blundell 1987
	Вс	auhinia bidentata Jack	Yellow to red	(+)A, (+) C	Thailand	Larsen <i>et al.</i> 1984
	Са	adia purpúrea (G.Piccioli) Aiton	White to dark pink	(+) A	Saudi Arabia	Collenette 1985

Table 1.1. (Continued)

	Taxon <sup>1</sup>	- Color change <sup>2</sup> Diamont involved <sup>3</sup>	Nativo babitat <sup>4</sup>	Poforonco		
Order Fam	ily Species		Pigment involveu*	Native habitat	Reference	
Faba	aceae					
	Castanospermum australe A.Cunn & C.Fraser ex Hook	Yellow to red	(+) A	Australia	Macoboy 1979	
	Colutea multiflora Ali	Yellow to pink-red	(+) A	Nepal	Polunin & Stainton 1984	
	Dichrostachys cinerea (L.) Wight & Arn.	Pink to white	(-) A	S. Africa	Van & Malan 1988	
	<i>Intsia bijuga</i> (Colebr.) Kuntze	White to pink	(+)A, (+) C	Seychelles	Friedmann 1986	
	Lotononis eriantha Benth.	Yellow to orange-red	(+) A, (+) C	South Africa	Van & Malan 1988	
	Lotononis laxa Eckl. & Zeyh	Yellow to orange	(+) C	South Africa	Van & Malan 1988	
	Lotus corniculatus L.	Yellow to deep orange	(+) A, (+) C	Europe, Asia	Jones & Cruzan 1982	
	Syrmatium tomentosum (Hook. & Arn.) Vogel	Yellow to red-orange	(+) A, (+) C	California	Weiss 1995	
	Syrmatium glabrumVogel	Yellow to deep orange	(+) C	California	Weiss 1995	
	Lotus pedunculatus	Yellow to red	(+) A	California	Weiss 1995	
	Mimosa rubicaulis Lam.	Pink to white	(-) A	Himalayas	Polunin & Stainton 1984	
	Pearsonia sessilifolia (Harv.) Dummer	Yellow to red	(+) A	South Africa	Van & Malan 1988	
	Psoralea esculenta Pursh	Lavender to pale yellow	(-) C	South Africa	Freeman & Schofield 1991	
	Saraca declinata Miq.	Pale yellow to red	(+)A, (+) C	Thailand	Larsen <i>et al</i> . 1984	
	Saraca indica L.	Pale orange to red	(+)A, (+) C	India, Malaysia	Macaboy 1982	
	Saraca thaipingensis Prain	Yellow to deep orange-red	(+)A, (+) C	Malaysia	Polunin 1987	
	Trifolium stellatum L.	White to deep pink	(+) A	Europe	Weiss 1995	
Poly	galaceae					
	Polygala cruciata L.	White to pink		SE. United States	Duncan & Foote 1975	
	Polygala curtissii A. Gray	White and yellow to pink		SE. United States	Duncan & Foote 1975	
Gentianales						
Rub	iaceae					
	Canthium lucidum R.Br.	White to yellow	(+) C	Australia	Brock 1988	
	Glionnetia sericea (Baker) Tirveng.	White to dark red	(+) A	Seychelles	Friedmann 1986	
	Hamelia patens Jacq.	Orange to red	(+) A	Costa Rica	Lackie <i>et al</i> . 1986	
	<i>Ixora javanica</i> (Blume) DC.	Yellow-orange to red	(+) A	SE. Asia	Polunin 1987	

Table 1.1. (Continued)

Taxon <sup>1</sup>		Color change <sup>2</sup>	Pigment involved <sup>3</sup>	Nativo habitat <sup>4</sup>	Deference
Order	Family Species	Color change	Pigment involved*	Native habitat	Reference
	Apocynaceae				
	Glossonema revoilii Franch.	White to yellow-bronze	(+) C, pH	E. Africa	Blundell 1987
	Oxypetalum coeruleum (D. Don ex Sweet) D	Pecne. Pale blue to pink	(+) C, pH	Brazil, Uruguay	Weiss 1995
Lamiale	les				
	Acanthaceae				
	Asystasia gangetica (L.) T.Anderson	Pale yellow to lavender	(+)A	India	Gracie 1993
	Avicennia marina (Forssk.) Vierh.	Yellow to orange	(+)A, (-) C*	Australia	Brock 1988
	Crossandra johanninae Fiori	White to lavender	(+)A	Saudi Arabia	Collenette 1985
	Bignoniaceae				
	Mansoa hymenaea (DC.) A.H.Gentry	Purple to white	(-) A	Central America	Barrows 1977
	Boraginaceae				
	Arnebia euchroma (Royle) I.M.Johnst.	Pink to deep purple	(+) A, pH	Himalayas	Weiss 1995
	<i>Cryptantha flava</i> (A.Nelson) Payson	White to yellow	(+) C	SW. United States	Welsh & Ratcliffe 1986
	Heliotropium arbainense Fresen.	Pale cream to deep yellow	(+) C	Saudi Arabia	Collenette 1985
	Pulmonaria officinalis L.	Red to blue	рН	Europe	Müller 1883
	Scrophulariaceae				
	Buddleja marrubiifolia Benth.	Yellow to red	(+) A	Chihuahuan desert	Weiss 1995
	Plantaginaceae				
	Hebe townsonii Cheeseman	Purple to white	(-) A	New Zealand	Weiss 1995
	Verbenaceae				
	Lantana camara L.	Yellow to red	(+)A, (-) C*	W. Indies	Mathur & Mohan Ram 1978
	Lantana viburnoides (Forssk.) Vahl	Yellow to deep pink	(+)A, (-) C*	Saudi Arabia	Collenette 1985
Liliales	5				
	Colchicaceae				
	Sandersonia aurantiacaHook.	Pale yellow to deep yellow	(+) C*	S. Africa	Nilsen <i>et al</i> 2003
	Liliaceae				
	Lilium brownii F.E.Br. ex Miellez	Yellow to white	(-) C*	China	Hai <i>et al</i> 2012

Table 1.1. (Continued)

	Taxon <sup>1</sup>		Color change <sup>2</sup>	Diamont involved <sup>3</sup>	Nativo habitat <sup>4</sup>	Poforonco
Order	Family	Species	color change	riginent involved	Native Habitat	Kelerence
	Melant	hiaceae				
		<i>Trillium ovatum</i> Pursh	White to deep pink	(+) A	W. United States	Weiss 1995
		Zigadenus glaberrimus Michx.	White to pale yellow	(+) C	Mississippi	Timme 1989
Malphi	ghialies					
	Malpig	hiaceae				
		<i>Byrsonima crassifolia</i> (L.) Kunth	Yellow to orange	(+) A	Costa Rica	Weiss 1995
		Byrsonima microphylla A.Juss.	Yellow to red	(+) A	Brazil	Costa <i>et al</i> . 2006
		Byrsonima gardnerana A.Juss.	Yellow to red	(+) A	Brazil	Costa <i>et al</i> . 2006
		Heteropterys alternifolia W.R. Anderson	Yellow to red	(+) A	Brazil	Costa <i>et al</i> . 2006
		Tetrapterys phlomoides (Spreng.) Nied.	Yellow to orange	(+) A	Brazil	Weiss 1995
	Passifle	praceae				
		Passiflora sp.	White to yellow	(+) C	New Zealand	Weiss 1995
	Rhizop	horaceae				
		Ceriops tagal (Perr.) C.B.Rob.	White to orange	(+) A	N. Australia	Brock 1988
	Violace	eae				
		Viola cornutaL.	White to purple	(+)A*		Farzad et al. 2002
Malval	es					
	Malvac	eae				
		Brachychiton discolor F.Muell.	White to purple	(+) A	Australia	Macaboy 1982
		Gossypium barbadense L.	Cream to pink	(+) A*		Tan <i>et al</i> . 2013
		Gossypium hirsutum L.	Cream to pink	(+) A*		Weiss 1995; Tan <i>et al</i> . 2013
		Hibiscus mutabilis L.	Yellow to orange-red	(+) A *	China	Chin 1977; Shanker & Vankar 2007
		Hibiscus tilliaceous L.	Yellow to orange-red	(+) A	Pantropical	Weiss 1995
		Lasiopetalum behrii	Pale yellow to pink	(+) A	Australia	Costermans 1981
		Malvaviscus arboreus Cav.	Red to pink	(-) A*	America	Gottsberger 1971

Table 1.1. (Continued)

		Taxon <sup>1</sup>	- Color change <sup>2</sup>	Diamont involved <sup>3</sup>	Nativo habitat <sup>4</sup>	Deference
Order	Family	Species		Pigment involved*	Native habitat	Kelefence
	Malva	ceae				
		Thespesia populnea (L.) Sol. ex Corrêa	Yellow to purple	(+) A*	Pantropical	Mabberley 1987; Lowry 1976
	Thyme	laeaceae				
		<i>Edgeworthia tomentosa</i> (Thunb.) Nakai	Yellow to white	(-) A, (-) C	China	Macoboy 1984
		Pimelea ferruginea Labill.	Pink to white	(-) A, (-) C	W. Australia	Weiss 1995
Myrtal	es					
	Combr	retaceae				
		Combretum farinosum Kunth	Green to orange	(+) A	S. and C. America	Schemske 1980
		Combretum indicum (L.) DeFilipps	White to deep red	(+) A	Burma	Weiss 1995
	Melast	omataceae				
		Tibouchina lepidota (Bonpl.) Baill.	Magenta to purple	(+) A*	Ecuador	Weiss 1995; Hendra & Keller, 2016
		<i>Tibouchina pulchra</i> (Cham.) Cogn.	White to dark pink	(+) A	Brazil	Brito <i>et al</i> . 2015
		Tibouchina sellowiana (Cham.) Cong.	White to red	(+) A	Brazil	Ludwing 1886
	Myrtad	ceae				
		Chamelaucium megalopetalum F.Muell. ex Benth.	White to red	(+) A	Australia	Erickson <i>et al</i> . 1973
		Darwinia citriodora (Endl.) Benth.	Yellow to red	(+) A	Australia	Lamont 1985
		Darwinia fascicularis Rudge	White to red	(+) A	Australia	Lamont 1985
		Hypocalymma angustifolium (Endl.) Schauer	White to deep pink	(+) A	Australia	Lamont 1985
		Myrtella sp.	White to pink	(+) A	Australia	Brock 1988
		Verticordia acerosa Lindl.	Yellow to orange-red	(+) A	Australia	Lamont 1985
		Verticordia chrysantha Endl.	Yellow to red	(+) A	Australia	Lamont 1985
		Verticordia grandiflora Endl.	Yellow to orange	(+) A	Australia	Erickson <i>et al</i> . 1973
		Verticordia huegelii Endl.	White to red	(+) A	Australia	Lamont 1985
	Onagr	aceae				
		Fuchsia excorticata (G.Forst.) L.f.	Green and black to deep red	(+) A*	New Zealand	Delph & Lively 1989; Webby & Bloor 2000

Table 1.1. (Continued)

	Taxon <sup>1</sup>	- Color change <sup>2</sup>	Pigmont involved <sup>3</sup>	Native habitat <sup>4</sup>	Poforonco
Order Family	Species	Color change	Fightent involved	Native Habitat	Kelerence
Onagra	aceae				
	Gaura coccinea Nutt. ex Pursh	White to maroon	(+) A	SW. United States	Welsh & Ratcliffe 1986
	Hauya sp.	White to deep pink	(+) A*	C. America	Averett & Raven 1984; Weiss 1995
	Oenothera epilobiifolia Kunth	Green to orange-red	(+) A	Venezuela	Weiss 1995
Vochys	siaceae				
	<i>Qualea multiflora</i> Mart.	White to pale yellow	(+) C	Brazil	Weiss 1995
Nymphaeales					
Nympł	naeaceae				
	Victoria amazonica (Poepp.) J.C. Sowerby	White to pale purple	(+) A	S. America	Weiss 1995
Oxialidales					
Elaeoc	arpaceae				
	Aristotelia fruticosa Hook.f.	White to red	(+) A	New Zealand	Salmon 1968
	Aristotelia serrata (J.R.Forst. & G.Forst.) Oliv.	White to red	(+) A	New Zealand	Weiss 1995
Poales					
Brome	liaceae				
	Aechmea sp.	Pink to blue	рН	Tropical America	Weiss 1995
Proteales					
Protea	ceae		( ) <b>.</b>		
	Banksia ilicifolia R.Br.	Yellow to red	(+) A	Australia	Lamont & Collins 1988
	Leucospermum oleaefolium R. Br.	Yellow to orange-red	(+) A	S. Africa	Van Der Spuy 1971
Rosales					
Rhamr	naceae			<b>A</b>	W////: 1070
2	Cryptandra amara Sm.	White to red	(+) A	Australia	Williams 1979
Kosace					
	Rosa chinensis f. mutabilis (Correvon) Rehder	Yellow to deep pink	(+) A*	China	Weiss 1995; Cai <i>et al</i> . 2005

Table 1.1. (Continued)

Taxon <sup>1</sup>		Taxon <sup>1</sup>	- Color change <sup>2</sup>	Pigmont involved <sup>3</sup>	Native babitat <sup>4</sup>	Poforonco
Order	Family	Species		Fightent involved	Native Habitat	Reference
Santala	ales					
	Lorant	haceae				
		Phragmanthera dshallensis (Engl.) M.G.Gilbert	Yellow to red	(+) A	E. Africa	Gill & Wolf 1975
		<i>Phragmanthera regularis</i> (Steud. ex Sprague) M.G. Gilbert	Yellow to red	(+) A	E. Africa	Blundell 1987
	Santala	aceae				
		Santalum freycinetianum F. Phil.	Greenish-yellow to red	(+) A*	Hawaii	Sohmer & Gustafson 1987
		Santalum haleakalae Hillebr.	White to red	(+) A	Hawaii	Sohmer & Gustafson 1987
		Santalum album L.	White to red	(+) A	Australia	Erickson <i>et al</i> . 1973; Harsha <i>et al</i> . 2013
	Schoep	ofiaceae				
		Quinchamalium chilense Molina	Yellow to orange	(+) A*	Andes	Riveros <i>et al</i> . 1987; Simirgiotis <i>et al</i> . 2012
Sapind	ales					
	Anacar	rdiaceae				
		Anacardium occidentale L.	White to pink	(+) A	Tropical America	Weiss 1995
	Rutace	ae				
		Boronia molloyae J.R.Drumm.	Magenta to deep red	(+) A	Old World	Weiss 1995
		Boronia polygalifolia Sm.	Pale yellow to pinkish	(+) A	Australia	Williams 1979.
Saxifra	gales					
	Paeoni	aceae				
		Paeonia ostii T.Hong & J.X.Zhang	White to pink	(+) A*		Li <i>et al.</i> 2011
	Saxifra	gaceae				
		<i>Tellima grandiflora</i> (Pursh) Douglas ex Lindl.	White to red	(+) A	W. United States	Weiss 1995
Solanal	les					
	Solana	ceae				
		Brugmansia versicolor Lagerh.	Cream to pink	(+) A	South America	Weiss 1995
		Brunfelsia calycina Benth.	Purple to white	(-) A*	Tropical America	Weiss 1995

 Table 1.1. (Continued)

Taxon <sup>1</sup>		Color change <sup>2</sup>	Diamont involved <sup>3</sup>	Nativo habitat <sup>4</sup>	Poforonco	
Order	Family	Species		Fightent involved	Native Habitat	Kelelence
Solanaceae						
	Solanum diploconos (N	Mart.) Bohs	Violet to yellow-ochre	(-) A, (+) C	Tropical America	Sazima <i>et al</i> . 1993
Solanum endopogon (Bitter) Bohs		Lavender to greenish-white	(-) A	Tropical America	Gracie 1993	
	Solanum sciadostylis (	Sendtn.) Bohs	Violet to yellow-ochre	(-) A, (+) C	Tropical America	Sazima <i>et al</i> . 1993
Nicotiana mutabilis Stehmann & Semir		White to pink	(+) A*	Australia	Stehmann <i>et al</i> . 2002	
	Solandra sp.		White to yellow	(+) C	Tropical America	Van Der Spuy 1971
	Streptosolen jamesoni	i (Benth.) Miers	Yellow to deep orange	(+) C	Peru	Weiss 1995

<sup>1</sup> Taxa for which the mechanism of color change was reported

<sup>2</sup> Color change described for sepals, petals or both.

<sup>3</sup> Mechanism of flower color change, pH or pigments. (+) A: anthocyanin accumulation, (-) A: anthocyanin degradation; (+) C: carotenoid accumulation, (-) C: carotenoid degradation, (+) B: betalain accumulation, (-) B: betalain degradation, N.: North, S.: South, E: East, W: West, SE.: Southeast, SW.: Southwest., Medt.: Mediterranean. \* Asterisks indicates that the pigment was experimentally addressed, in other cases, the mechanism was inferred by Weiss, 1995.

<sup>4</sup> N.: North, S.: South, E: East, W: West, SE.: Southeast, SW.: Southwest., Medt.: Mediterranean.

# 2. OBJECTIVES

The overall aim of this thesis was to explore the phenomenon of flower color change in *Tibouchina pulchra* by the chemical characterization of flower extracts and the study of the regulatory mechanisms involved in this temporal phenotypic change. Results were organized in the following to chapters:

# Chapter 2: Chemical characterization of Tibouchina Pulchra (Cham.) Cogn. flowers.

Chapter two describes the chemical characterization of petal pigments by isolation procedures and spectroscopic analysis (UV-Vis absorption spectra, mass spectra and nuclear magnetic resonance).

# <u>Chapter 3</u>: The Regulation of color change in *Tibouchina pulchra* flowers.

In this chapter, the goal was to investigate whether the color change is determined either by the differential accumulation of flavonoid derivatives and/or metals along the different stages of floral development. In order to investigate the putative transcriptional regulation of the biosynthetic enzyme encoding genes, key genes of the flavonoid biosynthesis were cloned and the transcript accumulation patterns profiled along floral development.

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# Chapter 2 – Chemical characterization of *Tibouchina pulchra* (Cham.) Cogn. flowers.

### 1. INTRODUCTION

*Tibouchina* Aubl. is one of the most representative genera within Melastomataceae A. Juss., the fifth largest angiosperm family in Brazil (BFG 2015). This family contains approximately 170 genera and 4,500 species with a pantropical distribution, but particularly rich and dominant in the Neotropics with about 3,000 species, being 1,367 Brazilian natives. Melastomataceae species can be easily recognized among eudicots by their leaves with characteristic acrodromous venation. It contains thirteen well differentiated tribes (APG 2016).

Recently, a study that addressed the reproductive strategies within three tribes (i.e. Miconieae, Microlicieae and Melastomeae) found that Miconieae encompasses pollinator independent and fleshy fruit bearing species, mainly bird dispersed. On the other hand, in Microlicieae and Melastomeae, although the dispersion is abiotic (dry fruit), the breeding strategy depends on pollinators, mainly bees (Brito et al. 2017). By means nuclear and plastid markers, a molecular phylogenetic analysis demonstrated that *Tibouchina*, the largest genus in Melastomeae, is polyphyletic once some species grouped together with other genera, such as Brachyotum, Bucquetia, Castratella, Centradenia, Chaetolepis, Heterocentron, Itatiaia, Microlepis, Monochaetum, Pilocosta, Svitramia, and Tibouchinopsis, which are all monophyletic (Michelangeli et al. 2013). Moreover, Tibouchina has several plesiomorphic characters as the core Melastomeae: tetramerous or pentamerous flowers that show anthers with developed pedoconnectives (i.e. the connective tissue below the anther locules) and short bilobed ventral anther appendages. These data clearly expose that *Tibouchina* phylogeny is unsolved and it has been proposed that either the genus is recognized as much-expanded, although no morphological synapomorphies for this clade were identified, or all the other genera could be maintained, while Tibouchina should be divided into smaller and diagnosable units (Michelangeli et al. 2013).

*Tibouchina* has approximately 460 species (Clausing & Renner 2001, Tropicos 2017, Guimarães 2014). High level of endemism was verified for Brazilian species, being 145 out of the 166 reported considered restricted to the country, occurring in Atlantic rainforest and Cerrado (BFG 2015, Guimarães 2016). Studies aiming to describe the biological activities of

*Tibouchina* species have shown promising results. Jiménez *et al.* (2015) reported the use of *Tibuchina kingii* Wurdack in traditional medicine due to its anti-inflammatory properties and demonstrated high antioxidant potential of the aqueous leaf extract. *T. pereirae* Brade & Markgr also exhibited antioxidant activity by free radical scavenging assay and antinociceptive effect of *n*-hexanic extract (Dias *et al.* 2016). Low antimicrobial activity was found in extracts of *Tibouchina candolleana* Cong. (Santos *et al.* 2012). Additionally, isolated compounds (*i.e.* 2,8-dihydroxy-7H-furo[2,3-f] chromen-7-one and isoquercitrin) from *Tibouchina paratropica* (Grised.) Cong. showed antibacterial and antifungal activity. Interestingly, 2,8-dihydroxy-7H-furo[2,3-f] chromen-7-one also displayed potent antiparasitic activity against *Leishmania donovani* (Tracanna 2015). Antifungal (Niño *et al.* 2003, Kuster *et al.* 2009), anticancer (Jones *et al.* 1981) and antiparasitic activities (Singha *et al.* 1992, Mpalantinos *et al.* 1998, Cunha *et al.* 2009) have also been reported for the genus.

Phytochemical studies in eleven species of *Tibouchina* (*T. candolleana, T. ciliares* (Vent.) Cong., *T. grandiflora* Cong., *T. lepidota* (Bonpl.) Baill., *T. multiflora* Cong., *T. paratropica, T. pereirae, T. pulchra, T. semidecantra* (Mart & Schrank ex DC.) Cong., *T. urvilleana* (DC.) Cong., *T. granulosa* (Desr.) Cong.) reported the presence of several natural products, such as flavonoids (flavonol glycosides and anthocyanins), isoflavonoids, phenolic derivatives, tannins and triterpenes (Table 2.1). Structural elucidation by Nuclear Magnetic Resonance (NMR) was performed for some triterpenoids, tannins, flavonols, and anthocyanins in *T. urvilleana* and *T. lepidota* (Terahara *et al.* 1993, Pérez-Castorena 2014, Hendra & Keller 2016). As reviewed above, few studies have addressed the chemical profile of *Tibouchina* species, which remains largely elusive.

Species	Class of metabolite	Compound	Organ	Reference
T. candolleana	triterpene/ flavone/ isoflavoid	$\alpha\text{-}$ and $\beta\text{-}amyrin,$ $\beta\text{-}$ sitosterol, ursolic and oleanolic acids/ luteolin/ genistein	aerial parts	Dos Santos <i>et al.</i> 2012
T. ciliaris	flavonol	kaempferol 7- <i>O-p</i> -coumaroyl, quercetin 3-O-rhamnopyranoside, quercetin 3-O-arabnoside	leaves	Colorado <i>et al.</i> 2007
T. grandiflora	flavonol/ anthocyanin	quercetin 3-O- $\beta$ -D-glucuronide, quercetin 3-O- $\beta$ -D-glucopyranoside, quercetin 3-O- $\beta$ -D-galactopyranoside, quercetin 3-O- $\alpha$ -L- rhamnopyranoside, quercetin 3-O- $\beta$ -L-arabinopyranoside, quercetin 3- O- $\beta$ -D-(6"- <i>p</i> -coumaroyl)-glucopyranoside/ peonidin 3-sophoroside, peonidin 3-sambubioside, malvidin 3,5-diglucose, malvidin 3-( <i>p</i> - coumaroyl)-sambubioside-5-glucoside	leaves/ flowers	Kuster <i>et al</i> . 2009, Bobbio <i>et al</i> . 1985
T. granulosa	flavonol/ flavone/ proanthocyanidin/ anthocyanin	Isorhamnetin 3-O-glucuronide, isorhamnetin 3-O-di-glucoside, isorhamnetin 3-O-rutinoside, quercetin 3-(O-galloyl)-hexoside/ hispidulin 7-O-glucoside/ B-type procyanidin monomer, dimer, trimer and pentamer/ petunidin, pelargonidin	flowers/ leaves	Okumura <i>et al.</i> 2002, Sobrinho <i>et</i> <i>al</i> . 2017
T. lepidota	tannin/ flavonol/ anthocyanin	gallic acid, 2,3,5-trihydroxybenzoic acid/ quercetin 3-O-arabnoside, quercetin, quercetin 3-glucoside, isorhamnetin 3-rutinoside/ malvidin 3- ( <i>p</i> -coumaryl-glucoside)-5-(acetyl-xyloside)	flowers	Hendra & Keller 2016
T. multiflora	tannin	nobotanins O and P	leaves	Yoshida <i>et al</i> .1999,
T. paratropica	phenolic derivative/ flavonoL	2,8-dihydroxy-7H-furo (2,3-f)-chromen-7-one/ isoquercitrin	aerial parts	Tracanna <i>et al.</i> 2015
T. pereirae	flavonol	Isorhamnetin 3-O-galloyl-glucoside, isorhamnetin-3-O-glucoside, kaempferol 3-O-rutinoside, quercetin 3-O-(galloyl)-glucoside, rutin	aerial parts	Dias 2013
T. pulchra	flavonol/ tannin	kaempferol 3-O-galactoside, kaempferol 3-O-glucoside, myricetin 3-O- galactoside, myricetin 3-O-glucoside, quercetin, myricetin, kaempferol, luteolin/ gallic acid	leaves	Furlan 2004, Motta <i>et al</i> . 2005

# **Table 2.1**- Natural products reported for *Tibouchina* species.

<b>Table 2.1</b> . (C	ontinued)
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Species	Class of metabolite	Compound	Organ	Reference
T. semidecantra	tannin/ flavonol/ anthocyanin/ proanthocyanidin	1,2,6-tri-O-galloyl-β-D-glucose, 1,4,6-tri-O-galloyl-β-D-glucose, 1,2,3,6-tetra-O-galloyl-β-D-glucose, nobotanin A, B, C, D, E and F, casuarictin, pedunculagin, praecoxin A and B, casuarinin, 2,3-O-(S)-hexahydroxydiphenoyl-D-glucopyranose, castalagin, vescalagin, 1-O-methylvescalaginnobotanins A, B, F, 3,3'-O-dimethyl ellagic acid 4-O-α-L-rhamnopyranoside/ quercetin, myricetin, quercetin 3-O-(6''-O-galloyl) galactoside, quercetin 3-O-α-L-(2''-O-acetyl) arabinofuranoside, quercetin 3-O-arabnoside, quercetin 3-O-rhamnopyranoside/ malvidin 3-( <i>p</i> -coumaroylglucoside)-5-glucoside/ leucodelphinidin, leucocyanidin	aerial parts	Sirat <i>et al</i> . 2010, Yoshida <i>et al</i> . 1991, Harbone 1964, Lowry 1975
T. urvilleana	flavonol/ flavone/ anthocyanin/ triterpene	quercetin 3-O-arabnoside/ hispidulin 7-O- $\beta$ -D-glucopyranoside / malvidin 3-O-(6-O-p-coumaryl- $\beta$ -D-glucopyranoside)-5-O-(2-O-acetyl- $\beta$ -D-xylopyranosyl)/ glutinol, taraxerol, $\alpha$ - and $\beta$ -amyrin, $\beta$ - sitosterol, ursolic and oleanolic acids	aerial parts	Perez-Castorena 2014, Terahara <i>et</i> <i>al</i> . 1993.

*T. granulosa* and *T. pulchra* have been characterized as possible biomonitors of air pollution, *i.e.* particulate matter and ozone (Klumpp *et al.* 2001, Furlan *et al.* 2004, Furlan *et al.* 2008, Santos & Furlan 2013, Zampieri *et al.* 2013, Espósito & Domingos 2014, Espósito & Domingos 2016, Pedroso *et al.* 2016). Moreover, as they occur mainly in open areas, such as forest edges and clearings, *Tibouchina* species are considered important for restoration/reforestation purposes (Ellison *et al.* 1993).

*Tibouchina pulchra* (Figure 2.1), popularly known as "manacá-da-serra", has been increasingly used for urban ornamentation and, nowadays several cultivars are available in the flower market. The beauty and fascinating feature of these flowers is the development-associated color change: the buds open as white flowers that the next day become pink, further, from the third day towards senescent, the flowers maintain a deep pink color. The well-determined stages of flower coloring in *T. pulchra* raise interesting questions about the genetic and chemical mechanisms involved in the regulation of this phenomenon, which has up to now been unexplored (see Chapter 3).

In this context, this chapter presents results of isolation and identification of the petal tissue pigments of *Tibouchina pulchra*.



Figure 2.1. Tibouchina pulchra tree and its flowers.

#### 2. MATERIAL AND METHODS

#### 2.1. Plant Material

Pink petals of *Tibouchina pulchra* (cv. "manacá-anão") flowers were sampled at Praça Carlos José Gíglio (Latitude: -23.57998, Longitude: -46.73403) in the most vigorous flowering period (May and June/2016) between 8 and 9 a.m. Petals were immediately frozen in liquid nitrogen and stored at -80°C until processing. Samples were freeze dried in a lyophilizer (K202, Liobras) and crushed in a ball mill (TissueLyser, Qiagen) for further analyses. An exsiccate was deposited on herbarium of Institute of Bioscience (SPF) of University of São Paulo (ID: Furlan73).

#### 2.2. Extraction and analysis by UPLC-MS

Phenolic compounds were extracted from 100 mg of petal powder with two times of 1.5 mL of 0.2 % HCl in methanol (MeOH). The samples were sonicated for 10 min and centrifugated at 10,000 rpm for 10 min. The extract was filtered (PTFE 0.45 µm) and analyzed by Ultra Performance Liquid Chromatography (UPLC) system with Diode Array Detector (DAD) (Dionex Ultimate 3000) and Electrospray Ionization Quadrupole Time-of-Flight High-Resolution Mass Spectrometry (ESI-QTOF-HRMS) detector (Bruker Maxis), MS/MS analysis was performed with a Broadband Collision Induced Dissociation (bbCID) detector. Separation was achieved by using a C18 column (Waters Acquity UPLC 100 x 2.1 mm - 1.7 µm) at a flow rate of 0.3 mL min<sup>-1</sup>, and 4 µL of injection volume, the column temperature was 45°C, and the solvent system composed of 1% formic acid in water (A) and 1% formic acid in acetonitrile (B). Gradient elution: 5 to 25% of B (0- 40 min), 25 to 100% of B (40- 42 min), 100% of B (42- 42.5 min), 100 to 5% of B (42.5- 43 min), 5% of B (43- 46 min). Separated compounds were first monitored using DAD (200 to 600 nm) and then MS scans were performed in positive ion mode (MS<sup>+</sup>), in the range m/z 75-1250 m/z, in the following conditions: capillary voltage set to 4,500 V, end plate offset at -500 V, nebulizer at 2 Bar, dry gas 12 L min<sup>-1</sup> and dry gas temperature at 200°C MS was calibrated using sodium formate. All data were processed using Data analysis software 4.2 (Bruker). This analysis was performed in the Department of Chemistry at Denmark Technical University under the supervision of Ph.D. Mads H. Clausen.

# 2.3. Isolation by preparative HPLC and identification by Nuclear Magnetic Resonance (NMR)

Also in partnership with Ph.D. Mads H. Clausen, flavonoids were isolated from 10 g of petal powder by extracting four times in 200 mL of hydrochloric acid 0.2% in MeOH. The samples were sonicated for 15 min, pillowed for 10 min, vacuum filtered and concentrated with a rotary evaporator. The crude extract was diluted to, approximately, 250 mg/mL and injected in a preparative High Performance Liquid Chromatography (HPLC- Waters) with Photodiode Array Detector (PAD). Separation was achieved on a C18 column (Waters X Bridge BEH ODB 30 x 150 mm, 5µm) at a flow rate of 20 mL min<sup>-1</sup>, using 1 mL of injection volume, and a solvent system composed of 1% formic acid in water (A) and 1% formic acid in acetonitrile (B). Gradient elution: 10% of B (0- 3 min), 10 to 15% of B (3- 30 min), 15% of B (30- 50 min), 15 to 20% of B (50- 60 min), 20 to 25% of B (60- 80 min), 25 to 35% of B (80- 90 min), 35 to 45% of B (90- 95 min), 45 to 100% of B (95- 96 min), 100% of B (96- 98 min), 100 to 10% of B (98- 98.5 min), 10% of B (98.5-102 min), and monitored using DAD (200 to 600 nm). Fractions were collected by time. All the fractions were concentrated by rotary evaporator. An aliquot was resuspended in 0.2% HCl in MeOH to check the purity by UPLC-MS. For the isolated compounds the dried sample was dissolved in deuterated dimethylsulfoxide (DMSO-d<sub>6</sub>) for NMR analysis. <sup>1</sup>H and <sup>13</sup>C NMR spectra were obtained by using a AVANCE III HD spectrometer (Bruker) operating at frequency of 800 MHz and equipped with a 5 mm TCI CryoProbe (Bruker). Two-dimensional analysis (HSQC and HMBC) were also performed. All data were processed using MestreNova software 11.0 (Mestrelab Research). The identification of compounds was performed in collaboration with Ph.D. Marcelo José Pena Ferreira from Department of Botany, University of São Paulo.

# 3. RESULTS AND DISCUSSION

#### 3.1. Analysis by UPLC-MS

HPLC analysis showed a high diversity of compounds in petal extract, being integrated 30 constituents (Figure 2.2). UV-Vis absorption spectra and mass fragmentation of each detected constituent allowed the identification of two phenolic classes: phenolic acids (constituents **1** to **6**) and flavonoids (constituents **7** to **30**). Flavonoid compounds showed glucosyl and acyl groups as substituents (Table 2.2 and Figures 2.3 to 2.8).

Since the chromatogram did not exhibit an accurate resolution of the phenolic acids, generating low-quality spectra (Supplemental Figures 2.1 and 2.2), the identification of these compounds was not properly performed.

Regarding flavonoids, the analysis revealed hexosyl, pentosyl, glucuronic acid, galloyl, acetylpentosyl, and *p*-coumaroyl as substituents. Kaempferol, quercetin and myricetin were the detected flavonol aglycones, while malvidin and petunidin were the base skeleton for the identified anthocyanins (Table 2.2).

In many of the obtained mass spectra, it was possible to observe adduct from [M+Na]<sup>+</sup>. Sodium and potassium adduct ions are often detected in first-order mass spectra obtained with ESI in the positive ion mode. These alkali metals are generally extracted from glass during solution storage and are more easily formed in flavonoids substituted at the 3-position (Cuyckens & Claeys 2004).



**Figure 2.2**. Chromatogram obtained by UPLC-ESI-QTOF-HRMS from *T. pulchra* petals extracted with acidified methanol. Separation was performed with a column Waters Acquity UPLC C18 (1.7  $\mu$ m 100 x 2.1 mm) at a flow rate of 0.3 mL min<sup>-1</sup>, using 4  $\mu$ L of injection volume, column temperature of 45°C, and a solvent system composed by 1% formic acid in water (A) and 1% formic acid in acetonitrile (B). Gradient elution: 5 to 25% of B (0- 40 min), 25 to 100% of B (40- 42 min), 100% of B (42- 42.5 min), 100 to 5% of B (42.5- 43 min), 5% of B (43- 46 min). Numbers correspond to the identification presented in Table 2.2.

Compound	RT (min)	UV/ Vis (nm)	Mass spectrum HRMS-MS	Suggestion
1	0.97	278	L.Q.	Phenolic acid
2	1.13	278, (sh) 308	L.Q.	Cinnamic acid derivative
3	1.55	278	L.Q.	Phenolic acid
4	1.76	278	L.Q.	Phenolic acid
5	2.95	278, (sh) 308	L.Q.	Cinnamic acid derivative
6	3.65	278	L.Q.	Phenolic acid
7	12.69	270	453.0083 [M+H] <sup>+</sup> , 303.0134 [M-150] <sup>+</sup>	N.I.
8	14.05	268, 294 (sh), 354	481.0967 [M+H] <sup>+</sup> , 319.0446 [M-162] <sup>+</sup>	Myricetin hexoside
9	14.70	268, 294 (sh), 354	481.0964 [M+H] <sup>+</sup> , 319.0445 [M-162] <sup>+</sup>	Myricetin hexoside
10	16.33	269, 290 (sh), 354	639.0946 [M+Na] <sup>+</sup> , 617.1121 [M+H] <sup>+</sup> , 303.0498 [M-314] <sup>+</sup>	Quercetin galloylhexoside
11	18.25	269, 290 (sh), 355	465.1020 [M+H] <sup>+</sup> , 303.0499 [M-162] <sup>+</sup>	Quercetin hexoside
12	18.89	269, 290 (sh), 355	479.0804 [M+H] <sup>+</sup> , 303.0493 [M-176] <sup>+</sup>	Quercetin glucuronide
13	19.36	266,290,350	601.1183 [M+H] <sup>+</sup> , 287.0552 [M-314] <sup>+</sup>	Kaempferol galloylhexoside
14	19.95	270	453.0083 [M+H] <sup>+</sup> , 303.0134 [M-150] <sup>+</sup>	N.I.
15	21.42	266, 346	471.0894 [M+Na] <sup>+</sup> , 449.1073 [M+H] <sup>+</sup> , 287.0549 [M-162] <sup>+</sup>	Kaempferol hexoside
16	22.00	266,290,350	601.1184 [M+H] <sup>+</sup> , 287.0551 [M-314] <sup>+</sup>	Kaempferol galloylhexoside
17	23.18	266, 348	449.1079 [M+H] <sup>+</sup> , 287.0551 [M-162] <sup>+</sup> / 463.0865 [M+H] <sup>+</sup> , 287.0551 [M-176] <sup>+</sup>	Mixture: Kaempferol 3-O-β-D- glucopyranoside (Astragalin)/ Kaempferol-(2''-O-methyl)-4'-O-α-D- glucopyranoside
18	23.93	266, 355	441.0790 [M+Na] <sup>+</sup> , 419.0971 [M+H] <sup>+</sup> , 287.0551 [M-132] <sup>+</sup>	Kaempferol pentoside
19	24.81	266,290,350	623.1000 [M+Na] <sup>+</sup> , 601.117 [M+H] <sup>+</sup> , 287.0547 [M-314] <sup>+</sup>	Kaempferol galloylhexoside
20	25.98	266, 355	441.0785 [M+Na] <sup>+</sup> , 419.0959 [M+H] <sup>+</sup> , 287.0545 [M-132] <sup>+</sup>	Kaempferol pentoside
21	27.87	268, 314	595.1445 [M+H] <sup>+</sup> , 287.0551 [M-308] <sup>+</sup>	Kaempferol <i>p</i> -coumaroylhexoside

**Table 2.2.** Phenolic constituents from *T. pulchra* petals extracted with acidified methanol and analyzed by UPLC-DAD-ESI-QTOF-HRMS.

 Table 2.2. (Continued)

Compound	RT (min)	UV/ Vis (nm)	Mass spectrum HRMS-MS	Suggestion
22	28.24	282, 305(om), 530	799.2077 [M+H] <sup>+</sup> , 625.1552 {M-174] <sup>+</sup> , 491.1176 [M-308] <sup>+</sup> , 317.0655 [M-482] <sup>+</sup>	Petunidin <i>p</i> -coumaroylhexoside acetylpentoside
23	29.04	268, 320, 530	499.0839 [M+Na] <sup>+</sup> , 477.1031 [M+H] <sup>+</sup> , 287.0547 [M-190] <sup>+</sup> / 771.2138 [M+H] <sup>+</sup> , 317.0665 [M-454] <sup>+</sup>	Mixture- Kaempferol 3-O-glucuronide-6''-O-methylester / Petunidin derivative
24	30.82	282, 310(om), 534	813.2243 [M+H] <sup>+</sup> , 639.1716 [M-174] <sup>+</sup> , 505.1336 [M-308] <sup>+</sup> , 331.0812 [M-482] <sup>+</sup>	Malvidin <i>p</i> -coumaroylhexoside acetylpentoside
25	32.68	271, 312	633.1203 [M+Na] <sup>+</sup> , 611.1393 [M+H] <sup>+</sup> , 303.0496 [M-308] <sup>+</sup>	Quercetin 3-O-(6''-O- <i>p</i> -coumaroyl)-β-D-glucopyranoside (Helichrysoside)
26	34.78	266, 349	593.0892 [M+H] <sup>+</sup> , 285.0603 [M-308] <sup>+</sup>	N.I.
27	35.27	268, 314	617.1258 [M+Na] <sup>+</sup> , 595.1437 [M+H] <sup>+</sup> , 287.0546 [M-308] <sup>+</sup>	Kaempferol 3-O-(6"-O-p-coumaroyl)- $\beta$ -D-glucopyranoside (Tiliroside)
28	36.17	268, 314	617.1256 [M+Na] <sup>+</sup> , 595.1418 [M+H] <sup>+</sup> , 287.0545 [M-308] <sup>+</sup>	Kaempferol p-coumaroylhexoside
29	37.48	270, 368	287.0546 [M+H]+	Kaempferol
30	37.98	268, 314	617.1248 [M+Na] <sup>+</sup> , 595.1455 [M+H] <sup>+</sup> , 287.0549 [M-308] <sup>+</sup>	Kaempferol <i>p</i> -coumaroylhexoside

\*RT (min): retention time in minutes, sh: shoulder, N.I: not identified, L.Q: low quality. Numbers highlighted in bold indicate compounds reported for the first time in *Tibouchina*.



Figure 2.3. Mass spectra (MS<sup>+</sup>) of compounds 7, 8, 9 and 10. Structures represent the proposed compound and its main fragmentation.



Figure 2.4. Mass spectra (MS<sup>+</sup>) of compounds 11, 12, 13 and 14. Structures represent the proposed compound and its main fragmentation.



Figure 2.5. Mass spectra (MS<sup>+</sup>) of compounds 15, 16, 17 and 18. Structures represent the proposed compound and its main fragmentation.



Figure 2.6. Mass spectra (MS<sup>+</sup>) of compounds 19, 20, 21 and 22. Structures represent the proposed compound and its main fragmentation.



Figure 2.7. Mass spectra (MS<sup>+</sup>) of compounds 23, 24, 25 and 26. Structures represent the proposed compound and its main fragmentation.


Figure 2.8. Mass spectra (MS<sup>+</sup>) of compounds 27, 28, 29 and 30. Structures represent the proposed compound and its main fragmentation



kaempferol p-coumaroylhexoside (21, 28, and 30)



petunidin *p*-coumaroylhexoside acetylpentoside (22)



malvidin p-coumaroylhexoside acetylpentoside (24)



quercetin glucuronide (12)



kaempferol hexoside (15)



myricetin hexoside (8 and 9)



quercetin galloylhexoside (10)



kaempferol pentoside (18 and 20)



quercetin hexoside (11)

Figure 2.9. Compounds of *T. pulchra* flowers identified by UV-visible absorption and mass spectra.

From the twenty-four flavonoids detected, twenty-one were identified by MS and UV-Vis analyses (Figure 2.9), and eight of these compounds were isolated for the accurate identification by NMR analysis. The most abundant flavonol found was kaempferol (m/z287.0549, constituents **13**, **15** to **21**, **23**, and **27** to **30**) with different substituents, besides three isomers of kaempferol galloylhexoside (constituents **13**, **16**, and **19**). Quercetin (m/z 303.0498, constituents **10**, **11**, **12**, and **25**) was the second most abundant flavonol found, followed by myricetin (m/z 319.0445, constituents **8** and **9**). Flavonols as kaempferol, myricetin, and quercetin derivatives were previously described in leaves of *T. pulchra* corroborating the present study (Furlan 2004, Motta *et al.* 2005).

The fragment m/z 153.0181 was observed in many constituents (**7**, **10**, **13**, **14**, **16**, **19**, **21**, and **26**). It is a typical ion signal from a fragment of A-ring, generated by Retro Diels–Alder cleavage of the C–ring bonds, resulting in structurally informative <sup>i,j</sup>A<sup>+</sup> and <sup>i,j</sup>B<sup>+</sup> ions. This reaction is the most diagnostic fragmentation for flavonoid aglycone identification, because it provides information on the number and type of substituents in the A- and B-rings (Cuyckens & Claeys 2004). The major routes of fragmentation result in cleavage of the C—C bonds at positions (i/j) 1/3, 0/2, 0/3, 0/4 or 2/4 of the C-ring (Figure 2.10). The fragmentation pathways strongly depend on the substitution pattern and the class of flavonoids, <sup>1,3</sup>A<sup>+</sup> (m/z 153), <sup>1,3</sup>B<sup>+</sup> + 2H, <sup>1,4</sup>A<sup>+</sup> + 2H, <sup>0,2</sup>A<sup>+</sup>, and <sup>0,2</sup>A<sup>+</sup> - CO are typically observed for flavonols, while <sup>1,3</sup>B<sup>+</sup>, <sup>0,4</sup>B<sup>+</sup> and <sup>0,4</sup>B<sup>+</sup> -H<sub>2</sub>O are found for flavones (Ma *et al.* 1997, Kyekyeku *et al.* 2017).



Figure 2.10. Schematic Retro Dies-Alder positions of cleavage in flavonols.

Compounds **8**, **9**, **11**, **15**, and **17** showed the loss of 162 amu indicating the presence of a hexose, probably a galactosyl or a glucosyl group. Pentoses, as arabinose, apiose or xylose, were also found as substituents, as in the case of compounds **18** and **20**, which exhibited a mass loss of 132 amu. Compounds **12** and **23** showed the presence of glucuronic acid as substituent, detected by the mass loss of 176 amu in compound **12** and 190 amu in compound **23**. This difference is due to a methyl group in the later compound (**23**) as consequence of an extraction artifact. Glucose is the sugar most commonly identified, while galactose, rhamnose, xylose, and arabinose are less frequent. On the contrary, mannose, fructose, glucuronic and galacturonic acids are rare (Markham 1982, Iwashina 2000).

Mass loss of 314 amu indicates the presence of galloylhexose. This is the case for compound **10**, a quercetin galloylhexoside, as well as, constituents **13**, **16**, and **19**, which are predicted as isomers of kaempferol galloylhexoside. The mass spectrum of compound **19** showed the fragment m/z 449.1071 that indicates a loss of 152 amu in relation to m/z 601.117  $[M+H]^+$ , corroborating the galloyl substitution (Ducrey *et al.* 1995, Mahmoud *et al.* 2001). Moreover, the mass loss of 162 amu [449- 287]<sup>+</sup> confirmed the hexose residue.

In relation to *p*-coumaroyl group, were identified eight compounds with this acylation pattern: 21, 22, 24, 25, 26, 27, 28, and 30. The mass loss of 308 amu is indicative of pcoumaroylhexose substituent, but it can also be a rutinosyl group (rhamnose-glucose). The fragment of m/z 147.0439 found in compounds 25, 27, 28, and 30 ensure the presence of pcoumaroyl or rhamnosyl group. However, acylation with hydroxycynamic acids, as p-coumaric acid, shifts Band I of the flavonol spectrum to lower wavelength, resulting in a peak or shoulder at 305-310 nm. Furhtermore, acylation on the sugar moiety also increases retention time characterizing this type of residue (Tamura et al. 1994, Wang et al. 2003, Atoui et al. 2005). Thus, the absorption in UV (310 nm) and the retention time of constituents 25, 27, 28, and 30 were decisive to propose the acylated identification. Excluding anthocyanins (22 and 24) these compounds were identified as kaempferol p-coumaroylhexoside. Regarding anthocyanins, compound **22** showed a fragment of m/z 799.2077 [M+H]<sup>+</sup>, which fragmentation resulted in *m/z* 625.1552 [M-174]<sup>+</sup>, *m/z* 491.1176 [M-308]<sup>+</sup>, and *m/z* 317.0665 [M-482]<sup>+</sup> corresponding to the neutral loss of acetylpentose and p-coumaroylhexose from a petunidin. The other anthocyanin (24) exhibited a fragment of m/z 813.2243 [M+H]<sup>+</sup>, which fragmentation retrieved *m/z* 639.1716 [M-174]<sup>+</sup>, *m/z* 505.1336 [M-308]<sup>+</sup>, and *m/z* 331.0812 [M-482]<sup>+</sup> consistent with the loss of acetylpentose and p-coumaroylhexose, but in this case, from a malvidin as the anthocyanidin. So, these compounds were identified as Petunidin p-coumaroyl-hexoside acetylpentoside (22) and Malvidin p-coumaroyl-hexoside acetylpentoside (24), respectively. This pattern of substituents of anthocyanins was previously reported in *T. lepdota* and *T* urvilleana (Hendra & Keller 2016, Terahara et al. 1993 - Table 2.1). Acylated anthocyanins are characteristic of Melastomatoideae (APG 2016).

## **3.2.** Isolation by preparative HPLC and identification by Nuclear Magnetic Resonance (NMR)

Compounds were isolated by preparative HPLC, and their purity was analyzed by UPLC-MS, being eight of them analyzed by NMR (Table 2.3 to 2.10). Isolated substances were: kaempferol  $3-O-(6''-O-galloyl)-\beta-D-glucopyranoside (13, 16 or 19), kaempferol <math>3-O-(2''-O-galloyl)-\beta-D-glucopyranoside (13, 16 or 19), kaempferol <math>3-O-glucopyranoside (17),$ kaempferol  $3-O-glucuronide-6''-O-methylester (23), quercetin <math>3-O-(6''-O-p-coumaroyl)-\beta-D-glucopyranoside (25), kaempferol <math>3-O-(6''-O-p-coumaroyl)-\beta-D-glucopyranoside (27),$ kaempferol (29) and quercetin (31). Even though the later was not detected by the UPLC-DAD-ESI-HRMS analysis, it was isolated by preparative HPLC and identified by UPLC-MS and NMR (Table 2.10). All the structures are shown in figure 2.11 and their NMR spectra in Supplemental Figure 2.3 to 2.47.

Kaempferol 3-O-(6"-O-galloyl)-β-D-glucopyranoside (13, 16 or 19) exhibited U.V.  $\lambda_{max}$ =266, 290, 350 and [M+H]<sup>+</sup> m/z 601.1183. <sup>1</sup>H NMR spectrum showed five aromatic hydrogen signals: two doublet signals were observed at  $\delta$  6.21 (1H, d, J= 2.0 Hz) and  $\delta$  6.45 (1H, d, J= 2.0 Hz), corresponding to a *meta*-coupling of these atoms, which were attributed to the H6 and the H8 of the flavonoid A-ring, respectively. Two other doublet signals with orthocoupling constants were found at  $\delta$  6.77 (2H, d, J= 8.8 Hz) and  $\delta$  7.94 (2H, d, J= 8.8 Hz), and were assigned to H3',5' and H2',6', respectively, suggesting that the B-ring of the compound was *para*-substituted. The last aromatic signal was a singlet at  $\delta$  6.92 (2H, s) attributed to H2<sup>'''</sup> and H6<sup>'''</sup> from the galloyl group. Additionally, a doublet at  $\delta$  5.45 (1H, d, J= 7.6 Hz) and two double-doublets at  $\delta$  4.17 (1H, dd, J= 12.0 Hz and J= 3.8 Hz) and  $\delta$  4.26 (1H, dd, J= 12.0 Hz and J= 2.0 Hz) were assigned to H1" and H6" from the sugar moiety. The HMBC spectrum showed two relevant correlations: the anomeric hydrogen (δ 5.45) with C3 (δ 133.52) from flavonoid C-ring and the H6" ( $\delta$  4.26) of the sugar moiety with C7" ( $\delta$  165.36) from the galloyl group. Therefore, the sugar and galloyl group positions were confirmed in the structure. HSQC and HMBC spectra confirmed the assignment of the carbon signals presented in Table 2.3, in agreement with previously reported data (Markham 1982, Braca et al. 2003).

Kaempferol 3-*O*-(2''-*O*-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**) displayed U.V.  $\lambda_{max}$ = 266, 290, 350 and [M+H]<sup>+</sup> *m/z* 601.1183. <sup>1</sup>H NMR spectrum also showed five aromatic hydrogen signals: two from the flavonoid A-ring at δ 6.21 (1H, d, *J*= 2.0 Hz, H6) and δ 6.45 (1H, d, *J*= 2.0 Hz, H8), two from the B-ring, δ 6.83/ 6.93 (2H, d, *J*= 8.8 Hz, H3' and 5') and δ 8.04 (2H, d, *J*= 8.8 Hz, H2' and 6'), and one from the galloyl group at  $\delta$  6.90 (2H, s, H2''' and 6'''). Anomeric carbon appeared as a doublet at  $\delta$  5.47 (1H, d, *J*= 7.6 Hz, H1''). Signals at  $\delta$  3.62 and  $\delta$  3.38 were evidences of 2'' substitution. The HMBC spectrum showed one relevant correlation: the anomeric hydrogen ( $\delta$  5.47) with C3 ( $\delta$  133.22) from the flavonoid C-ring. None hydrogen of the sugar moiety coupled with C7''' to confirm the galloyl group position, but literature data corroborates this substitution in C2''. HSQC and HMBC spectra reinforced the assignment of the carbon signals presented in Table 2.4, which is also in accordance to previously published data (Isobe *et al.* 1980, Markham 1982, Braca *et al.* 2003).

Compound **17** was confirmed by NMR as a mixture of both compounds: kaempferol 3-O- $\beta$ -D-glucopyranoside and kaempferol-(2''-O-methyl)-4'-O- $\alpha$ -D-glucopyranoside (Table 2.5).

Kaempferol 3-*O*-β-D-glucopyranoside (**17**) showed U.V.  $\lambda_{max}$ = 266, 348 nm and [M+H]<sup>+</sup> *m/z* 449.1079. <sup>1</sup>H NMR spectrum showed four aromatic hydrogen signals: two from the flavonoid A-ring at δ 6.22 (1H, d, *J*= 2.0 Hz, H6) and δ 6.46 (1H, s, H8), two from the B-ring, δ 6.89 (2H, d, *J*= 8.4 Hz, H3' and 5') and δ 8.04 (2H, d, *J*= 8.4 Hz, H2' and 6'). Anomeric carbon appeared as a doublet at δ 5.46 (1H, d, *J*= 7.6 Hz, H1''). Signals among δ 3.09 and δ 3.58 were attributed to the sugar moiety. The HMBC spectrum showed one relevant correlation: the anomeric hydrogen (δ 5.46) with C3 (δ 133.63) from the flavonoid C-ring, confirming the position of the sugar moiety. <sup>13</sup>C NMR spectrum showed nineteen carbon signals: C2 (156.61), C3 (133.63), C4 (177.92), C5 (161.65), C6 (99.16), C7 (164.62), C8 (94.12), C9 (156.82), C10 (104.45), C1' (122.33), C2' and 6' (131.33), C3' and 5' (115.58), C4' (160.44), C1'' (101.81), C2'' (74.67), C3'' (76.88), C4'' (70.33), C5'' (77.96), and C6'' (61.28) (Table 2.5). Signal attributions were performed according to similar data from the literature (Markham 1982, Wei *et al.* 2011).

Kaempferol-(2''-*O*-methyl)-4'-*O*-α-D-glucopyranoside (**17**) showed the same U.V ( $\lambda_{max}$ = 266, 348) of compound described above, but the [M+H]<sup>+</sup> was *m/z* 463.0865, suggesting the presence of an additional methyl group in the structure, which was confirmed by the signals at δ 3.26 (s, 3H) and δ 54.74. <sup>1</sup>H NMR spectrum also showed four aromatic hydrogen signals: two from the flavonoid A-ring at δ 6.21 (1H, d, *J*= 2.0 Hz, H6) and δ 6.46 (1H, s, H8), two from the flavonoid B-ring, δ 6.93 (2H, d, *J*= 8.5 Hz, H3' and 5') and δ 8.04 (2H, d, *J*= 8.5 Hz, H2' and 6'). Anomeric carbon appeared as a doublet at δ 4.51 (1H, d, *J*= 3.6 Hz, H1'') and this small coupling constant suggests an α-sugar series. The HMBC spectrum showed one relevant correlation: the anomeric hydrogen (δ 4.51) with 2''*O*Me (δ 54.74) from the methyl group, confirming this group at C2'' position on the sugar moiety. Although the correlation among

the anomeric hydrogen with carbon C4' of the flavonoid has not been observed, the sugar unity was bonded in that position because the <sup>13</sup>C NMR chemical shifts at C2 ( $\delta$  147.27) and C3 ( $\delta$  136.10) positions of the flavonoid ensure the presence of a free hydroxyl group at C3. <sup>13</sup>C NMR spectrum showed twenty carbon signals: C2 (147.27), C3 (136.10), C4 (176.35), C5 (161.13), C6 (98.68), C7 (164.40), C8 (93.95), C9 (156.71), C10 (103.48), C1' (122.10), C2' and 6' (129.95), C3' and 5' (115.91), C4' (159.68), C1'' (100.12), C2'' (73.84), C3'' (72.44), C4'' (73.04), C5'' (70.79), C6'' (61.42), and OMe-C2'' (54.74) (Table 2.5). Signal attributions of flavonoid were performed through HSQC and HMBC correlations. Indeed, the methyl group could be an artefact due to the use of acidified methanol in extraction procedure, further analyses are necessary to confirm this substitution. Moreover, only kaempferol 4'-*O*- $\beta$ -D-glucopyranoside was described (Scheer & Wichtl 1987), being the kaempferol-(2''-*O*-methyl)-4'-*O*- $\alpha$ -D-glucopyranoside or kaempferol 4'-*O*- $\alpha$ -D-glucopyranoside an inedited substance.

Carbon number	<sup>1</sup> Η	<sup>13</sup> C*	НМВС	<sup>1</sup> H (CD <sub>3</sub> OD) (Braca <i>et al.</i> 2003)	<sup>13</sup> C (CD <sub>3</sub> OD) (Braca <i>et al.</i> 2003)
2	-	157.15	-	-	158.5
3	-	133.52	-	-	136.0
4	-	-	-	-	179.4
5	-	161.9	-	-	163.0
6	6.21 d ( <i>J</i> = 2.0 Hz)	99.58	C8, C10, C7, C5	6.19 d ( <i>J</i> = 2.0 Hz)	100.0
7	-	164.5	-	-	166.0
8	6.45 d ( <i>J</i> = 2.0 Hz)	94.91	C10, C9, C7, C6	6.35 d ( <i>J</i> = 2.0 Hz)	95.0
9	-	157.26	-	-	159.0
10	-	104.66	-	-	105.7
1'	-	121.04	-	-	122.0
2',6'	7.94 d ( <i>J</i> = 8.8 Hz)	131.19	C2, C4', C6'or 2'	7.94 d ( <i>J</i> = 8.0 Hz)	132.2
3',5'	6.77 d ( <i>J</i> = 8.8 Hz)	116.46	C4', C1', C5' or 3'	6.73 d ( <i>J</i> = 8.0 Hz)	116.1
4'	-	150.58	-	-	161.5
1''	5.45 d ( <i>J</i> = 7.6 Hz)	102.17	C3	5.21 d ( <i>J</i> = 7.7 Hz)	104.4
ר ''2		74.52	C1'', C3''	-	75.9
3''		76.51	C2'', C4''	-	78.1
4''	≻ 3.24 - 3.50*	69.89	C2'', C6''	-	71.5
5'' J		74.60	-	-	75.8
6''	4.17 dd ( <i>J</i> = 12 and 3.8 Hz), 4.26 dd ( <i>J</i> = 12 and 2.0 Hz)	63.15	C5'', C4'', C7'''	-	64.3
1'''	-	120	-	-	121.3
2''',6'''	6.92 s	109.70	C7''', C2'''or 6''', C3'''or 5''', C4''', C1'''	6.94 s	110.3
3''',5'''	-	146.11	-	-	146.4
4'''	-	138.94	-	-	139.3
7'''	-	165.36	-	-	168.2
OH-C5	12.52 s	-	-	-	-
OH-C7	10.87 s	-	-	-	-
OH-C4'	10.06 s	-	-	-	-

**Table 2.3**. NMR data of <sup>1</sup>H, <sup>13</sup>C and HMBC for Kaempferol 3-O-(6"-O-galloyl)- $\beta$ -D-glucopyranoside (**13**, **16** or **19**) comparing to the literature.

\*Chemical shifts of <sup>13</sup>C NMR were obtained by correlations in HSQC and HMBC.

Carbon number	<sup>1</sup> Η	<sup>13</sup> C*	НМВС	<sup>1</sup> H (CD <sub>3</sub> OD) (Braca <i>et al.</i> 2003)	<sup>13</sup> C (DMSO-d6) (Isobe <i>et al.</i> 1980)
2	-	155.46	-	-	156.6
3	-	133.22	-	-	132.8
4	-	-	-	-	177.4
5	-	160.00	-	-	161.4
6	6.21 d ( <i>J</i> = 2.0 Hz)	98.02	C8, C10, C7, C5	6.19 d ( <i>J</i> = 2.0 Hz)	98.6
7	-	163.30	-	-	165.6
8	6.45 d ( <i>J</i> = 2.0 Hz)	93.05	C10, C6, C7, C9	6.35 d ( <i>J</i> = 2.0 Hz)	94.1
9	-	155.50	-	-	156.6
10	-	102.70	-	-	104.3
1'	-	120.90	-	-	121.1
2',6'	8.04 d ( <i>J</i> = 8.8 Hz)	129.54	C2'or 6', C4', C2, C3'''or 5'''	7.94 d (J= 8.0 Hz)	131.2
3',5'	6.89/6.93 d (J= 8.8 Hz)	114.66	C3'''or 5''', C4', C1'	6.73 d ( <i>J</i> = 8.0 Hz)	115.6
4'	-	158.40	-	-	160.3
1''	5.47 d ( <i>J</i> = 7.6 Hz)	100.38	C3	5.21 d ( <i>J</i> = 7.7 Hz)	99.1
ן "2		73.68	C1, C3	-	74.6
3''	200 222*	75.87	C2, C4	-	78.0
4''	3.09 - 3.22*	69.48	C5	-	70.5
ر <sub>"5</sub>		76.74	C3, C4	-	74.6
6''	3.62 and 3.38*	60.68	C5, C4	-	61.1
1'''	-	124.00	-	-	119.9
2''',6'''	6.90 s	108.45	C7''', C4''', C3'''ou 5''', C1''', C2''' or 6'''	6.94 s	109.4
3''',5'''	-	145.96	-	-	145.7
4'''	-	138.86	-	-	138.7
7'''	-	162.11	-	-	164.4

**Table 2.4.** NMR data of <sup>1</sup>H, <sup>13</sup>C and HMBC for Kaempferol 3-*O*-(2"-*O*-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**) comparing to the literature.

\*Chemical shifts of <sup>13</sup>C NMR were obtained by correlations in HSQC and HMBC.

	Kaempferol 3-O-β-D-glucopyranoside (Astragalin)					Kaem	pferol-(2	''-O-methyl)-4'-O	-α-D-glucopyranosid	е
Carbon number	<sup>1</sup> Η	<sup>13</sup> C	НМВС	<sup>1</sup> H (DMSO-d6) (Wei <i>et al</i> 2011)	<sup>13</sup> C (DMSO- d6) (Wei <i>et al</i> 2011)	<sup>1</sup> Η	<sup>13</sup> C	НМВС	<sup>1</sup> H (DMSO-d6) (Scheer & Wichtl, 1987)	<sup>13</sup> C (DMSO- d6) (Scheer & Wichtl, 1987)
2	-	156.62	-	_	156.72	-	147.27	_	_	145.91
3	-	133.63	-	-	133.66	-	136.10	-	-	136.30
4	-	177.92	-	-	177.94	-	176.35	-	-	176.05
5	-	161.65	-	-	161.69	-	161.13	-	-	160.69
6	6.22 d ( <i>J</i> = 2.0 Hz)	99.16	C5, C7, C8, C10	6.31 d (J= 2.4 Hz)	99.15	6.21 d ( <i>J</i> = 2.0 Hz)	98.68	C5, C7, C8, C10	6.19 d ( <i>J</i> = 2.0 Hz)	98.33
7	-	164.62	-	-	164.59	-	164.40	-	-	164.32
8	6.46 sl	94.12	C4, C6, C7, C9, C10	6.55 d ( <i>J</i> =2.4 Hz)	94.11	6.46 sl	93.95	C6, C7, C9	6.45 d ( <i>J</i> = 2.0 Hz)	93.60
9	-	156.82	-	-	156.84	-	156.71	-	-	156.27
10	-	104.45	-	-	104.48	-	103.48	-	-	103.02
1'	-	122.33	-	-	121.36	-	122.10	-	-	124.41
2',6'	8.04 d ( <i>J</i> = 8.4 Hz)	131.33	C2, C4', C3' or 5', C2' or 6'	8.16 d ( <i>J</i> = 9.0 Hz)	131.34	8.04 d ( <i>J</i> = 8.5 Hz)	129.95	C2, C3' or 5', C4'	8.13 d (J= 9.0 Hz)	129.13
3',5'	6.89 d ( <i>J</i> = 8.4 Hz)	115.58	C1, C3' or 5', C4'	6.84 d ( <i>J</i> = 9.0 Hz)	115.56	6.93 d ( <i>J</i> = 8.5 Hz)	115.91	C1', C3' or 5', C4'	7.18 d ( <i>J</i> = 9.0 Hz)	116.11
4'	-	160.44	-	-	160.41	-	159.68	-	-	158.44
1''	5.46 d ( <i>J</i> =7.6 Hz)	101.81	C3, C5''	5.29 d (J=7.8 Hz)	101.35	4.51 d (J=3.6 Hz)	100.12	C2'', OMe	4.98 d (J=7.2 Hz)	99.93
2''	3.18	74.67	C1", C3", C4"	)	74.68	3.37	73.84	C3''	-	73.17
3''	3.22	76.88	C2'', C4''		76.89	3.18	72.44	C2''	-	76.6
4''	3.09	70.33	C6'', C5''	> 3.27 - 3.67	70.37	3.29	73.04	C1''	-	69.61
5''	3.09	77.96	C6'', C4''		77.95	3.04	70.79	C6'', C4''	-	77.12
6''	3.58 and 3.33	61.28	C5'', C4''	J	61.31	3.62 and 3.44	61.42	C5'', C4''	-	60.60
2''OMe	-	_	-	-	-	3.26 s	54.74	C1''	-	-

**Table 2.5.** NMR data of <sup>1</sup>H, <sup>13</sup>C and HMBC for a mixture of both compounds: kaempferol  $3-O-\beta-D$ -glucopyranoside and kaempferol-(2''-O-methyl)-4'-O- $\alpha$ -D-glucopyranoside (**17**) comparing to the literature.

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Kaempferol 3-*O*-glucuronide-6''-*O*-methylester (**23**) exhibited U.V.  $\lambda_{max}$ = 268, 320 and [M+H]<sup>+</sup> *m/z* 477.1031. <sup>1</sup>H NMR spectrum also showed four aromatic hydrogen signals: two from the flavonoid A-ring at  $\delta$  6.23 (1H, d, *J*= 2 Hz, H6) and  $\delta$  6.45 (1H, d, *J*= 2 Hz, H8), and two from the B-ring,  $\delta$  6.89 (2H, d, *J*= 8.9 Hz, H3' and 5') and  $\delta$  8.02 (2H, d, *J*= 8.9 Hz, H2' and 6'). Anomeric carbon appears as a doublet at  $\delta$  5.47 (1H, d, *J*= 7.7 Hz, H1''). A singlet signal at  $\delta$  3.57 reveals the presence of a methyl group. The HMBC spectrum showed three relevant correlations: the anomeric hydrogen ( $\delta$  5.47) with C3 ( $\delta$  133.58) from the flavonoid C-ring, the H5'' ( $\delta$  3.73) from the sugar moiety with C6'' ( $\delta$  169.52) and the methoxyl group ( $\delta$  3.57) with C6'' from the carboxyl group in the glucuronic acid. Therefore, the sugar and the methyl group positions were confirmed in the structure. HSQC and HMBC spectra confirmed the assignment of the carbon signals presented in Table 2.6 in accordance with previously reported data (Markham 1982, Jung *et al.* 2003).

Quercetin 3-O-(6"-O-p-coumaroyl)- $\beta$ -D-glucopyranoside (**25**) exhibited U.V.  $\lambda_{max}$ =271, 312, and  $[M+H]^+$  m/z 611.1393. <sup>1</sup>H NMR spectrum showed seven aromatic hydrogen signals: two doublet signals were observed at  $\delta$  6.14 (1H, d, J= 2.1 Hz) and  $\delta$  6.38 (1H, d, J= 2.1 Hz), corresponding to the meta-coupling of H6 and H8 from the flavonoid A-ring, respectively, two other doublet signals were found at  $\delta$  6.78 (1H, d, J= 8.4 Hz) and  $\delta$  7.52 (1H, d, J= 2.3 Hz) assigned to H5' and H2', respectively, a double doublet was detected at  $\delta$  7.66 (1H, dd, J= 8.5 Hz and J= 2.3 Hz) corresponding to H6', suggesting that the B-ring of the compound was substituted with two hydroxyl groups at C3' and C4'. The last two aromatic signals were doublets at δ 6.83 (2H, d, J= 8.4 Hz, H3<sup>'''</sup> and 5<sup>'''</sup>) and δ 7.37 (2H, d, J= 8.4 Hz, H2<sup>'''</sup> and 6<sup>'''</sup>) attributed to a *trans*-double bond, were assigned to a *p*-coumaroyl group. Two signals from the double bond were also detected at  $\delta$  6.13 (2H, d, J= 15.7 Hz, H8''') and  $\delta$  7.36 (2H, d, J= 15.7 Hz, H7"). Additionally, a doublet at  $\delta$  5.42 (1H, d, J= 7.8 Hz) and a double-doublet at  $\delta$ 4.12 (1H, dd, J = 11.5 Hz and J = 4.6 Hz) were assigned to H1" and H6" from the sugar moiety. The HMBC spectrum was not done and the group positions were suggested by comparing signals to literature. HSQC spectra allowed assignment of carbon signals presented in Table 2.7 in agreement with data from the literature (Markham 1982, Lavault & Richomme et al. 2004).

Carbon number	<sup>1</sup> Η	<sup>13</sup> C*	НМВС	<sup>1</sup> H (DMSO- d6) (Jung <i>et al.</i> 2003)	<sup>13</sup> C (DMSO- d6) (Jung <i>et al.</i> 2003)
2	-	157.17	-	-	156.4
3	-	133.58	-	-	133.0
4	-	-	-	-	177.2
5	-	161.58	-	-	161.2
6	6.23 d (J= 2.0 Hz)	99.53	C8, C10, C7	6.21 d (J= 2.0 Hz)	98.9
7	-	164.78	-	-	164.3
8	6.45 d (J= 2.0 Hz)	94.1	C6, C10, C7, C9	6.43 d (J= 2.0 Hz)	93.8
9	-	156.94	-	-	156.6
10	-	104.46	-	-	104.0
1'	-	121.05	-	-	120.6
2',6'	8.02 d ( <i>J</i> = 8.9 Hz)	131.73	C3, C2' or 6', C4'	8.02 d (J= 8.0 Hz)	130.9
3',5'	6.89 d ( <i>J</i> = 8.9 Hz)	115.5	C1', C3' or 5', C4'	6.89 d (J= 8.0 Hz)	115.1
4'	-	160.62	-	-	160.2
1''	5.47 d ( <i>J</i> = 7.7 Hz)	101.74	C3	5.47 d (J= 7.4 Hz)	101.4
ן "2		74.6	C1'', C3'',	-	73.9
3'' }	3.24-3.36*	75.92	C2'', C4''	-	75.5
4'' J		71.95	C5'', C6''	-	71.5
5''	3.73 d ( <i>J</i> = 9.7 Hz)	76.13	C1'', C6'', C4''	-	75.6
6''	-	169.52	-	-	169.0
MeO-C6"	3.57 s	52.32	C6''	3.57 s	-
OH-C5	12.51 s	-	-	12.51 s	-
OH-C7	10.97 s	-	-	-	-
OH-C4'	10.22 s	-	-	-	-

**Table 2.6.** NMR data of <sup>1</sup>H, <sup>13</sup>C and HMBC for Kaempferol 3-O-glucuronide-6"-O-methylester (**23**) comparing to the literature.

\*Chemical shifts of <sup>13</sup>C NMR were obtained by correlations in HSQC and HMBC.

Carbon number	<sup>1</sup> Η	<sup>13</sup> C*	<sup>1</sup> H (CD <sub>3</sub> OD) (Lavault & Richomme <i>et al.</i> 2004)	<sup>13</sup> C (CD <sub>3</sub> OD) (Lavault & Richomme <i>et al.</i> 2004)
2	-	-	-	158.8
3	-	-	-	135.5
4	-	-	-	179.6
5	-	-	-	163.2
6	6.14 d ( <i>J</i> = 2.1 Hz)	99.19	6.21 d ( <i>J</i> = 1.9 Hz)	100.6
7	-	-	-	167.3
8	6.38 d ( <i>J</i> = 2.1 Hz)	93.90	6.38 d ( <i>J</i> = 1.9 Hz)	95.3
9	-	-	-	158.8
10	-	-	-	105.6
1'	-	-	-	123.4
2'	7.52 d ( <i>J</i> = 2.3 Hz)	116.42	7.68 d ( <i>J</i> = 1.7 Hz)	115.0
3'	-	-	-	146.2
4'	-	-	-	146.9
5'	6.78 d ( <i>J</i> = 8.4 Hz)	116.01	6.90 d ( <i>J</i> = 8.4 Hz)	116.2
6'	7.66 dd (J= 8.4 Hz and 2.3 Hz)	122.38	7.66 dd (J= 8.3 Hz and 2.0 Hz)	123.4
1''	5.42 d ( <i>J</i> = 7.8 Hz)	102.03	5.32 d ( <i>J</i> = 7.0 Hz)	104.2
2''	-	-	-	76.1
3''	-	-	-	78.3
4''	-	-	-	72.0
5''	-	-	-	75.9
6''	4.12 dd ( <i>J</i> = 11.5 Hz and 4.6Hz)	63.62	4.29 m, 4.37 m	64.6
1'''	-	-	-	131.5
2''',6'''	7.37 d ( <i>J</i> = 8.4 Hz)	130.62	7.40 d ( <i>J</i> = 8.5 Hz)	117.6
3''',5'''	6.83 d ( <i>J</i> = 8.4 Hz)	115.61	6.87 d ( <i>J</i> = 8.8 Hz)	131.5
4'''	-	-	-	161.5
7'''	7.36 d ( <i>J</i> = 15.7 Hz)	145.31	7.48 d ( <i>J</i> = 15.7 Hz)	140.2
8'''	6.13 d ( <i>J</i> = 15.7 Hz)	114.10	6.16 d ( <i>J</i> = 15.7 Hz)	115.0
9'''	-	-	-	169.3
OH-C5	12.64 s	-	-	-
OH-C7	10.84 s	-	-	-
OH-C3'	9.16 s	-	-	-
OH-C4'	9.73 s	-	-	-
OH-C4'''	10.01 s	-	-	-

**Table 2.7.** NMR data of <sup>1</sup>H, <sup>13</sup>C and HMBC for Quercetin 3-*O*-(6"-*O*-*p*-coumaroyl)-β-D-glucopyranoside (**25**) comparing to literature.

\*Chemical shifts of <sup>13</sup>C NMR were obtained by correlations in HSQC.

Kaempferol 3-O-(6"-O-*p*-coumaroyl)-β-D-glucopyranoside (**27**) displayed U.V.  $\lambda_{max}$ =268, 314, and [M+H]<sup>+</sup> *m/z* 595.1418. <sup>1</sup>H NMR spectrum showed six aromatic hydrogen signals: two from the flavonoid A-ring at δ 6.14 (1H, d, *J* = 2.1 Hz, H6) and δ 6.40 (1H, d, *J* = 2.1 Hz, H8), two from *para*-substituted B-ring, δ 6.78 (2H, d, *J* = 8.4 Hz, H3' and 5') and δ 8.05 (2H, d, *J* = 8.4 Hz, H2' and 6'), and two from *p*-coumaroyl group at δ 6.86 (2H, d, *J* = 8.7 Hz, H2''' and 6'''). Two signals from the double bond were also detected at δ 6.11 (2H, d, *J* = 15.9 Hz, H8''') and δ 7.34 (2H, d, *J* = 15.9 Hz, H7'''). Anomeric carbon appears as a doublet at δ 5.41 (1H, d, *J* = 7.7 Hz, H1''). The signal at δ 4.10 (1H, d, *J*=6.2 Hz) is an evidence of 6'' substitution. The HMBC spectrum showed one relevant correlation confirming the *p*-coumaroyl position: the H6'' (δ 4.10) from the sugar moiety with the C9''' from the *p*-coumaroyl group. None of the hydrogens of the flavonoid moiety showed correlation with carbons from the sugar moiety, however, literature data corroborate the substitution in C3 (Nikaido *et al.* 1986). HSQC and HMBC spectra reinforced the assignment of the carbon signals presented in Table 2.8 in accordance to the literature (Markham 1982, Nikaido *et al.* 1986).

Kaempferol (**29**) showed U.V.  $\lambda_{max}$ = 270, 368, and [M+H]<sup>+</sup> *m/z* 287.0545. <sup>1</sup>H NMR spectrum displayed four aromatic hydrogen signals. Two doublet signals were observed at  $\delta$  6.21 (1H, d, *J*= 2.0 Hz) and  $\delta$  6.46 (1H, d, *J*= 2.0 Hz), attributed to a *meta*-coupling of H6 and H8 from the flavonoid A-ring, respectively. Two other doublet signals, with *ortho*-coupling constants were found at  $\delta$  6.94 (2H, d, *J*= 8.9 Hz) and  $\delta$  8.05 (2H, d, *J*= 8.9 Hz), and were assigned to H3',5' and H2',6', respectively, suggesting that the B-ring of the compound was *para*-substituted. <sup>13</sup>C NMR spectrum showed thirteen carbon signals: C2 (147.27), C3 (136.11), C4 (176.36), C5 (161.15), C6 (96.68), C7 (64.38), C8 (93.95), C9 (156.63), C10(103.49), C1' (122.12), C2' and 6' (129.95), C3' and 5' (115.91), C4' (159.67) (Table 2.9). Attribution was performed according to similar data from the literature (Markham 1982, Wahab *et al.* 2014, Guo *et al.* 2016).

Quercetin (**31**) showed  $[M+H]^+ m/z$  302.94. HSQC spectrum allowed the assignment of the hydrogen and carbon signals presented in Table 2.10, which were in accordance to previously published data (Markham 1982, Joshi *et al.* 2009). The signals at  $\delta$  6.18 and 6.42, corresponding to the flavonoid A-ring C6 and C8, respectively, and at  $\delta$  6.88, 7.52 and 7.65 from the B-ring were enough to confirm the presence of quercetin.

Carbon number	<sup>1</sup> H	<sup>13</sup> C*	НМВС	<sup>1</sup> H (DMSO- d6) (Nikaido <i>et al.</i> 1986)	<sup>13</sup> C (DMSO-d6) (Nikaido <i>et a.l</i> 1986)
2	-	156.96	-	-	155.7
3	-	-	-	-	132.5
4	-	-	-	-	176.5
5	-	161.69	-	-	155.7
6	6.14 d ( <i>J</i> = 2.1 Hz)	99.30	C8, C10, C5	6.41 d ( <i>J</i> = 1.8 Hz)	93.2
7	-	164.56	-	-	163.6
8	6.40 d ( <i>J</i> = 2.1 Hz)	94.19	C10, C9, C7, C6	6.18 d (J= 1.8 Hz)	98.3
9	-	156.55	-	-	160.4
10	-	104.20	-	-	103.3
1'	-	125.58	-	-	120.2
2',6'	8.05 d (J= 8.4 Hz)	131.47	C2, C4', C6'or 2'	6.90 d (J= 8.8 Hz)	130.1
3',5'	6.78 d (J= 8.4 Hz)	116.35	C1', C5' or 3'	8.02 d (J= 8.8 Hz)	114.5
4'	-	160.38	-	-	159.2
1''	5.41 d ( <i>J</i> = 7.7 Hz)	102.04	-	5.48 d ( <i>J</i> = 8 Hz)	100.6
ר "2		71.47	C1	-	73.9
3''		70.32	-	-	76.0
4''	≻ 3.51 - 3.7	73.40	C5"ou C3", C1", C6"or C2"	-	69.7
ل <sub>"5</sub>		69.05	-	-	73.9
6''	4.10 d ( <i>J</i> = 6.2 Hz)	63.71	C9''', C4'', C5''or C3''	4.08 dd ( <i>J</i> = 2 e 12 Hz)	62.7
1'''	-	121.33 or 125.05	-	-	124.3
2''',6'''	7.35 d (J= 8.7 Hz)	130.66	C7''', C4'''	6.83 d (J= 8.8 Hz)	139.4
3''',5'''	6.86 d ( <i>J</i> = 8.7 Hz)	115.52	C1''', C5''or C3'', C4'''	7.39 d ( <i>J</i> = 8.8 Hz)	115.2
4'''	-	160.53	-	-	158.8
7'''	7.34 d ( <i>J</i> = 15.9 Hz)	145.24	C9''', C2'''or C6'''	7.38 d ( <i>J</i> = 16.1 Hz)	143.4
8'''	6.11 d ( <i>J</i> = 15.9 Hz)	114.29	-	6.14 d ( <i>J</i> = 16.1 Hz)	114.8
9'''	-	166.85	-	-	167.1
OH-C5	12.59 s	-	-	-	-
OH-C7	10.87 s	-	C7, C6, C8	-	-
OH-C4'	10.03 s	-	-	-	-
OH-C4'''	10.20 s	-	-	-	-

**Table 2.8.** NMR data of <sup>1</sup>H, <sup>13</sup>C and HMBC for Kaempferol  $3-O-(6''-O-p-coumaroyl)-\beta-D-glucopyranoside ($ **27**) comparing to literature.

\*Chemical shifts of <sup>13</sup>C NMR were obtained by correlations in HSQC and HMBC.

Carbon number	<sup>1</sup> Η	<sup>13</sup> C	<sup>1</sup> H (DMSO-d6) (Guo <i>et al.</i> 2016)	<sup>13</sup> C (DMSO-d6) (Wahab <i>et al.</i> 2014)
2	-	147.27	-	146.3
3	-	136.11	-	135.2
4	-	176.36	-	175.2
5	-	161.15	-	160.4
6	6.21 d (J= 2.0 Hz)	96.68	6.19 d ( <i>J</i> = 1.7 Hz)	98.4
7	-	164.38	-	163.7
8	6.46 d (J= 2.0 Hz)	93.95	6.44 d ( <i>J</i> = 1.7 Hz)	93.8
9	-	156.63	-	156.7
10	-	103.49	-	103.1
1'	-	122.12	-	122.1
2',6'	8.05 d (J= 8.9 Hz)	129.95	8.04 d (J= 8.8 Hz)	129.4
4'		159.67		158.7
3',5'	6.94 d (J= 8.9 Hz)	115.91	6.93 d (J= 8.8 Hz)	115.3
OH-C3	9.38 s		9.37 s	
OH-C5	12.48 s	-	12.48 s	-
OH-C7	10.83 s	-	10.78 s	-
OH-C4'	10.13 s	-	10.10 s	-

 Table 2.9. NMR data of <sup>1</sup>H, <sup>13</sup>C e HMBC for Kaempferol (29) comparing to literature.

Carbon number	<sup>1</sup> Η	<sup>13</sup> C*	<sup>1</sup> H (DMSO-d6) (Joshi <i>et al.</i> , 2009)	<sup>13</sup> C (DMSO-d6) (Joshi <i>et al.</i> , 2009)
2	-	-	-	145.05
3	-	-	-	135.72
4	-	-	-	175.83
5	-	-	-	160.70
6	6.18*	98.75	6.16 d (J= 2.0 Hz)	98.34
7	-	-	-	163.80
8	6.42*	93.93	6.40 d (J= 2.0 Hz)	93.34
9	-	-	-	156.13
10	-	-	-	103.01
1'	-	-	-	121.96
2'	7.65*	115.47	7.47 d (J= 2.0 Hz)	115.00
3'	-	-	-	145.05
4'	-	-	-	147.69
5'	6.88*	116.13	6.85 d (J= 9.0 Hz)	115.60
6'	7.52*	120.52	7.62 dd (J= 2.0 e 9.0 Hz)	119.90
OH-C3	9.60 s	-	9.60 s	-
OH-C5	12.46 s	-	12.48 s	-
OH-C7	10.86 s	-	-	-
OH-C3'	9.36 s	-	-	-
OH-C4'	9.31 s	-	-	-

Table 2.10. NMR data of <sup>1</sup> H, <sup>13</sup> C e HMBC for Quercetin comparing to literatur	e.
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\*Chemical shifts of <sup>13</sup>C NMR were obtained by correlations in HSQC.



Figure 2.11. Compounds of *T. pulchra* flowers identified by UV-visible absorption, mass and NMR spectra.

Considering aerial parts of *T. pulchra*, sixteen compounds are described for the first time: quercetin galloylhexoside (**10**), quercetin hexoside (**11**), quercetin glucuronide (**12**), kaempferol  $3-O-(2''-O-galloyl)-\beta-D-glucopyranoside ($ **13**,**16**or**19** $), kaempferol <math>3-O-(6''-O-galloyl)-\beta-D-glucopyranoside ($ **13**,**16**or**19** $), kaempferol-(2''-O-methyl)-4'-O-<math>\alpha$ -D-glucopyranoside (**17**), kaempferol pentoside (**18** and **20**), kaempferol *p*-coumaroylglucoside (**21**, **28**, and **30**), petunidin *p*-coumaroylhexoside acetylpentoside (**22**), kaempferol 3-O-glucopyranoside (**23**), malvidin*p*-coumaroylhexoside acetylpentoside (**24** $), quercetin <math>3-O-(6''-O-methyl)-\beta-D-glucopyranoside ($ **25** $), and kaempferol <math>3-O-(6''-O-p-coumaroyl)-\beta-D-glucopyranoside ($ **27**).

Fourteen compounds are described for the first time for the genus: myricetin hexoside (8 and 9), kaempferol 3-O-(6"-O-galloyl)-β-D-glucopyranoside (13, 16 or 19), kaempferol 3- $O-(2''-O-galloyl)-\beta-D-glucopyranoside$  (13, 16 or 19), kaempferol-(2''-O-methyl)-4'-O- $\alpha$ -Dglucopyranoside (17), kaempferol pentoside (18 and 20), kaempferol p-coumaroyl-glucoside (21, 28, and 30), petunidin p-coumaroyl-hexoside acetylpentoside (22), kaempferol 3-Oglucuronide-6"-O-methylester (23), and kaempferol 3-O-(6"-O-p-coumaroyl)-β-Dglucopyranoside (27). Flavonols, mainly myricetins are characteristic of Mytales (APG, 2016). Although the chemical characterization of *Tibouchina* is extremely scarse, quercetin and isorhamnetin have been the most commonly found flavonols in aerial parts and already identified in T. ciliaris, T. grandiflora, T. granulosa, T. lepidota, T. paratropica, T. pereirae, T. pulchra T. semidecantra, and T. urvelleana. On the other hand, kaempferol has only been found in T. ciliaris, T. pereirae, and T. pulchra. Regarding anthocyanins, malvidin has already been identified in T. lepidota, T. grandiflora, T. semidecantra, and T. urvelleana, while petunidin has exclusively been described in T. granulosa (Table 2.2). Here, we report for the first time anthocyanins (i.e. malvidin and petunidin) in T. pulchra.

The most common acyl groups generally found as flavonoid substituents are hydroxycinnamic acids (*e.g.* caffeic, ferulic and *p*-coumaric acids) (Willians 2004). Flavonoids with *p*-coumaroyl group have already been found in *Tibouchina* species, such as *T. ciliaris* (kaempferol 7-*O*-*p*-coumaroyl), *T. grandiflora* (malvidin 3-(*p*-coumaroyl)-sambubioside-5-glucoside), malvidin 3-(*p*-coumaroyl-glucoside)-5-glucoside), and *T. urvilleana* (malvidin 3-*O*-(6-*O*-*p*-coumaryl-β-D-glucopyranoside)-5-*O*-(2-*O*-acetyl-β-D-xylopyranosyl)) (Colorado *et al.* 2007, Lowry *et al.* 1995, Bobbio *et al.* 1985, Terahara *et al.* 1993), this is in agreement with results found in *T. pulchra*.

Acylation with hydroxybenzoic acids, such as gallic acid, is rare in angiosperms because the active production of hydrolysable tannins is restricted to certain orders: Hammamelidales, Fagales, Dilleniales, Theales, Ericales, Rosales, Myrtales, Cornales, Proteales, Sapindales, Geraniales, and Juglandales (Haslam 2007). Tannin occurrence has been described in *Tibouchina* genus (Table 2.1) in at least three species: *T. semidecantra*, *T. pulchra* and *T. multiflora* (Yoshida *et al.* 1991, Furlan 2004, Motta *et al.* 2005, Santos & Furlan 2013, Yoshida *et al.* 1999). Although hydrolysable tannins were not identified neither in *T. ciliaris* nor in *T. granulosa*, the presence of quercetin 6"-*O*-gallate and quercetin 3-(*O*-galloyl)-hexoside (Colorado *et al.* 2007, Sobrinho *et al.* 2017), respectively, is indicative of the existence of this class of metabolites. Here, we found quercetin galloylhexoside and three isomers of kaempferol galloylhexoside (kaempferol-3-*O*-(6"-*O*-galloyl)- $\beta$ -D-glucopyranoside, kaempferol-3-*O*-(2"-*O*-galloyl)- $\beta$ -D-glucopyranoside, and one not determined the galloyl position), in agreement with the finding of tannins in *T. pulchra* (Furlan 2004, Motta *et al.* 2005, Santos & Furlan 2013).

Although in Melastomataceae many compounds have been isolated and identified by extensive spectral analyses, considering the size of the family, the number of studied species is still small. Only three of the nine tribes have been chemical investigated (Melastomeae, Miconieae and Sonerileae). The most commonly found natural products in this family belong to terpenoids, simple phenolics, quinones, lignans and their glycosides, flavonoids, as well as a vast range of tannins, mainly hydrolyzable tannins (Serna & Martínez 2015).

Below, according to Serna & Martinéz (2015), we discuss the presence of *T. pulchra* constituents identified by NMR in the context of the natural products fully structural elucidated from Melastomataceae.

The results obtained describe, for the first time, the presence of kaempferol 3-O-(6"-O-galloyl)- $\beta$ -D-glucopyranoside (**13**, **16** or **19**), kaempferol 3-O-(2"-O-galloyl)- $\beta$ -Dglucopyranoside (**13**, **16** or **19**), kaempferol-(2"-O-methyl)-4'-O- $\alpha$ -D-glucopyranoside (**17**), kaempferol 3-O-glucuronide-6"-O-methylester (**23**), and kaempferol 3-O-(6"-O-*p*coumaroyl)- $\beta$ -D-glucopyranoside (**27**) in Melastomataceae. Kaempferol 3-O- $\beta$ -D-(2",6"-di-O*trans*-coumaroyl)-glucopyranoside was found in *Miconia cabucu* Hoehne and *M. rubiginosa* (Bonpl.) DC. (Rodrigues *et al.* 2007), but usually kaempferol is 7-O substituted in this family (Serna & Martinéz 2015). Kaempferol aglycone (**29**) was also found in *Medinilla magnifica* Lindley and *Centradenia floribunda* Planch (Serna & Martinéz 2015). Regarding anthocyanins, this was the first description of petunidin *p*coumaroylhexoside acetylpentoside (**22**) in *T. pulchra*, but further studies on this anthocyanin are necessary for its precisely identification, since it was identified by MS and UV analysis. The anthocyanins previously described in Melastomataceae include pelargonidin, cyanidin, peonidin, delphinidin, and malvidin glycosides or acylglycosides. In this work malvidin *p*coumaroylhexoside acetylpentoside (**24**) was the major anthocyanin in *T. pulchra*, which is in agreement with the proposition of malvidin as the most common anthocyanidin moiety in Melastomataceae (Serna & Martinéz 2015).

## 4. CONCLUSION

Melastomataceae and, in particular, *Tibouchina* are chemically poorly characterized taxons. Here, out of the sixteen compounds described for the first time in *T. pulchra*, five of them described for the first time in Melastomataceae, moreover an unpublished compound was identified as kaempferol 4'-O-(2''-methyl)- $\alpha$ -D-glucopyranoside. The advances in spectrometric techniques offer a unique opportunity to improve our knowledge about the chemical structure of natural products. Studies about flower anthocyanins are scarce and the understanding of their structure, biosynthesis and the regulatory mechanisms involved in their accumulation pattern can improve our knowledge about plant secondary metabolism, the relationship between flower and pollinators, and bring new insights for future biotechnological applications.

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## 6. SUPPLEMENTAL MATERIAL



**Supplemental Figure 2.1.** MS<sup>+</sup> spectra of compounds **1**, **2**, **3**, and **4**.



**Supplemental Figure 2.2.**  $MS^+$  spectra of compounds **5** and **6**.



**Supplemental Figure 2.3.** <sup>1</sup>H spectra (DMSO-*d*6) of kaempferol 3-*O*-(2"-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**).



Supplemental Figure 2.4. Zoom from 5.2 to 8.5 ppm, <sup>1</sup>H spectra (DMSO-*d*6) of kaempferol 3-O-(2"-galloyl)-β-D-glucopyranoside (13, 16 or 19).



**Supplemental Figure 2.5.** HSQC spectra (DMSO-*d*6) of kaempferol 3-*O*-(2"-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**).



Supplemental Figure 2.6. Zoom from 2.7 to 3.8 ppm, HSQC spectra (DMSO-*d*6) of kaempferol 3-O-(2"-galloyl)-β-D-glucopyranoside (13, 16 or 19).



**Supplemental Figure 2.7.** HMBC spectra (DMSO-*d*6) of kaempferol 3-O-(2"-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**).



**Supplemental Figure 2.8.** Zoom from 2.7 to 3.9 ppm, HMBC spectra (DMSO-*d*6) of kaempferol 3-O-(2"-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**).



**Supplemental Figure 2.9.** Zoom from 5.2 to 8.6 ppm, HMBC spectra (DMSO-*d*6) of kaempferol 3-O-(2"-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**).



**Supplemental Figure 2.10.** <sup>1</sup>H spectra (DMSO-*d*6) of kaempferol 3-*O*-(6"-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**).



**Supplemental Figure 2.11.** Zoom from 5.1 to 8.1 ppm, <sup>1</sup>H spectra (DMSO-*d*6) of kaempferol 3-O-(6"-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**).


**Supplemental Figure 2.12.** HSQC spectra (DMSO-*d*6) of kaempferol 3-O-(6"-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**).



**Supplemental Figure 2.13.** Zoom from 3.0 to 4.6 ppm, HSQC spectra (DMSO-*d*6) of kaempferol 3-O-(6"-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**).



**Supplemental Figure 2.14.** HMBC spectra (DMSO-*d*6) of kaempferol 3-O-(6"-galloyl)-β-D-glucopyranoside (**13**, **16** or **19**).



Supplemental Figure 2.15. Zoom from 2.8 to 4.9 ppm, HMBC spectra (DMSO-*d*6) of kaempferol 3-O-(6"-galloyl)-β-D-glucopyranoside (13, 16 or 19).



Supplemental Figure 2.16. Zoom from 5.2 to 8.4 ppm, HMBC spectra (DMSO-*d*6) of kaempferol 3-O-(6"-galloyl)-β-D-glucopyranoside (13, 16 or 19).



Supplemental Figure 2.17. <sup>1</sup>H spectra (DMSO-*d*6) of mixture (17).



Supplemental Figure 2.18. Zoom from 5.2 to 8.2 ppm, <sup>1</sup>H spectra (DMSO-*d*6) of mixture (17).



Supplemental Figure 2.19. <sup>13</sup>C spectra (DMSO-*d*6) of mixture (17).



Supplemental Figure 2.20. HMBC spectra (DMSO-*d*6) of mixture (17).



Supplemental Figure 2.21. Zoom from 2.4 to 4.5 ppm, HMBC spectra (DMSO-d6) of mixture (17).



Supplemental Figure 2.22. Zoom from 4.8 to 8.6 ppm, HMBC spectra (DMSO-d6) of mixture (17).



**Supplemental Figure 2.23.** <sup>1</sup>H spectra (DMSO-*d*6) of kaempferol 3-O-glucoronide-6"-O-methylester (23).



Supplemental Figure 2.24. Zoom from 5.1 to 8.1 ppm, <sup>1</sup>H spectra (DMSO-*d*6) of kaempferol 3-O-glucoronide-6"-O-methylester (23).



**Supplemental Figure 2.25.** HSQC spectra (DMSO-*d*6) of kaempferol 3-O-glucoronide-6"-O-methylester (23).



Supplemental Figure 2.26. Zoom from 2.9 to 4.5 ppm, HSQC spectra (DMSO-*d*6) of kaempferol 3-O-glucoronide-6"-O-methylester (23).



**Supplemental Figure 2.27.** HMBC spectra (DMSO-*d*6) of kaempferol 3-O-glucoronide-6"-O-methylester (23).



Supplemental Figure 2.28. Zoom from 2.7 to 4.0 ppm, HMBC spectra (DMSO-d6) of kaempferol 3-O-glucoronide-6"-O-methylester (23).



Supplemental Figure 2.29. Zoom from 5.0 to 8.6 ppm, HMBC spectra (DMSO-*d*6) of kaempferol 3-O-glucoronide-6"-O-methylester (23).



**Supplemental Figure 2.30.** <sup>1</sup>H spectra (DMSO-*d*6) of Quercetin 3-*O*-(6"-*p*-coumaroyl)-β-D-glucopyranoside (**25**).



**Supplemental Figure 2.31.** Zoom from 5.1 to 8.1 ppm, <sup>1</sup>H spectra (DMSO-*d*6) of Quercetin 3-O-(6"-*p*-coumaroyl)-β-D-glucopyranoside (**25**).



**Supplemental Figure 2.32.** HSQC spectra (DMSO-*d*6) of Quercetin 3-O-(6"-*p*-coumaroyl)-β-D-glucopyranoside (25).



**Supplemental Figure 2.33.** Zoom from 2.7 to 4.7 ppm, HSQC spectra (DMSO-*d*6) of Quercetin 3-*O*-(6"-*p*-coumaroyl)-β-D-glucopyranoside (**25**).



**upplemental Figure 2.34.** Zoom from 5.0 to 8.8 ppm, HSQC spectra (DMSO-*d*6) of Quercetin 3-O-(6"-*p*-coumaroyl)-β-D-glucopyranoside (**25**).



**Supplemental Figure 2.35.** <sup>1</sup>H spectra (DMSO-*d*6) of Kaempferol 3-O-(6"-*p*-coumaroyl)-β-D-glucopyranoside (**27**).



**Supplemental Figure 2.36.** Zoom from 5.1 to 8.1 ppm, <sup>1</sup>H spectra (DMSO-*d*6) of Kaempferol 3-O-(6"-*p*-coumaroyl)-β-D-glucopyranoside (**27**).



**Supplemental Figure 2.37.** HSQC spectra (DMSO-*d*6) of Kaempferol 3-*O*-(6"-*p*-coumaroyl)-β-D-glucopyranoside (**27**).



**Supplemental Figure 2.38.** Zoom from 3.0 to 4.6 ppm, HSQC spectra (DMSO-*d*6) of Kaempferol 3-O-(6"-*p*-coumaroyl)-β-D-glucopyranoside (**27**).



**Supplemental Figure 2.39.** HMBC spectra (DMSO-*d*6) of Kaempferol 3-*O*-(6"-*p*-coumaroyl)-β-D-glucopyranoside (**27**).



**Supplemental Figure 2.40.** Zoom from 2.9 to 4.8 ppm, HMBC spectra (DMSO-*d*6) of Kaempferol 3-O-(6"-*p*-coumaroyl)-β-D-glucopyranoside (27).



**Supplemental Figure 2.41.** Zoom from 5.2 to 8.4 ppm, HMBC spectra (DMSO-*d*6) of Kaempferol 3-O-(6"-*p*-coumaroyl)-β-D-glucopyranoside (27).



**Supplemental Figure 2.42.** <sup>1</sup>H spectra (DMSO-*d*6) of kaempferol (2**9**).



Supplemental Figure 2.43. Zoom from 6.1 to 8.2 ppm, <sup>1</sup>H spectra (DMSO-*d*6) of kaempferol (29).



Supplemental Figure 2.44. <sup>13</sup>C spectra (DMSO-*d*6) of kaempferol (29).



Supplemental Figure 2.45. HSQC spectra (DMSO-d6) of quercetin.



Supplemental Figure 2.46. Zoom from 3.2 to 4.0 ppm, HSQC spectra (DMSO-d6) of quercetin.



Supplemental Figure 2.47. Zoom from 5.4 to 8.8 ppm, HSQC spectra (DMSO-d6) of quercetin.
# Chapter 3 – The regulation of floral color change in *Tibouchina pulchra*.

# 1. INTRODUCTION

*Tibouchina pulchra* (Cham.) Cogn. (Melastomataceae, Myrtales), named as "manacá da serra", is a native species from the Atlantic Forest. Its use for urban ornamentation is increasing due to its beautiful flowers. In this sense, the ornamental dwarf cultivar of *T. pulchra* ("manacá-anão") blooms with the same temporal pattern of floral color as the wild parental: buds, first stage (S1), open as white flowers (S2) that, one day after, become light pink (S3), further, flowers are deep pink (S4), from the third day towards senescence (Figure 3.1). As reviewed in chapter 1, the temporal alteration of flower color can be regulated by pollination, gene expression regulation, the presence of copigments and flavonoid-metal complexation, resulting in the differential accumulation of phenolic acids, flavonols, and anthocyanins extensively characterized in chapter 2.



Figure 3.1. Stages of floral color change of *T. pulchra*. S1- buds (day 0), S2- white flowers (day 1), S3- light pink (day 2), S4- dark pink (day 3).

*T. pulchra* flowers are nectarless, with poricidal anthers, hermaphrodite, heterantherous, herkogamous and self-compatible (Renner 1989, Brito & Sazima 2012). Concerning pollinators, Pereira *et al.* (2011) observed fifteen species of floral visitors in *T. pulchra*, being only four described as effective pollinators (*Bombus morio, B. pauloensis, Xylocopa frontalis* and *X. brasilianorum*). These large bees interact with poricidal anthers transferring the pollen to the stigmas by vibration. *Xylocopa* species collected pollen exclusively from white flowers, while *Bombus* visited pink flowers only at the end of the flowering season, when white flowers are scarce. Moreover, by performing controlled pollination experiments, it was demonstrated that the pollen viability, stigma receptivity and fruit setting were similar in white and pink flowers. This data together with the bee color preference described above suggest that flower color

change is not associated to pollination but, according to Da Silva (2006), may enlarge the visitor diversity.

As aforementioned, the pigments responsible for the color of *T. pulchra* flowers are phenolic acids and flavonoids (flavonols and anthocyanins). These phenolic compounds are produced by the combination of products from shikimic (SKM) and malonate (MAL) pathways, whose building blocks are derived from primary metabolism (Figure 3.2). Many simple benzoic acid derivatives (*e.g.* gallic acid,  $C_6C_1$ ) and phenylpropanoids (*e.g. p*-coumaric,  $C_6C_3$ ) are produced by SKM pathway, while flavonoids ( $C_6C_3C_6$ ) are biosynthesized by intermediates from both, SKM and MAL pathways (Davies *et al.* 2009).

PHENYLALANINE AMMONIA LYASE (PAL) is the first enzyme in the phenylpropanoid and flavonoid pathway, it catalyzes the nonoxidative deamination of phenylalanine yielding *trans*-cinnamic acid. CINNAMATE 4-HYDROXYLASE (C4H) is a CYTOCHROME P450 MONOOXYGENASE (P450) enzyme involved in the hydroxylation of *trans*-cinnamic acid at C4 position to produce *p*-coumaric acid. This is a branching point that can channel the metabolic flux towards phenylpropanoids or flavonoids. By the action of COUMARATE 3-HYDROXYLASE (C3H), cinnamic acid derivatives as caffeic, ferulic, and sinapic acids are synthetized. On the other hand, *p*-coumaric acid is activated to the corresponding Co-A thioester facilitating further conversion to flavonoids, this reaction is catalyzed by 4-COUMARATE:CO-A LIGASE (4CL) (Dewick 2009, Saito *et al.* 2013).

The first produced flavonoid class comprises the chalcones, which are formed by the action of CHALCONE SYNTHASE (CHS). This enzyme belongs to a family of type III polyketide synthases and is the first committed enzyme in biosynthesis of flavonoids. Three sequential additions of the extender molecule malonyl-CoA in a *p*-coumaroyl-CoA is the reaction catalyzed by CHS (Dewick 2009, Saito *et al.* 2013). Chalcones are the first colored flavonoid, and provide yellow coloration to petals of a few plant species (*e.g. Dianthus caryophyllus* L, Caryophyllaceae, Caryophyllales) (Ono *et al.* 2006a, 2006b). CHALCONE ISOMERASE (CHI) acts in the stereospecific cyclization of naringenin chalcone to flavanone. FLAVANONE 3-HYDROXYLASE (F3H) is responsible for the oxygenation at C3 position of flavanone to form dihydroflavonol (dihydrokaempferol). FLAVONOID 3'-HYDROXYLASE (F3'H) and FLAVONOID 3'-5'-HYDROXYLASE (F3'5'H) are P450 that catalyze the hydroxylation at C3'and C3'-C5'-positions of the B-ring of flavonoid, respectively, rendering dihydroflavonols.

FLAVONOL SYNTHASE (FLS) and DIHYDROFLAVONOL REDUCTASE (DFR) are competitors for the same substrate, dihydroflavonol, a branching point for flavonols and anthocyanins formation (Pelletier *et al.* 1997). FLS catalyzes the double bond formation between C2 and C3, while DFR reduces the C4-keto group of dihydroflavonol to the corresponding leucoanthocyanidin. FLS, F3H and ANTHOCYANIDIN SYNTHASE (ANS) are the three 2-OXOGLUTARATE/FE(II)-DEPENDENT DIOXYGENASE (2-ODD) enzymes in the flavonoid biosynthesis pathway. The enzyme ANS catalyzes the formation of anthocyanidin from leucoanthocyanidins exhibit a conjugated double bond system encompassing A, B and C rings, the later with two double bonds carrying a positive charge. Anthocyanidins are the chromophore moiety of anthocyanins, which are formed after glycosylation, acylation and/or methylation by the action of specific enzymes as FLAVONOID GLUCOSYLTRANSFERASE (FGT), ANTHOCYANIDIN ACYLTRANSFERASE (ACT) and/or *O*-METHYLTRANSFERASE (OMT) (Figure 3.2). In *Arabdopsis thaliana* (L.) Heynh. (Brassicaceae, Brassicales), FGT can transfer glucose to the C3 position of both, flavonol and anthocyanin aglycones (Yonekura *et al.* 2012).

Due to anthocyanins appeal as natural food colorant, nutraceutic and bioactive compounds (Norberto *et al.* 2013, De Ferrars *et al.* 2014, Xiaonan *et al.* 2016, Gowd *et al.* 2017), the regulation of anthocyanin biosynthetic pathway has been extensively explored. Most of the structural genes have been isolated (Winkel-Shirley 2001) and several regulatory genes have been characterized. Petunia (*Petunia hybrid* Vilm., Solanaceae, Solanales) and snapdragon (*Antirrhinum majus* L., Plantaginaceae, Lamiales) are the major model species for the study of anthocyanin biosynthesis, especially in floral tissues, however, maize (*Zea mays* L., Poaceae, Poales) and *Arabdopsis thaliana* have been extensively studied as well (Cone *et al.* 1986, Goodrich *et al.* 1992, Quattrocchio *et al.* 1998, 1999, Broun 2005, Dixon *et al.* 2005, Koes *et al.* 2005, Grotewold 2006).

Regarding the position in the pathway, structural genes can be grouped into two classes: early biosynthetic genes (*e.g. PAL, C4H, CHS*, and *FLS*) and late biosynthetic genes (*e.g. DFR* and *ANS*) (Tornielli *et al.* 2009). The early genes of anthocyanin pathway were functionally characterized in many plant species such as *A. thaliana, Solanum lycopersicum* L. (Solanaceae, Solanales), *Vitis vinifera* L. (Vitaceae, Vitales), *P. hybrida*, and *A. majus* (Sparvolli *et al.* 1994, Brugliera *et al.* 1999, Mol *et al.* 1983, Guo *et al.* 2009, Huang *et al.* 2010), while late genes (*DFR* and *ANS*) have been characterized mostly in *P. hybrid* and *A. thaliana*. (Pelletier *et al.* 1997, Devic *etal.* 1999). However, much less is known about the subsequent modifications

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(substituents), such as glycosylation, acylation, and methylation (Grotewold 2006). In fact, there are several types of substituents that guarantee the chemical diversity in flower color. For example, *P. hybrida* synthesizes methylated anthocyanins with an acylated rutinose moiety, while *A. thaliana* accumulates more complex anthocyanins with seven sugar and acyl moieties but no methyl groups (Brouillard *et al.* 2003, Saito *et al.* 2013).



**Figure 3.2.** Simplified schematic representation of flavonoid biosynthetic pathway. Primary metabolism is highlighted in green. Abbreviations indicate the following enzymes: 4-COUMARATE:CO-A LIGASE (4CL), ANTHOCYANIDIN ACYL TRANSFERASE (AACT), **ANTHOCYANIDIN SYNTHASE (ANS)**, CAFFEIC ACID *O*-METHYLTRANSFERASE (COMT), **CINNAMATE 4-HYDROXYLASE (C4H)**, CHALCONE ISOMERASE (CHI), CHALCONE REDUCTASE (CHR), **CHALCONE SYNTHASE (CHS)**, Coenzyma A (CoA), COUMARATE 3-HYDROXYLASE (C3H), DIHYDROFLAVONOL 4-REDUCTASE (DFR), erythrose 4-phosphate (E4P), FLAVANONE 3-HYDROXYLASE (F3H), FLAVONE SYNTHASE (FS), FLAVONOID 3'-HYDROXYLASE (F3'H), FLAVONOID GLYCOSYL TRANSFERASE (FGT), **FLAVONOL SYNTHASE (FLS)**, HSCoA: coenzyme A, ISOFLAVONE SYNTHASE (IFS), malonate (MAL), *O*-METHYLTRANSFERASE (OMT), **PHENYLALANINE AMMONIA LYASE (PAL)**, phosphoenolpyruvate (PEP), SHIKIMATE DEHYDROGENASE (SDH), and shikimate (SKM). The enzymes studied in the present work are highlighted in bold. Dotted lines represent more than one enzymatic reaction. Adapted from Dewick *et al.* (2009).

Numerous studies have demonstrated the transcriptional regulation of anthocyanin biosynthesis. A transcription factor complex (MBW complex) composed by a MYELOBLASTOSIS (MYB), a BASIC HELIX-LOOP-HELIX (bHLH), and a BETA-TRANSDUCIN REPEAT (WD40) protein, has shown to regulate the expression of the structural genes (Koes et al. 2005, Hichri et al. 2011, Xu et al. 2014, 2015). In vegetative tissues of A. thaliana, an eudicot, anthocyanin early biosynthetic genes are co-activated by bHLH and WD40 transcription factors in a MYBindependent manner, while the late biosynthetic genes require all three proteins to form the active MBW complex. Interestingly, in maize, a monocot, early and late biosynthetic genes are transcriptionally activated exclusively by MYB and bHLH proteins (Petroni & Tonelli 2011). By the characterization of mutant genotypes and transgenic lines, the transcriptional induction of anthocyanin biosynthesis in flowers has been addressed in several species, where distinct combinations of the aforementioned transcriptional factors have shown to be needed to determine floral pigmentation. MYB transcription factors induce early and late biosynthetic genes in Nicotiana tabacum L. (Solanaceae, Solanales), Brassica oleracea L. (Brassicaceae, Brassicales), Anthurium andraeanum Linden ex André (Araceae, Alismatales), Mimulus lewisii Pursh (Scrophulariaceae, Lamiales), and M. cardinalis Douglas ex Benth. flowers (Chiu et al. 2010, Pattanaik et al. 2010, Li et al. 2016, Yuan et al. 2016). In Ipomea purpurea (L.) Roth (Convolvulaceae, Solanales) transposon-insertion double mutants for bHLH and MYB genes resulted in snow-white flowers where the expression of the late biosynthetic genes was completely abolished (Park et al. 2007).

Additionally, the existence of post-transcriptional regulatory mechanisms has also been verified to control anthocyanin biosynthesis. In *A. thaliana*, it has been reported that the expression of the anthocyanin-associated MYB transcription factors is regulated by miRNAs and tasiRNAs-mediated hetero-silencing (Rajagopalan *et al.* 2006, Hsieh *et al.* 2009). In particular, the miR156 targets SQUAMOSA PROMOTER BINDING PROTEIN-LIKE (SPL) mRNA, whose product destabilizes the MBW transcriptional activation complex. Thus, increased miR156 activity promotes the accumulation of anthocyanins by the MBW complex-mediated transcriptional induction of the late biosynthetic genes (Gou *et al.* 2011). Moreover, in *Petunia*, *CHS* has also shown to be post-transcriptionally regulated by siRNA (Koseki *et al.* 2005). Interestingly, anthocyanin accumulation in maize kernels is controlled by the regulation of the splicing patter of *Bz2* gene, a GLUTATHIONE S-TRANSFERASE (GST) that mediates transfer of cytoplasmic anthocyanin into the vacuole (Pairoba & Walbot 2003).

Finally, post-translational regulation has also been reported for anthocyanin biosynthesis. In *A. thaliana*, KELCH REPEAT F-BOX proteins have shown to physically interact with four PAL isozymes inducing their 26S proteasome- mediated ubiquitination, resulting in the downregulation of phenylpropanoid biosynthesis (Zhang *et al.* 2013).

Although for some biological systems the regulation of anthocyanin biosynthesis has been elucidated, including the identification of master transcription factors (Butelli *et al.* 2008), many gaps remain unsolved for the complete understanding of the mechanisms underneath the accumulation of these pigments. Improving the knowledge about the multilevel regulation of gene expression will favor the development of new biotechnological tools for the generation of new varieties with enhanced health-promoting properties and visual appeal (Hichri 2011).

Some studies showed that metals can influence the color by establishing a flavonoidmetal complex. In cornflowers (*Centaurea sp.*, Asteraceae, Asterales), the bright blue color is the result of a supramolecular structure with stacking between copigments, pigments and metals: six molecules of apigenin, six molecules of cyanidin and four metal ions (Fe<sup>3+</sup>, Mg<sup>2+</sup> and two Ca<sup>2+</sup>) (Shiono *et al.* 2005). Hortensia flowers (*Hydrangea macrophylla* (Thumb.) Ser., Hydrangeaceae, Cornales) are the best-known example of flavonoid-metal complexation influencing the corolla color. Under high aluminum content and acidity of the soil, the flowers are blue due to a nonstoichiometric complex with Al<sup>3+</sup>, delphinidin 3-glucosides, quinic acid derivatives, while upon low aluminum concentrations, the same metal-pigment-copigment complex became light pink, as the result of soil pH change (Yoshida *et al.* 2003, Kondo *et al.* 2005, Oyama *et al.* 2015, Trouillas *et al.* 2016).

The pH affects the final color of flower petals by determining the conformation and absorption spectrum of anthocyanins. The light blue petals of morning glory (*Ipomoea tricolor* Cav., Convolvulaceae, Solanales) owe their color to the effect of the alkaline pH on the accumulated anthocyanins. The closed buds of these flowers are purplish red and their cells have a pH of 6.6. However, when the flowers open, the petal cell pH increases up to 7.7, and the pigment changes color to sky blue (Yoshida *et al.* 1995). Recently, studying the genetic determinants of *P. hybrida* flower pigmentation, Verweij *et al.* (2016) have identified PH3 protein, a WRKY-type transcriptional factor that regulates the expression of the proteins responsible for the vacuolar acidification.

The well-determined stages of flower coloring in *T. pulchra*, raise interesting questions about the genetic and chemical mechanisms involved in the regulation of this phenomenon,

which has been unexplored up to now. Here, we addressed this issue by cloning key flavonoid biosynthetic genes and profiling mRNA levels, pigment composition and metal content along flower development.

#### 2. MATERIAL AND METHODS

#### 2.1. Plant material

Petals at S1 to S4 stages (Figure 3.1) were sampled from five *Tibouchina pulchra* (cv. "manacá-anão") plants located in Praça Carlos José Gíglio (Latitude: -23.57998, Longitude: -46.73403) in the most vigorous flowering period (May and June/2016) between 8 and 9 a.m. For each developmental stage a pool of petals (~ 5 g) from each plant was considered a biological replicate (n = 5). Samples were collected with gloves, in a sterile tube and immediately frozen in liquid nitrogen. Petals were stored at -80°C for mRNA extraction, or freeze dried in a lyophilizer (K202, Liobras) for chemical analysis. All material was crushed in a ball mill (TissueLyser, Qiagen) for further analyses. For gene cloning a pool of all stages collected in May and June/2014 was used. An exsiccate was deposited on herbarium of Institute of Bioscience (SPF) of University of São Paulo (ID: Furlan73).

#### 2.2. Pigment profile

Phenolic profile was assessed analyzing petal extracts by Ultra Performance Liquid Chromatography (UPLC) system with Diode Array Detector (DAD) (Dionex Ultimate 3000) and Electrospray Ionization Quadrupole Time-of-Flight High-Resolution Mass Spectrometry (ESI-QTOF-HRMS) detector (Bruker Maxis). The extraction was performed as described in chapter 2 (item 2.2). To quantify the compounds, the areas of MS<sup>+</sup> chromatograms were compared with standard curves of *p*-coumaric acid, kaempferol and cyanidin. This analysis was performed in the Department of Chemistry at Denmark Technical University under the supervision of Ph.D. Mads H. Clausen.

## 2.3. Gene cloning and expression

## 2.3.1. Sequence analysis for primers design

The sequences of the functionally characterized genes *PAL* (AT2G37040.1), *C4H* (AT2G30490.1), *CHS* (AT5G13930.1), *FLS* (AT5G08640.1) and *ANS* (AT4G22880.1) from *Arabdopsis thaliana* (Wanner *et al.* 1995, Bell-Lelong *et al.* 1997, Pelletier *et al.* 1997, Feinbaum & Ausubel 1998) were used as query to identified, by tBLASTx program (Altschul *et al.* 1997), the orthologous sequences from *Vitis vinifera*, *Eucalyptus grandis* W.Hill (Myrtaceae, Myrtales), and *Solanum lycopersicum* fully sequenced genomes available in the Phytozome v 9.0 database

(http://www.phytozome.net/). The amino acid coding sequences were aligned using the MUSCLE program following the standard configurations of the MEGA 6 package (Tamura *et al.* 2011). After the identification of conserved domains, the alignments were converted to nucleotides for degenerated primer design based on *Eucalyptus grandis* sequences, which is the *T. pulchra* closest related species with completely sequenced genome (Supplemental Figure 3.1 and Supplemental Table 3.2). The quality of the designed primers was verified using the software Oligo Analyzer 3.1 (http://www.idtdna.com).

Additionally, using the same strategy described above, primers were designed for the cloning of two reference genes for Reverse Transcriptase Quantitative Polymerase Chain Reaction (RT-qPCR) normalization. The chosen genes were those previously described for gene expression analyses in *Petunia hybrida* flowers, *ELONGATION FACTOR 1* $\alpha$  (*EF1*) and *RIBOSOMAL PROTEIN S13* (*RPS*) (Mallona *et al.* 2010).

#### 2.3.2. RNA extraction and cDNA synthesis

Total RNA from 3 g of petal samples was extracted with CTAB (Cetyl trimethylammonium bromide- Chang *et al.* 1993) (Supplemental Figure 3.2). DNA was removed with 100 U of amplification-grade DNase (Invitrogen) following the recommended protocol. cDNA was synthesized from 1 µg of RNA using oligo dT (for gene cloning) or random primers (for RT-qPCR) and the SuperScript III kit (Invitrogen). cDNA quality was confirmed by PCR using *RPS* constitutive gene primers.

## 2.3.3. Cloning

Gene fragments were amplified by PCR in a total volume of 50  $\mu$ L containing 0.2 mM each dNTPs, 0.2  $\mu$ M each primer (Supplemental Table 3.2), 1X reaction buffer, 1.5 mM MgCl<sub>2</sub>, 50 ng of cDNA, and 2 U of Taq DNA polymerase (Invitrogen). The amplification conditions were: 94°C for 3 min, 35 cycles of 94°C for 30 s, 50°C for 30 s, and 72°C for 1 min, and then 72°C for 10 min. Amplification products were purified (Kit GFX Amersham Biosciences) and cloned into a TOPO-TA vector (Dual promoter Kit Invitrogen) following the recommended protocols. Transformation was carried by using 50  $\mu$ L of *Escherichia coli* DH10B competent cells with 2  $\mu$ L of the ligation product. The mixture was kept on ice for 30 min, and subjected to a thermal shock of 45 sec at 65°C and 2 min on ice. 500  $\mu$ L of SOC (Super Optimal Broth) medium (Hanahan 1983) was added for a shaking incubation (45 min at 37°C). Samples were centrifuged (3 min, 3,000 rpm at room temperature) and cells resuspended and plated on Luria-Bertani (LB)

agar (Bertani *et al.* 1951) with kanamycin, X-Gal (5-bromo-4-chloro-3-indoxyl- $\beta$ -D-galactopyranoside) and IPTG (isopropil  $\beta$ -D-1-tiogalactopiranosida). Five positive colonies for each gene were grown in liquid LB and plastidial DNA was purified and sequenced with vector universal primers.

#### 2.3.4. Sequence analyses

To verify the identity of the cloned *T. pulchra* gene fragments, phenetic analyses were performed. Orthologous sequences from *Arabdopsis thaliana* (AT) and *Brassica rapa* L. (Brara)-Brassicaceae, Brassicales, *Eucalyptus grandis* (Eucgr)- Myrtaceae, Myrtales, *Mendicago trunculata* (Medtr) and *Trifolium pratense* L. (Tp) – Fabaceae, Fabales, and *Solanum lycopersicum* (Soly) – Solanaceae, Solanales, were retrieved from Phytozome v 12.1 database (http://www.phytozome.net /) by tBLASTx program (Altschul *et al.* 1997). The species were selected according to the following criteria: i) fully sequenced genome, ii) genes belonging to gene families whose functional annotation was associated to phenolic compound biosynthesis (gene family 94476114 for *PAL*, 94469571 for *C4H*, 94475332 for *CHS*, 94475392 for *FLS*, 94470119 for *ANS*, 94475526 for *RPS* and 94476231 for *EF1*), and iii) species placed in Rosids clade in APG IV classification (AT, Brara, Eucgr, Medtr and Tp, and Soly as outgroup). The coding sequences were aligned using the Clustal program following the standard configurations of the MEGA 6 package (Tamura *et al.* 2011), with manual verification of each codon according to the amino acid sequence. The phenograms were constructed with the following parameters: *Neighbor-joining, Bootstrap* of 1,000 replicates and the best model test for each analysis.

#### 2.3.5. RT-qPCR

Based on *T. pulchra* sequences, gene specific primers for expression analysis were designed to amplify fragments of approximately 160 bp (Supplemental Table 3.2). To quantify mRNA levels by RT-qPCR, reactions were carried out in a QuantStudio 6 Flex Real-Time PCR system (Applied Biosystems) using 2X Power SYBR Green Master Mix reagent (Life Technologies) in 14  $\mu$ L final volume. Absolute fluorescence data were analyzed using the LinRegPCR software package (Ruijter *et al.* 2009) to obtain quantitation cycle (Cq) values and calculate primer efficiencies (the primer efficiency values ranged from 1.85 to 1.98). Expression values were normalized against the geometric mean of the two reference genes (*EF1* and *RPS*), according to Quadrana *et al.* (2013). A permutation test lacking sample distribution assumption

(Pfaffl *et al.* 2002) was applied (formula below) to detect statistical differences (P<0.05) in expression ratios using the algorithms in the fgStatistics software package (Di Rienzo 2009).

$$RE = \frac{Pe_{goi}^{(control Ct mean- sample Ct mean)}}{Pe_{ref}^{(control Ct mean- sample Ct mean)}}$$

RE: relative expression, Pe: primer efficiency, goi: gene of interest, Ct: treshold cycle, control Ct: Ct mean of control biological replicates, sample Ct: Ct mean of sample biological replicates, ref: reference gene.

#### 2.4. Metal content

To analyze the content of metals 500 mg of freeze-dried powder petals were sent to Laboratory of Vegetal Tissue (ESALQ, USP). After a nitro-perchloric digestion, contents of Mg, K, Na, Ca, Cu, Mn, Zn and Fe<sup>3+</sup> were determined by Atomic Absorption Spectroscopy (AAS).

#### 2.5. Data analyses

Statistical analysis for pigment and metal profiles were performed using Infostat package (Di Rienzo *et al.* 2011). When the data set showed normality, ANOVA followed by Tukey test (P<0.05) were performed to compare differences between groups. In the absence of normality, a non-parametric comparison was performed by applying the Kruskal-Wallis test (P<0.05). All values represent the mean of five biological replicates.

#### 3. RESULTS

#### 3.1. Pigment profile

Pigment profile revealed that *T. pulchra* flowers contain a diverse mixture of six phenolic acids, twenty-two flavonols and two anthocyanin (Table 3.1)- see details about identification in chapter 2. Flavonols were the most abundant pigments ranging from 20 to 25 mg g<sup>-1</sup> DW, followed by phenolic acids that varied between 10 and 18 mg g<sup>-1</sup> DW and anthocyanins displaying values of 0.028 to 0.076 mg g<sup>-1</sup> DW.

Pigment abundance along flower development revealed that the amount of total phenolic acids increases during the transition from S1 to S2 maintaining constant levels from S2 onwards. This increment is due to the accumulation of the non-identified compound **6**. Regarding flavonols, except for kaempferol *p*-coumaroylhexoside (compound **30**), which showed increment from S2 to S3, the other twenty-one compounds showed a decreasing trend along flower development. Both identified anthocyanins were detected from the S3 stage exhibiting a significant increment in S4.

Table 3.1 - Pigment	profile in each develo	pmental stage of T.	pulchra flowers.
			,

Compound	S1 (mg g <sup>-1</sup> DW)	S2 (mg g⁻¹DW)	S3 (mg g⁻¹DW)	S4 (mg g⁻¹DW)
1- Phenolic acid	4.43 ± 2.96	5.9 ± 2.67	3.76 ± 2.46	7.6 ± 3.08
2- Cinnamic acid derivative	3.89 ± 2.28	4.18 ± 3.85	6.16 ± 1.50	3.53 ± 3.41
3- Phenolic acid	$0.84 \pm 0.91$	$1.55 \pm 0.17$	1.85 ± 0.20	$1.92 \pm 0.21$
4- Phenolic acid	0.38 ± 0.75	0.14 ± 0.22	0.14 ± 0.21	0.15 ± 0.20
5- Cinnamic acid derivative	0.23 ± 0.23	0.57 ± 0.62	0.22 ± 0.16	0.26 ± 0.27
6- Phenolic acid	0.90 <sup>b</sup> ± 0.61	5.24 <sup>a</sup> ± 1.33	4.19 <sup>a</sup> ± 1.08	4.23 <sup>a</sup> ± 0.94
Phenolic acids total	10.69 <sup>b</sup> ± 3.43	17.58° ± 2.66	16.39 <sup>a</sup> ± 2.24	17.69 <sup>a</sup> ± 0.49
Compound	S1 (mg g <sup>-1</sup> DW)	S2 (mg g <sup>-1</sup> DW)	S3 (mg g⁻¹DW)	S4 (mg g <sup>-1</sup> DW)
7- N.I.	$0.40 \pm 0.41$	$0.70 \pm 0.13$	$0.49 \pm 0.40$	0.57 ± 0.34
8- Myricetin hexoside	0.77 ± 0.47	$0.56 \pm 0.19$	$0.56 \pm 0.08$	$0.58 \pm 0.18$
9- Myricetin hexoside	$0.29 \pm 0.21$	0.22 ± 0.13	0.23 ± 0.07	0.27 ± 0.06
10- Quercetin galloylhexoside	$0.41 \pm 0.30$	$0.22 \pm 0.10$	$0.23 \pm 0.06$	$0.23 \pm 0.16$
11- Quercetin hexoside	$1.26 \pm 0.65$	$1.02 \pm 0.37$	$0.98 \pm 0.19$	$0.94 \pm 0.16$
12- Quercetin glucuronide	$0.16 \pm 0.07$	$0.06 \pm 0.03$	$0.07 \pm 0.04$	$0.11 \pm 0.06$
13- Kaempferol galloylhexoside	$1.61 \pm 0.85$	$1.36 \pm 0.61$	$1.34 \pm 0.22$	1.24 ± 0.33
14- N.I.	$0.19 \pm 0.12$	$0.14 \pm 0.08$	$0.14 \pm 0.08$	$0.17 \pm 0.09$
15- Kaempferol hexoside	$3.40 \pm 1.30$	3.20 ± 0.91	3.12 ± 0.55	2.92 ± 0.42
16- Kaempferol galloylhexoside	2.16 ± 0.94	$1.78 \pm 0.72$	$1.78 \pm 0.46$	$1.70 \pm 0.23$
17- Kaempferol 3-O-β-D-glucopyranoside	6.48 ± 1.51	6.32 ± 0.85	6.44 ± 0.50	6.21 ± 0.47
18- Kaempferol pentoside	2.75 ± 1.43	2.41 ± 1.07	2.23 ± 0.63	$1.96 \pm 0.30$
19- Kaempferol galloylhexoside	$0.24 \pm 0.14$	$0.22 \pm 0.11$	$0.18 \pm 0.07$	$0.16 \pm 0.04$
20- Kaempferol pentoside	$0.23 \pm 0.15$	$0.15 \pm 0.10$	$0.14 \pm 0.07$	$0.11 \pm 0.05$
21- Kaempferol p-coumaroylhexoside	0.53 ± 0.29	0.46 ± 0.26	$0.40 \pm 0.18$	0.29 ± 0.07
23- Mixture- Kaempferol 3-O-glucuronide-6''-O-methylester / Petunidin derivative	$0.28 \pm 0.15$	0.29 ± 0.09	$0.21 \pm 0.01$	$0.17 \pm 0.05$

#### Table 3.1. (Continued)

Compound	S1 (mg g⁻¹DW)	S2 (mg g <sup>-1</sup> DW)	S3 (mg g⁻¹DW)	S4 (mg g⁻¹DW)
25- Quercetin 3-O-(6''-O- <i>p</i> -coumaroyl)-β-D-glucopyranoside	$0.68 \pm 0.38$	$0.52 \pm 0.19$	$0.44 \pm 0.08$	$0.40 \pm 0.05$
26- N.I.	$0.43 \pm 0.21$	$0.35 \pm 0.20$	$0.31 \pm 0.13$	0.26 ± 0.04
27- Kaempferol 3-O-(6''-O-p-coumaroyl)-β-D-glucopyranoside	$1.44 \pm 0.65$	$1.20 \pm 0.46$	$1.05 \pm 0.00$	$0.94 \pm 0.12$
28- Kaempferol p-coumaroylhexoside	$0.19 \pm 0.09$	$0.16 \pm 0.06$	$0.10 \pm 0.03$	$0.08 \pm 0.02$
29- Kaempferol	$0.94 \pm 0.48$	$1.05 \pm 0.44$	0.83 ± 0.20	$0.71 \pm 0.08$
30- Kaempferol <i>p</i> -coumaroylhexoside	$0.00^{b} \pm 0.00$	0.08 <sup>b</sup> ± 0,00	0.13 <sup>a</sup> ± 0.03	0.13 <sup>a</sup> ± 0.03
Flavonol total	24.99 ± 10.31	22.17 ± 6.66	21.25 ± 3.13	19.94 ± 2.64
Compound	S1 (µg g⁻¹DW)	S2 (µg g⁻¹DW)	S3 (µg g⁻¹DW)	S4 (µg g⁻¹DW)
22- Petunidin <i>p</i> -coumaroylhexoside acetylpentoside	0.00 <sup>c</sup> ± 0.00	0.00 <sup>c</sup> ± 0.00	8.43 <sup>b</sup> ± 2.21	22.42 <sup>a</sup> ± 8.00
24- Malvidin p-coumaroylhexoside acetylpentoside	0.00 <sup>c</sup> ± 0.00	0.00 <sup>c</sup> ± 0.00	19.87 <sup>b</sup> ± 5.28	53.72 <sup>a</sup> ± 20.18
Anthocyanin total	0.00 <sup>c</sup> ± 0.00	0.00 <sup>c</sup> ± 0.00	28.29 <sup>b</sup> ± 7.47	76.14 <sup>a</sup> ± 28.11

\* S1- buds (day 0), S2- white flowers (day 1), S3- light pink (day 2), S4- dark pink (day 3). DW- Dry weight. NI- not identified. Compounds highlighted in bold showed statistically significant differences between stages. Letters indicate statistically significant different values.

## 3.2. Gene cloning and expression analysis

In order to analyze the transcriptional profile of flavonoid biosynthetic genes and since no information about genome sequence is available for *T. pulchra*, the first aim proposed was the cloning of partial gene sequences for some key enzymes. The genes encoding for PAL, C4H, CHS, FLS, and ANS enzymes were chosen because they have been reported to be transcriptional regulated in other species. Additionally, two constitutive genes previously used for expression analyses in *Petunia hybrida* were also selected to be used as reference genes in RT-qPCR experiments. The gene fragment cloning was based on the identification of conserved protein domains and degenerated primers were designed based on the *E. grandis* paralog sequences, the *T. pulchra* most closely related species with complete sequenced genome (see item 2.3.1 in Material and Methods). Gene fragments with approximately the predicted size were amplified (Supplemental Figure 3.3), cloned and sequenced (Supplemental Figure 3.4).

The identity of the obtained gene fragments was corroborated by a phenetic analysis. Although the topology did not strictly follow the species phylogenetic relationship, species belonging to the same order grouped together: *B. rapa* and *A. thaliana* (Brassicales), *M. trunculata* and *T. pratense* (Fabales), and *T. pulchra* and *E. grandis* (Myrtales) (Figure 3.3). It is worth mentioning that the aim of this analysis was exclusively to confirm the identity of the cloned sequences corroborating its homology to sequences predicted as the corresponding genes. All five *T. pulchra* sequences grouped together to the corresponding putative homologous from *E. grandis* demonstrating that the strategy was successful to clone partial sequences of flavonoid biosynthetic genes.



**Figure 3.3.** Phenograms for *PHENYLALANINE AMMONIUM LYASE (PAL)* (A), *CINAMMATE 4-HYDROXYLASE (C4H)* (B), *CHALCONE SYNTHASE (CHS)* (C), *FLAVONOL SYNTHASE (FLS)* (D), and *ANTHOCYANIDIN SYNTHASE (ANS)* (E) genes. Sequences from species with full sequenced genomes were included in the analyses: *Eucalypstus grandis* (Eucgr), *Arabdopsis thaliana* (AT), *Brasica rapa* (Brara), *Mendicago trunculata* (Medtr), *Trifolium pratense* (Tp) and *Solanum lycopersicum* (Soly). Trees were constructed using the following parameters: *Neighbor-joining, Bootstrap* of 1,000 replicates and the best model test for each analysis. Grey boxes show the clusters for *T. pulchra* (T.pulchra) and *E. grandis* sequences.



**Figure 3.4.** Expression profile of *PHENYLALANINE AMMONIUM LYASE (PAL), CINAMMATE 4-HYDROXYLASE (C4H), CHALCONE SYNTHASE (CHS), FLAVONOL SYNTHASE (FLS),* and *ANTHOCYANIDIN SYNTHASE (ANS)* genes along flower development. Different letters indicate statistically different relative expression ratios. S1- buds (day 0), S2- white flowers (day 1), S3- light pink (day 2), S4- dark pink (day 3).

То address whether the transcriptional regulation of gene expression is involved in T. pulchra flower color change, the mRNA accumulation of the cloned enzyme-encoding genes was profiled in four flower stages (S1 to S4) analyzed (Figure 3.4). As expected, since the gene fragments were cloned from petal cDNA, the presence of mRNA from all five genes was identified. With the exception of C4H that maintained constant levels of mRNA, the other four genes analyzed showed to be transcriptionally regulated along flower development.

The expression of the first committed gene in the phenolic compound biosynthetic pathway, PAL, peaked in the white opened flower (S2), while CHS mRNA reached its highest level at the first pink stage (S3). FLS, whose product diverges the route towards flavonols synthesis, showed maximum expression level in buds (S1), decreasing onwards. Finally, the ANS enzyme encoding gene, responsible for the synthesis of the anthocyanin chromophores, is 30-fold upregulated from white to pink stages (S2 to S3) and remains highly expressed until dark pink stage (S4).

#### 3.3. Metal content

Due to the eventual impact of metal concentration on pigment color, the metal content was also profiled along flower development. Among the analyzed metals, Fe<sup>3+</sup> and K were the most abundant macro and micronutrients, respectively (Figure 3.5). The only metal that showed fluctuation between flower stages was Fe<sup>3+</sup>, which increased from S3 to S4.



**Figure 3.5.** Metal quantification along *T. pulchra* flower development. Different letters indicate statistically different values. S1- buds (day 0), S2- white flowers (day 1), S3- light pink (dav 2), S4- dark pink (dav 3).

#### 4. DISCUSSION

Flower color change is a widespread phenomenon in angiosperms and poorly studied from the chemical and molecular point of view. In this work, the phenomenon of *T. pulchra* floral color change was addressed by a comprehensive analysis of pigments, gene expression and metal profiling along flower development.

In general, phenolic acids and, especially, flavonols are associated to white flowers (*e.g. Petunia hybrid* (Saito *et al.* 2006), *Trifolium repens* L., Fabaceae, Fabales (Abeynayake *et al.* 2012), and *Cymbidium hybrid*, Orchidaceae, Asparagales (Wang *et al.* 2014)). These copigments are also found in pink, purple and red flowers, although the color is attributed to the presence of anthocyanins, *e.g. I. tricolor* (Park *et al.* 2014), *P. hybrida* (Saito *et al.* 2006), *Trifolium repens* (Abeynayake *et al.* 2012), *Cymbidium hybrid* (Wang *et al.* 2014), *Dendrobium* sp, Orchidaceae, Asparagales (Willians *et al.* 2002), *Rosa rugosa* Thunb., *R. multiflora* Thunb., and *Prunus persica* (L.) Batsch Rosaceae, Rosales, *Dianthus caryophyllus* L., Caryophyllaceae, Caryophyllales, *Camellia japonica* L., Theaceae, Ericales (Luo *et al.* 2016), *Rhododendron* sp, Ericaceae, Ericales (Du *et al.* 2017).

Ferulic, caffeic and p-coumaric acids derivatives are the phenolic acids commonly found in flower tissues, while, kaempferol (acylated glycoside, diglycoside, glycoside, glucuronide), quercetin (acylated glycoside, diglycoside, glycoside, glucuronide), myricetin (acylated glycoside, diglycoside, glycoside), and isorhamnetin derivatives (glycoside) are the most abundant flavonols (Mulinacci et al. 2000, Saito et al. 2006, Abeynayake et al. 2012, Carocho et al. 2014, Wang et al. 2014, Snoussi et al. 2016). In pink flowers, the anthocyanins that have been identified are cyanidin (acylated glycoside, diglycoside, glycoside), perlargonidin (diglucoside), peonidin (acylated triglycoside, acylated glycoside, diglycoside, glycoside), petunidin (acylated diglycoside) and malvidin (acylated diglycoside) (Bobbio et al. 1985, Terahara et al. 1993, Willians 2004, Saito et al. 2006, Schmitzer et al. 2010, Wang et al. 2014, Gao et al. 2016, Hendra & Keller 2016). Pigments identified in T. pulchra were mainly the same as those mentioned above, moreover, an inedited substance, kaempferol 4'-O-(2"methyl)- $\alpha$ -D-glucoside was characterized for the first time. The kaempferol p-coumaroylhexoside (30) accumulation profile accompanied that observed for both identified anthocyanins that exhibited significant increment from white to pink floral stages, suggesting that this flavonol may act as an important copigment in pink flowers.

Gene expression analysis in *T. pulchra* flowers was a challenging enterprise because not only the genome is not sequenced, but there is no genomic sequence available for phylogenetic related species, being the most closely related species, *E. grandis*, from a different family, Myrtaceae. However, the approach used here was successful and allowed the transcriptional profiling of five key genes of flavonoid biosynthesis. Although five clones were sequenced for each gene, unique sequences were obtained. However, we cannot rule out that paralog genes, or even allelic variants, are missing, keeping in mind that we didn't use systematically bred germplasm and *T. pulchra* genomic complement is unknown.

Despite of the mentioned reservations, our results showed robustness and clear correlations between pigment and gene expression profiles from both, early and late biosynthetic genes (Figure 3.6). *PAL* expression increases in S2 stage, providing precursors for the increment in phenolic acids (S2) and the further accumulation of anthocyanins (S3 and S4). Downstream, the *CHS* is upregulated in S3 stage, directly related with anthocyanin detection. Interestingly, *FLS* was the only enzyme-encoding gene that showed reduction in the expression profile from S1 onwards in accordance with the decreasing trend observed in flavonol contents, which was the most abundant pigment along all flower development. The late biosynthetic gene *ANS* showed a pronounced increment in mRNA level from white (S1 and S2) to pink (S3 and S4) stages confirming that the expression of this gene is responsible for the anthocyanin accumulation in *T. pulchra* flowers.

The obtained results are in clear agreement with those reported for *Viola cornuta* L. (Violaceae, Malpighiales), whose flowers open white turning to lavender and further to purple (Farzad *et al.* 2003). By using a similar approach to that adopted in the present work, the authors have found that the expression of *CHS* and *DFR* increase along flower development. Moreover, *ANS* is strongly upregulated in colored stages (Farzad *et al.* 2003). Furthermore, a study in *Nicotiana mutabilis* flowers has found that the color change from white to red, passing through different shades of pink, is determined by anthocyanin accumulation (*i.e.* malvidin) and correlated with *CHS* expression (Macnish *et al.* 2010).

Recently, a transcriptomic analysis in *Paeonia ostii* T.Hong & J.X.Zhang (Paeoniaceae, Saxifragales) aimed to understand the mechanism underneath the flower color change from white to pink and demonstrated a positive correlation between the concentration of anthocyanins and the expression of biosynthestic genes. *CHI*, *F3H*, *F3'H*, *DFR* and *ANS* exhibited higher levels of mRNA in pink flowers than in white ones. Interestingly, the genes encoding the

MBW transcription factor complex did not showed differential expression during flower color change, instead, the expression of a MYB transcriptional factor encoding gene (*PoMYB2*) showed an inverse pattern compared to that observed for anthocyanin accumulation (Gao *et al.* 2016). Transcriptional repressors of anthocyanin accumulation have also been suggested in *P. hybrida*, *Z. mays*, *A. thaliana*, and *Fragaria ananassa* Duchesne ex Rozier (Rosaceae, Rosales) (Aharoni *et al.* 2001, Albert *et al.* 2011, Petroni & Tonelli 2011). Additionally, other metabolic pathways have shown to influence flower color change in *P. ostii*, such as carbohydrate and fatty acid metabolism, as well as ethylene signaling (Gao *et al.* 2016).

In vegetative tissues, the regulation of flavonoid biosynthesis is intimately associated with environmental changes to enhance plant survival under stressful environmental conditions. Various factors, such as UV, visible light, cold, osmotic stress, pollution and pathogen infection, can induce flavonoid biosynthesis. In particular, the role of phytochromemediated signaling transduction pathway, associated to temperature and light perception, has been abundantly explored (Guo et al. 2008, Maier et al. 2013, Maier & Hoecker 2015, Li et al. 2016, Kim et al. 2017). In this sense, various phytochrome signaling components in A. thaliana, such as the PHYTOCROME INTERACTING FACTORS (PIFs) and PROTEIN LONG HYPOCOTYL 5 (HY5), have shown to act as positive regulators of anthocyanin biosynthesis (Shin et al. 2007, Liu et al. 2015, Kim et al. 2017). In other plants this relationship between light perception and anthocyanin accumulation has also been addressed (Czemmel et al. 2012, Zhang et al. 2016, Zong et al. 2017). In T. pulchra we performed a pilot experiment to evaluate the role of light in flower pigment accumulation. When the plants were maintained indoor under low light irradiance, the flowers did not homogenously turn from white to pink and fell down in 24 hours (Supplemental Figure 3.5). This was indicative that, at least, the transcriptional regulation demonstrated for ANS might be light-dependent. Similarly, in cotton flowers (Gossypium sp., Malvaceae, Malvales), which turn from white to pink, gene expression analysis has shown that light and shade regulate anthocyanin biosynthesis, in particular ANS (Tan et al. 2013). Likewise, V. cornuta flowers did not undergo color change in the darkness (Farzad et al. 2003).

Finally, as discussed in the introduction, flavonoid-metal complexing and pH alterations influence flower color. Here, we have identified an increment in Fe<sup>3+</sup> concentration accompanying the transition from pink to dark pink stage (Figure 3.6), thus an effect of this ion in the color change seems likely. Whether Fe<sup>3+</sup> is involved in color change by pigment complexation or pH alteration remains to be further investigated.



**Figure 3.6.** Comprehensive schematic representation of the obtained results. Flavonoid biosynthetic pathway showing heatmaps representing pigment content (orange to green) and mRNA expression profile (blue to red). Dotted lines represent more than one enzymatic reaction. Quantified compounds are highlighted in green. Asterisks indicate statistically significant different values compare to S1 stage. Abreviations: *PHENYLALANINE AMMONIUM LYASE (PAL), CINAMMATE 4-HYDROXYLASE (C4H), CHALCONE SYNTHASE (CHS), FLAVONOL SYNTHASE (FLS)* and *ANTHOCYANIDIN SYNTHASE (ANS)* genes, S1- buds (day 0), S2- white flowers (day 1), S3- light pink (day 2), S4- dark pink (day 3).

## 5. CONCLUSION

In this work, by applying a combined approach of detailed biochemical and molecular genetic techniques, we characterized the pigment profile and investigated the mechanism that determines *T. pulchra* flower color change. The thirty phenolic compounds identified in chapter 2 were quantified showing variations in pigment and copigment contents along flower development. The partial cloning of five key genes of flavonoid biosynthetic pathway allowed to profile mRNA levels that explained, at least in part, the pigment accumulation pattern. Collectively these data indicate that the flower color change in *T. pulchra* is regulated by the transcriptional control of the structural genes of the flavonoid biosynthetic pathway that determine the accumulation of petunidin and malvidin, as well as, the copigment kaempferol *p*-coumaroylhexoside. Additionally, the quantification of metal contents suggested that Fe<sup>3+</sup> might influence the saturation of the color at dark pink stage.

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## 7. SUPPLEMENTAL MATERIAL

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**Supplemental Figure 3.1.** Alignments used for primer design to clone the gene partial sequences of *PHENYLALANINE AMMONIUM LYASE (PAL)* (A), *CINAMMATE 4-HYDROXYLASE (C4H)* (B), *CHALCONE SYNTHASE* (*CHS)* (C), *FLAVONOL SYNTHASE (FLS)* (D) and *ANTHOCYANIDIN SYNTHASE (ANS)* genes. Boxes indicates the primer region.

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Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-	ATGGCTCGTCTTACC	TCCTTCGTTTA	TGCAACTCTC	CTTGCATCCA	COCTCATTAT	TCTCGCCAAC CCTGGAGAAG	ACCCTCCTGG	CCCGCTTGTC GC	CATTCAACTA	CTTTCTGGT	CGGCCTCTCT	TGTCGTCTCT GGCCATCGCG	CTTOCTCTOC GTCTCCAAGC	TCACTTGCTT TCCGG	IGTTC
Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-	160 1   <b>TACCTATTTTTCGAC</b> GC	170 CGTGGCGCCAA AAGCGGTTCCG		190 GCCCTCTGG GCCCCCTCTGG	CCATCCCCAT	210 TTCGGCAPC CTTCGGCAPC	220 IGGCTCCAGG	230 TOGOCAATGA TCOGOCGACGA		250   . COCCTTCTAG CGCAACCTCA	260 1	270 CAAGACTTAC CAAGAOGTTC	280 	290 TCCTOCTTA TCCTOCTCCC	300 AACTC 3CATG
Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-	GGCGTGAAGAAGCTA GGGCAGCGCAACCTC	STCGTGGTGTGTC GTGGTCGTCTC	CGACCCTGAG	CTGACCACAC	ACGTCCTCCA	TACCEAAGEC	TCGAATTCG	GATCCCGCCC	TAGAAACGTT	STGTTCGACA STCTTCGACA	TATTCACGGG	CANTEGECAN	GACATGGTGT GACATGGTGT	TCACCATCTA TCACGGTGTA	ATGGC ACGGG
Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-	460 GATCATTGGCGCAAA GAGCACTGGCGCAAG	470 ATGCGTCGCAT	480	490 CCATTCTTTA CCCTTCTTCA	500 CCAACAAGGT	510 GGTGAACACT GGTGCAGCAG	520 IACAGTGATA IACCGGGAGG	530 TGTGGGAACA GGTGGGAGGO	540 GGAAATGGAT CGAGGCCGCC	550 TIGGTIGTCA SCCGTCGTCG	560 GCEACCTAAC AGGACGTCAA	570 	580 GATGAGO AACCCCG	590 GAGTGCGAAG	600 J JCGAA JCGAG
Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-	610 GGGATCGTTATCAGG GGGATCGTGCTCCGG	620 AAGCGTCTGCA CGCAGGCTGCA	630 IGATGATGCTC GATGATGATG	640 TACAACATCA TACAACAACA	650 TGTATCGAAT TGTACCGGAT	660 GATGTTCGAC	670 ICCAAGTTTG	680 AGTCCATGCA ACAGCGAGGA	690 GGATCCGTTG CGACCCGCTC	700 ITCATTGAAG	710 CCACCAAGTT TCAAGGCCCT	720 CAACTCCGAA CAACGGCCGAG	730 AGGAGTCGC- CGGAGCCGC-	740 	750 NAAGT NGAGC
Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-	760 TTOGACTACAATTAC TTOGACTACAACTAC	770 GOGGATTICAT GOCGACTICAT	780 TCCCATCCTC CCCCATCCTC	790 11. AGACCTCTCT AGGCCCTTCC	800 TGAAAGGTTA TGCGAGGCTA	810 CTTGAACAAG CCTGAAGATC	820 IGCAGGGACT IGCAAGGAGG	830   . TGCAGCGCAG TCAAGGAGCG	840 GCGCCTTGCC CAGGCTGCAG	850 ITTTTTAACA	860 	870 	880   . Aggaaaataa Aaaaaactgg	890 	900 1 NT GCCGG
Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-	910 	920 ATAAGTTGTGC CTGAAGTGCGC	930 CATOGACTACI	940 	950 	960 	970 IIIII AGCGAAGCCA AACGAGGACA	980 	990 CATCOTOGAG CATCOTOGAG	1000 AACATTAATG AACATCAACG	1010 TGGCAGCGAT	1020 AGAACAACC CGAGACGACA	1030 CTITGGTCCA CTCTGGTCCA	1040 TGGAATGGGG TTGAGTGGGG	1050 CATA SAATC
Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-	1060 GCACAGCTGGTCAAT GCGCAGCTCGTGAAC	1070 CATCCGAATGT CACCCCCAGAT	1080 CCAACGCAAA CCAGAGGAAG	1090 ATCAGGGAGG CTGCGGGAGG	AGATCTCCCC AGATCGACCC	1110 TGTCCTGAAA GGTCCTGGGC	1120 GGGGAGC CCTGGCGTGC	1130 CGGTCACGGA CAGTCACCGA	1140 ATCGAACCTG GCCCGAGCTC	1150 CACGAGTTAC FACAAGCTGC	1160 CCTATCTGCA CGTACCTCCA	1170 AGCAACAGTG GGCGGTGATC	1180 AAGGAAACAC AAGGAGACCC	TGAGGCTCCC	1200 ACACG SGATG
Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-		1220 GTGCCCCACAT GTGCCCCCACAT	1230 GAACCTGGAA	1240 GAAGCCAAGC GACGCCAAGC	1250 TAGGCGGCTA TCGGGGGGCTA	1260 CACCATCCCT. CGACATCCCO	1270 NAAGAGTCAA SCCGAGAGCA	1280 AGGTGGTGGT AGATCCTGGT	1290 GAATGCGTGG CAACGCGTGG	1300 IGGTTGGCCA	1310 ACAACCCTGA	1320 ATGGTGGAAG CCACTGGAAG	1330 GACGCGGAAG	1340 ACTTTAGGCC AGTTCCGGCC	1350 CAGAG
Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-	1360 AGGTTCTTGGAAGAG CGGTTCCTGGAGGAG	1370 CAAAGCGGGAA GAGGCGAAGGT	1380 CAGATGCAGTO	1390 GCGGGCGGGA GGGAAC	1400 AGGTGGACTT	1410 CAGGTACTTG CCGGTACCTC	1420 CCATTTGGGG CCCTTCGGAG	1430 TGGGGAGGAG TCGGCCGGAG	1440 GAGCTGCCCG GAGCTGCCCT	1450 SGCATTATTC SGGATCATCC	1460 TCGCCCTTCC TGGCCCTGCC	1470 GATATTGGGG CATCCTCGGG	1480 CTTGTCGTCG GTCACCATCG	1490 CCAAGTTGGT GCCAGTTGGT	1500 IGTCC IGCAG
Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-	1510 	1520 GTCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCC	1530   . IAATG IACAA	1540 GACAJGCTCG TCGAJGCTCG	1550 ACACGAGCGA ACACCACTGA	1560 CAAAGGAGGA GAAGGGTGGC	1570	1580 TGCATATCGO TGCACATATT	1590 CAACCATTCT GAAGCACTCC	1600 ACTGTCGTCT ACCATCGTCT	1610 TTGAGCCCGT TGAAGCCAAG	1620 TGCTCCCCAG ATCCTTTTGA	1630   . GAGATT	1640	1650
Eucgr.C00065.1.egr.23569548/1- Eucgr.J01844.1.egr.23599860/1-	1660   TGA			3 <b>-</b>			- 5								

Supplemental Figure 3.1. (Continued)

С	310	320	330	340	350	360	370	380	390	400	410	420	430	440	450
Eucar D01632 1	GACACCATCCCAAA	CTCGGCGAGG	AGGCGGCGT	GAAGGCCATC	AAGGAATGGG	SCCAGCCCA	AGTCCAAGATC	ACCCACATA	GTCTTCTGCTC	CACCTCTGGC	GTCGACATG	CTGGTGCCG	CTACCAGCT	ATGCGCCTTC	TOGGC
Eucgr.D01635.1	GACACCATCCCAAA	SCTCGGTGAG	AGGCGGCGT	GAAGGCCATC	AAGGAATGGG	GCCAGCCCA	AGTCCAAGATC	ACCCACATA	GTCTTCTGCTC	GACCTCTGGC	GTCGACATG	CCTGGCGCCG	CTACCAGCT	ATGCGCCTTC	TOGGC
Eucgr.H00087.1	TCGGAAGTGCCGAAG	SCTGGCGAAGO	AGGCGGCGGG	GAACGCCCTG	AAGGAATGGG	GCAGCCCA	AGTCGAAGATC	ACCCACCTG	GTGTTCTGCAG	CACCAGCGGT	GTCGACATG	CCGGGCTGCG	CTACCAGCTO	ACCAAGCTCC	TOGGC
Eucgr.H02828.1	GTGGAAGTCCCGAA	SCTCGGCAAGO	AGGCGGCCG	CAAGGCCATC	AAGGAGTGGG	GCCAGCCCA	AGTCCAAGGTC	ACCCACCTG	ATCTTCTGCAG	CACCAGCGGG	GTCGACATG	CCCGCCCCG	ACTACCAGCT	ACCAAGCTGO	TOGGC
Eucgr.H03914.1	CCAGAAGTGCCGAAG	GCTGGCGAAGG	AGGCGGCGG	GAACGCCCTC	AAGGAATGGG	GCAGCCCA	AGTCGAAGATC	ACCTACCTG	GTGTTCTGCAC	CACCAGCGGT	GTCGACATG	CCGGGCTGCG	CTACCAGCT	CACCAAGCTCC	TCGGC
Eucgr.K01974.1	GTCAGCATCCCGAA	SCTCGGCGAGO	BAGGCAGCGAG	GGAGGCCATC	CAAGGAATGGG	GCCAACCCA	AGTCCAAGATC	ACCCACATT	GTATTCTGCTC	CCTCTCCGGG	GCCGACATG	CTGGTGCTG	ACTACCAGCTO	ATGCGCCTCC	TOGGC
Eucgr.K01978.1	GACGCCATCCCGAA	SCTCGGCAAGO	GAGGCGGCAT	GAAGGCCATC	CAAGGAATGGG	SCCAACCCA	AGTCCAAGATC	ACCCACATT	GTCTTCTGCTC	CTTCTCCGGG	GTCGACATG	CCTGGCGCCG	ACTACCAGCTO	ATGCGCCTTC	CTOGGT
	100	470	100	100	500		500	5.20	5.40	550	5.50	670	500	500	<u> </u>
	460	470	480	490	500	510	520		540			570	580	590	
Eucgr. D01632.1	CTCAGCCCCTCCGTC	CAAGCGTGTCA	TGCTCTACC	CCAGGGCTGC	TTTGCTGGTG	SCACCGTCC'	TCCGTGTTGCC	AAGGACCTT	GCTGAGAACA	TGGCGGCGCT	CGCGTCCTC	STCGTCTGCTC	TGAGATCAC	GTCACCACCT	TCCGA
Eucgr.D01635.1	CTCAGCCCCTCCGTC	CAAGCGGGGTCA	TGCTCTACC	CCAGGGCTGC	TTTGCTGGTG	SCACCGTCC'	TCCGTGTTGCC	AAGGACCTT	GCCGAGAACAA	TGCCGGCGCT	CGCGTCCTC	STCGTCTGCT	TGAGATCAC	GTCACCACCT	TCCGA
Eucgr.H00087.1	CTCCGCCCCACCGT	CAAGCGGTACA	TGATGTACCA	GGTGGGCTGC	TACGGCGCTG	SCCTGGTCC	TCCGCATCGCC	AAGGATCTT	GCTGAGAACAA	CAAGGGGGGCT	CGCGTCCTC	STCGTGTGCT	CGAGATCAC	GTCTGCACCT	TCCGC
Eucgr.H02828.1	CTCCGCCCCTCCGT	CAAGCGGTACA	TGATGTACCA	GCAGGGCTGC	TTCGCCGGCG	GCACGGTCC	TCCGCCTGGCC	AAGGACCTG	GCCGAGAACAA	CAAGGGGGGCC	CGCGTCCTC	STCGTGTGCT	CCGAGATCAC	GCCGTCACCT	TCCGC
Eucgr.H03914.1	CTCCGCCCCTCCGT	CAAGCGGTACA	ATGATGTACCI	GGTGGGCTGC	TACGGCGCTG	SCCTGGTCC'	<b>FCCGCCTAGC</b>	AAGGATCTT	GCTGAGAACAA	CAAGGGGGGCT	CGCGTCCTC	STCGTGTGCT	CCGAGATCACO	CGTCTGCACTI	TTCCGC
Eucgr.K01974.1	CTCAGCCCCTCCGT	CAAGCGIGTCA	ATGCTCTACCI	CCTGGGCTGC	CTTTGCAGGCG	SCACTGCCC'	FCCGTATCGCC	AAGGACCTT	GCTGAGAACAA	TGCTGGTGCG	CGCGTCCTC	ATTGTCTGCT	CTGAGATCAC	TGTCGCGTCCT	TCCGA
Eucgr.K01978.1	CTGAGCCCCTCCGT	CAAGCGTGTCA	ATGCTCTACCA	CCTGGGCTGC	CTTTGCTGGCG	STACTGCCC	rccgcatcgca	AAGGACCTC	GCCGAGAACAA	TGCCAGTGCC	CGTGTCCTCO	STCGTCTGCT	TGATATCAC	TGTCACCGCCI	TCCGA
	610	→ 3'	620	640	650	660	670	600	600	700	710	720	720	740	750
									1						1
Eucgr.D01632.1	GGCCCGTCTGCTGA	SAACCTTGCCA	ACCTTGTCG	SCCAGGCCATC	TTTGGGGACG	SAGCGGCTG	CGCTCATCATA	GGGTCCGAC	CCAGAGATCCO	TGCTGAGCGG	CCCATATTCO	AACTCGTGTG	GACGGCGGA	ACCGTTTTGC	CAAAC
Eucgr.D01635.1	GGCCCGTCTGCTGA	GAACCTTGCCA	ACCTTGTCG	SCCAGGCCATA	ATTTGGGGACG	GAGCGGCTG	CGCTCATCATZ	GGGTCCGAT	CCAGAGATCCO	TGCTGAGCGG	CCCATATTCO	CAACTCGTGTG	GACGGCGGA	BACCGTTTTGC	CAAAC
Eucgr.H00087.1	GGCCCCACCGAGACO	CTACCTGGACA	ACCTCGTGGG	SCCAGGCTCT	STTCGGCGATG	GCGCCTCCT	CGGTCATCGTC	GGCTCGGAC	CCCGTCCCGG	GATCGAGAG	CCCCTGTAC	GAGATCGTCT	CACCTCCCA	SACTATCCTCC	CAAAC
Eucgr.H02828.1	GGCCCTACCGACAC	CCACCTCGACA	GCCTCGTGGG	GCAGGCCCTC	STTCGGGGGACG	GCGCGGCGGG	CCGTCATCGTC	GGGGCCGAC	CCGGTGCCCGG	CGTCGAGAAG	CCGATGTTC	SAGCTCGTCT	GACGGCCCA	SACCATTCTCC	CGGAC
Eucgr. H03914.1	GGCCCCTCCGAGGCC	TTCCTGGACA	ACCTCGTGGG	GCCAGGCCCTG	GTTCGGCGACG	GCGCCTCCT	CGGTCATCGTC	GGCGCGGGAC	CCCGTCCCGGG	GATCGAGAGG	CCCCTGTAC	SAGATCGTCTC	CACGTTCCA	SACTCTCCTCC	CAAAC
Eucgr K01974.1	GGCCCGTCCGCTGA	AACCATGGCA	ACCTTGTCG	CCAGGCCATC	TTTGGGGACG	SAGCGGCCGC	GCTCATCAT	GGGTCGGAC	CCAGAGATCCC	TGCTGAGCAG	CCCGTATTC	AACTCGTGTGTG	GACTGCGGA	ACCGTTTTGC	CAAAC
Buogrinorproir	ooccorrectoria			ocnoocani c				0001000000	ounononicou	1001010000	0000111110		one recount		- Children
	760	770	780	790	800	810	820	830	840	850	860	870	880	890	900
									• • • •   • • • •						
Eucgr.D01632.1	TCCGAGGGCGCCAT	GAGGGTCAAC	TGCGCGACG	TGGCCTCACC	CTTCGGGTTAT	CCGACAGGG	FCCCCCATTTO	ATCGCAGAC	CACATCGAGA	GAGCTTGAAA	GAGGCGTTC	SATTCGCTCG	CATAAGTGA	TGGAACTCCA	TCTTT
Eucgr. D01635.1	TCCGAGGGCGCCATC	GAGGGTCAAC	TGCGCGACG	TGGCCTCACC	TTCGGGTTAT	CCGACAGGG	PCCCCCATTTC	ATCGCAGAC	CACATCGAGA	GAGCTTGAAA	GAGGCGTTC	SATTCGCTCG	CATAAGTGAG	TGGAACTCCA	TCTTT
Eucgr H02828 1	AGCGAGGGGGGGCCAT	GACGGGCACC	TCCGGGAGAGA	GGGGCTCACC	STTCCACCTAC	TCAAGGACG	reccesseere	ATCTCGAAG	AACATCGAGA	GAGCCTGGTG	GAGGCGTTC	ACCCACTOG	CATCTCGGA	TGGAACTCCC	TCTTC
Eucgr. H03914.1	AGCGAGGAGGCCAT	GCTGGGCACO	TCCGGGAGA	CGGGCTCACC	ATCCACCTCC'	TCAAGGACT	FGCCGGCCCTC	GTCTCCAAG	AACATCGACA	GAACATGGCT	GAGGCGTTC	ACCCACTGG	CATCACCGA	TGGAACTCGC	TCTTC
Eucgr.K01974.1	TCAGAGAGCACCGT	GAGAGCCAAG	TGCGTGACG	GGGGGCTCACC	TTAGTTTTAT	CCGACAGGG	TCCCCCTATTO	ATCGCAAGC	AACATCGAGA	GAGCTTGAAT	GAGGCGTTC	ATTCTCTCG	CATAACCGA	TGGAAC-CCA	TCTTC
Eucgr.K01978.1	TCTGAGTGCGCCAT	GAGGGGCAAC	TGCGCGACG	TGGCTTCACC	TTCGGGTTAT	CCGACAGGG	TCCCGCATTTO	ATCGCGAGC	AACATCGAGA	GAGCTTGAAA	GAGGCGTTC	ATTCGCTTG	CATAAGTGA	TGGAACTCCA	TCTTT
	910	920	930	940	950	960	970	980	990	1000	1010	1020	1030	1040	1050
Fucar D01632 1	TGGGTGCCTCATCC	GGTGGCCCCG	CCATACTCA	TCADATCCAG	GCCAAGCTCC	SATTGAAAA	GGAGAGGCT	CGAGCCGTC	AGGCACGTCC	CAGAGAGTAC	GGGAACATC	TCGAGCGCGT	CGTCTCCTT	ATCCTCGATO	AGATG
Eucar, D01635.1	TGGGTGCCTCATCC	GGTGGCCCGG	CCATACTCA	TCAAATCGAG	GCCAAGCTGG	GATTGAAAA	AGGAGAGGGCT	CGAGCCGTG	AGGCACGTCCT	CAGAGAGTAG	GGGAACATG	CGAGCGCGT	CGTCTCGTT	ATCCTCGATO	AGATG
Eucgr. H00087.1	TTCATCGTGCACCC	GGCGGGCCCG	CGATCCTCG	CGACGTCGAG	GCCAAGTTGA	AGCTCAAGCO	CCGAGAAGATO	GAGGCTTCC	AGGTATGTGCT	CGGCGAGTAT	GGCAACATG	TCTAGCGCAT	GCGTCTTCTTC	ATAATGGACG	AGATG
Eucgr.H02828.1	TGGATCGCGCACCC	Gecegecce	CCATCCTGG/	CCAGGTCGAG	GAGAAGCTGA	AGCTCAAGG	CCGAGAAGCTO	CGCGCGACG	CGCCACGTGCT	GAGCGAGTAT	GGCAACATG	TCGAGCGCCT	CGTGCTGTTC	ATCCTGGACO	AGATG
Eucgr.H03914.1	TTCATAGTGCACCCO	CGCCGGGCCCC	GATCCTCG	CCAGGTCGAG	GCCAAGTTGA	AGCTCAAGCO	CCGAGAAGATT	GAGCTTTCC	AGGTATGTGCT	CGCCGAGTAT	GGCAACATG	TCTAGCGCAT	GCGTCTTCTTC	ATAATGGACG	AGATG
Eucgr.K01974.1	TGGGTGCCTCATCC	GGG			CCAAGCTAG	GGTTGAGAA	AGGAGAGGCTO	CGAGCGGTG	AGGCACGTGCT	TAGAGAGTAG	GGGAACATG	FCGAGCGCGT	TGTCTTCTTC	CGTCCTTGATO	AGATG
Eucgr.K01978.1	TGGGTGCCTCATCC	GGTGGCCCGG	SCCATACTCA	CCAAATCGAG	GCCAAGCTAG	GGCTGAAAAA	AGGAGAGGCT	CGAGCCGTG	AGGCAGGTGCT	TGGAGAGTAG	GGGAACATG	rcgagcgcaad	STGTCTTGTT	CATCCTCGATO	AGATG
	1000	1070	1000	1000	1100	1110	1100	1120	1110	1150	1100	1170	1100		
	1060	1070	1080	1090	1100		1120		1140			11/0			
Eucgr.D01632.1	CGAAGAAAGTCGGCC	GAGGAAGGG	TTCCCACCCC	CCCCCACCCC	TTGGAGTGGG	SCOTCCTCT	TCGGTTTTGGG	CCAGGCCTC	ACGGTGGAGAG	AGTGGTTCTT	CACACOCO				
Fuger DOIGE 1	and the second se		TIGCOMCCCC	COOGGAGGGGG				CCHOOCCICI			CACAGCGIG	CAACAGTGAG	CCATGGGTG	A	
Eucgr.D01655.1	CGAAGGAAGTCGGC	GAGGAAGGG	TTGCTACCC	CGGCGAGGGG	TTGGAGTGGG	GCGTCCTCT	TCGGTTTTGG	CCAGGCCTC	ACGGTGGAGAG	GGTGGTTCTT	CACAGCGTG	CAACAGTGAG	CCATGGGTG	4	
Eucgr. H00087.1	CGAAGGAAGTCGGCG	GGAGGAAGGGC CAAGAAAGGGC	TTGCTACCCO	CGGCGAGGGG	TTGGAGTGGG CTCGAATGGG	SCGTCCTCT' SCGTGCTCT'	ICGGTTTTGGG ICGGATTTGGI	CCAGGCCTC	ACGGTGGAGAG ACCGTCGAGAG	GGTGGTTCTT CGTCGTCCTC	CACAGCGTG	CCAACAGTGAG CCAACAGTGAG ICTGCTTAA	CCATGGGTG	A -	
Eucgr.H00087.1 Eucgr.H02828.1	CGAAGGAAGTCGGCC AGGAAGAAATCGATC AGGAAGAAGTCGACC	GAGGAAGGGC CAAGAAAGGGC CGAGGACGGGC	TTGCTACCCO TCAAGACCAO	CGGC GAGGGG TGGAGAGGGG CGGC GAGGGG	STTGGAGTGGG SCTCGAATGGG SCTCGAGTGGG	SCGTCCTCT' SCGTGCTCT' SCGTGTTGT'	rcggttttgg rcggatttgg rcgggttcgg	CCAGGCCTC CCCGGCCTC CCCGGGCTC	ACGGTGGAGAG ACCGTCGAGAG ACCGTCGAGAG	GGTGGTTCTT CGTCGTCCTC CGTGGTCCTC	CACAGCGTG CACAGCGTG CACGCCCTG CACAGCCTC	CCAACAGTGAG CCAACAGTGAG ICTGCTTAA SCGGCGCAGGG	CCATGGGTG CCATGGGTG CTTAA	4 - -	
Eucgr.H00087.1 Eucgr.H02828.1 Eucgr.H03914.1	CGAAGGAAGTCGGCC AGGAAGAAATCGATC AGGAAGAAGTCGACC AGGAAGAAATCGATC	GGAGGAAGGGC CAAGAAAGGGC CGAGGACGGGC CAAGAAAGGGC	TTGCTACCCC TCAAGACCAC TGAAGACCAC TCAAGACCAC	CGGC GAGGGG TGGA GAGGGG CGGC GAGGGG TGGA GAGGGG	STTGGAGTGGG SCTCGAATGGG SCTCGAGTGGG SCTCGAATGGG	SCGTCCTCT' SCGTGCTCT' SCGTGTTGT' SCGTGCTCT'	rcggttttgg rcggatttgg rcgggttcgg rcggatttgg	CCAGGCCTC CCCGGCCTC CCCGGGCTC	ACGGTGGAGAG ACCGTCGAGAG ACCGTCGAGAG ACCGTCGAGAG	GGTGGTTCTT CGTCGTCCTC CGTGGTCCTC CGTCGTCCTC	CACAGCGTG CACGCCCTG CACGCCCTG CACGCCCTG	CCAACAGTGAG CCAACAGTGAG ICTGCTTAA BCGGCGCAGGG ICTACTTAG	CCATGGGTG CCATGGGTG	A A - -	
Eucgr.H00087.1 Eucgr.H02828.1 Eucgr.H03914.1 Eucgr.K01974.1	CGAAGGAAGTCGGCO AGGAAGAAATCGATO AGGAAGAAGTCGACO AGGAAGAAGTCGATO CGCAGGAAGTCGGTO	GGAGGAAGGGC CAAGAAAGGGC CGAGGACGGGC CAAGAAAGGGC CGAGGAAGGA	TTGCTACCCC TTCAAGACCAC TTGAAGACCAC TTCAAGACCAC TTCGCGACCAC	CGGC GAGGGG TGGA GAGGGG CGGC GAGGGG TGGA GAGGGG CGGC GAGGGG	STTGGAGTGGG GCTCGAATGGG GCTCGAGTGGG GCTCGAATGGG GTTGGAGTGGG	SCGTCCTCT SCGTGCTCT SCGTGTTGT SCGTGCTCT SCGTCCTTT	rcggttttggg rcggatttggt rcgggttcgg rcggatttggt rtgggtttgg	CCAGGCCTC CCCGGCCTC CCCGGCCTC CCCGGCCTC	ACGGTGGAGAG ACCGTCGAGAG ACCGTCGAGAG ACCGTCGAGAG ACGGTGGAGAG	GGTGGTTCTT CGTCGTCCTC CGTGGTCCTC CGTCGTCCTC GGTGGTGCTT	CACAGCGTG CACAGCCCTG CACAGCCCTG CACAGCCCTG CACAGCCCTG	CCAACAGTGAC CCAACAGTGAC FCTGCTTAA GCGGCGCAGGC FCTACTTAG CCAATAGTGCC	CCCATGGGTG CCCATGGGTG CTTAA	A - - - A	
Eucgr.H001835.1 Eucgr.H02828.1 Eucgr.H03914.1 Eucgr.K01974.1 Eucgr.K01978.1	CGAAGGAAGTCGGCO AGGAAGAAATCGATO AGGAAGAAGTCGACO AGGAAGAAATCGATO CGGAGGAAGTCGGTO CGGAGGAAGTCGGCO	GGAGGAAGGG CAAGAAAGGG CGAGGACGGG CAAGAAAGGG CGAGGAAGGA	TTGCTACCC TTGAAGACCA TGAAGACCA TGAAGACCA TCAAGACCA TCGCGACCA TCGCGACCA	CGGC GAGGG TGG2 GAGGG CGGC GAGGG TGG2 GAGGG CGGC GAGGG CGGC GAGGGG	STTGCAGTGGG GCTCCAATGGG GCTCCAGTGGG GCTCCAATGGG STTGCAGTGGG STTGCAGTGGG	SCGTCCTCT SCGTGCTCT SCGTGTTGT SCGTGCTCT SCGTCCTTT SCGTCCTCT	ICGGTTTTGGG ICGGATTTGGT ICGGGTTCGGG ICGGATTTGGT ITGGGTTTGGG ITGGGTTTGGG	CCAGGCCTC CCCGGCCTC CCCGGCCTC CCCGGCCTC CCCGGCCTC	ACGGTGGAGAG ACCGTCGAGAG ACCGTCGAGAG ACCGTCGAGAG ACGGTGGAGAG ACGGTGGAGAG	GGTGGTTCTT CGTCGTCCTC CGTGGTCCTC CGTCGTCCTC GGTGGTGCTT GGTGGTGCTT	CACAGCGTG CACGCCCTG CACAGCCTG CACAGCCTG CACAGCGTG CACAGCGTG	CCAACAGTGAC CCAACAGTGAC ICTGCTTAA SCGGCGCAGGC ICTACTTAG CCAATAGTGCC CCAACAGTGAC	CCCATGGGTG CCCATGGGTG CTTAA CCCCCGGATG CCCCCGGATG	4 - - 4	

Supplemental Figure 3.1. (Continued)

D	10	20	30	40	50	60	70	80	90	100	110	120	130	140	150
Eucgr.F03761.1 Eucgr.F03763.1 Eucgr.L00738.1 Eucgr.L00738.2	ATGGAAGTGGAGAGA ATGGAAGTGGAAAGA ATGGAGGTGGAGAGA ATGGAGGTGGAGAGA 3'	ATTCAAGCATT ATTCAGGCATT GTGCAGGCCGT <u>GTGCAGGC</u> CGT > 5'	GGCCG GGCCG GGCGTCTGCT GGCGTCTGCT	CGTTCGGCCTC CGGTCGGCCTC CTGATGACCGC CTGATGACCGC	G-ACACGCTC G-ACACGCTC GGAGACCATC GGAGACCATC	CCCAGCCCAG CCCAGCCCAG CCCCGCCGAT CCCCGCCGAT	STTTGTCCGCC STTTGTCCGCC TTCATCCGGC TTCATCCGGC	CCTCCCACGA CCTCCCACGA CGGAGCAGGA CGGAGCAGGA	GCGGCCAGAG GCAGCCAGAG GCGCCCCGCC GCGCCCCGCC	BAACAGC-CC BAACAGC-CC CATCACCACC CATCACCACC	CGGCCGTGGA- CGGCCGTGGA- TACCAGGGCC TACCAGGGCC	-GGGCGTCACC -GGGCGTCACC CCGGCCCCCGAC CCGGCCCCCGAC	GTGCCGGTGA GTGCCGGTGA ATCCCCACCA ATCCCCACCA	ICTCGCTATCC ICTCACTATCC TCGACCTGAGC TCGACCTGAGC	CAG CAG CTG CTG
Eucgr.F03761.1 Eucgr.F03763.1 Eucgr.L00738.1 Eucgr.L00738.2	160 CCCCACCACGCT CCCCACCACGCT CCTGACCGCGAGTCC CCTGACCGCGAGTCC	170 CTTATCCGGGA CTCATCCGGGA GTCGTGCGGGGC GTCGTGCGGGGC	180 	190 	200 CTGGGGATTC CTGGGGATTC GTGGGGGATC GTGGGGGGATC	210 CTTCCTTGTC CTTCCTTGTC CTTTCAGGTC CTTTCAGGTC	220 SACCGATCACG SACCGATCACG GTGAACCACG GTGAACCACG	230 GGATACCACC GGATACCGCT GCATCCCCAC GCATCCCCAC	240 GGAGCTGATC GGAGCTGATC GGAGCTCATC GGAGCTCATC	250 CGGGCAGCTO CGGGCAGCTO CTCCCGCCTO CTCCCGCCTO	260   GCAGGAGGCGG GCAGGAGGCGG GCAGGCGGCCG GCAGGCGGCCG	270 SGCCGCGAGTTC SGCCGTGAGTTC SGGAAGCACTTC	280 TTCAGGCTCC TTCAGGCTCC TTCGACCTCC TTCGACCTCC	290 CGCAGGAGGAG CGCAGGAGGAG CCCAGGAGGAG CCCAGGAGGAG	300   AAG AAG AAG
Eucgr.F03761.1 Eucgr.F03763.1 Eucgr.L00738.1 Eucgr.L00738.2	310 GAGAAGTACGCGAAC GAGAAGTACGCCAAC GAGGCCTACGCCAAG GAGGCCTACGCCAAG	320 GATCCGTCCAA GATCCGTCCAG CCCCCGGGGTC CCCCCGGGGTC	330   . AGGGAATTTC AGGGAATTTC CCAGTCCATC CCAGTCCATC	340    GAAGGGTACGG GAGGGCTACGG GAGGGCTACGG GAGGGCTACGG	350 GACGAAGATG GACGAAGATG CACCCGGGTC CACCCGGGTC	360 GACCAAGAAC GACCAAGAAC CTCCAAGGAG CTCCAAGGAG	370 CCATGACGAAA CCATGACGAAA GGTCAACGGCA GGTCAACGGCA	380   . AGGTTGAGTG AGGTTGAGTG AGCGGATGTG AGCGGATGTG	390 GGTGGATTAC GGTGGATTAC GAACGACCAC GAACGACCAC	400 CTTTTTTCCAC CTTTTTTTCCAC CCTCTTCCAC CCTCTTCCAC	410 TTCATGCATC CATCATGCATC CAAGGTCTGGC CAAGGTCTGGC	420 CCATCCGGCAAG CCATCTGGCAAG CCTCCTTCCTCC CCTCCTTCCTCC	430 GTCAGGCATG GTCAGGCATG GTCAACTACC GTCAACTACC	440 AAATCTGGCCT AAATTTGGCCT GCTTCTGGCCC GCTTCTGGCCC	450   'CAA 'CGA 'CGA 'ACG 'ACG
Eucgr.F03761.1 Eucgr.F03763.1 Eucgr.L00738.1 Eucgr.L00738.2	460 CACCCGCCTTCTTAC CACCCGCCTTCTTAC CACCCCCCGAACTAC AACCCCCCGAACTAC	470 	480 	490 	500 CACTGAAAGTG CACTGAAAGTG GAGGGAGGTG	510 SACAGAGGTO SACAGAGGTO GGCGGACAAG GGCGGACAAG	520 CTGCTGGAGC CTGCTGGAGC CTGTTCCTCT CTGTTCCTCT	530 TGATCTCGGA TGATCTCCGA GCCTCTCCCT GCCTCTCCCT	540 GGGCCTAGGZ GGGCCTAGGZ GGGGCTTGGT GGGGCTTGGT	550 ACTGGAGAAA ACTGGAGAAA TCTGGAGGAG TCTGGAGGAG	560    AAATGTCTTGA AATGTCTTGA SACCGCCGTGA SACCGCCGTGA	570 AAAGCAAGTTTG AAAGCAAGTTTG AAGGAGGCGATC AAGGAGGCGATC	580 GGAGGTGACC GGAGGTGACC GGCGGGGAAG GGCGGGGAAG	590 AGATGGAGCTA AGATGGAGCTA AGCTTGAGTAC AGCTTGAGTAC	600   IGAG IGAG ZAAC ZAAC
Eucgr.F03761.1 Eucgr.F03763.1 Eucgr.L00738.1 Eucgr.L00738.2	610    ATGAAAATAAACCTG ATGAAAATAAACCTG CTCAAGATCAACTAC CTCAAGATCAACTAC	620 TACCACCTG TACCCACCTG TACCCACCTG TACCCGCCCTG	630 	640 CAGCTGGCCCT CAGCTGGCCCT GACCTCGCCCT GACCTCGCCCT	650 CGGCGTCGAG CGGCGTCGAG GGGGGTGGCT GGGGGTGGCT	660 GCCTCACACC GCCTCACACC FGCCCACACC FGCCCACACC	670 CGACATGTCGG CGACATGTCGG CGACCTGTCCA CGACCTGTCCA	680 CAGTCACCAT CAGTCACCAT CGATCACCAT CGATCACCAT	690 ACTGGTCCCC ACTGGTCCCC CCTCGTGCCC CCTCGTGCCC	700 AACGATGTC AACGATGTC AACGATGTC AACGATGTC	710 CCCGGGTCTCC CCCGGGTCTCC CCCCGGCCTCC	720 AAGTTTGGAAA AAGTTTGGAAA AGGTCCTCAAA AGGTCCTCAAA	730 GACGGCAATT GATGGCAATT GACGGGCGCT GACGGGCGCT	740 GGTTCGCCGCA GGTTCGCCGCA GGATCGATGCC GGATCGATGCC	750  IGAT IGAT IGAT CAAG
Eucgr.F03761.1 Eucgr.F03763.1 Eucgr.L00738.1 Eucgr.L00738.2	760 TACTTGCCCGATGCT TACTTGCCCAATGCT TACTTGCCCAACGCC TACATCCCCAACGCC	770 CTTTTCTGCCA CTTTTCTTCCA CTCATCGTCCA CTCATCGTCCA	780 	790 	800 TCTAAGCAAC TCTAAGCAAC CCTGAGCAAT CCTGAGCAAT	810 CGGAAAGTAC CGGAAAGTAC TGGACGGTAT	820 CAAAAGTGTAC CAAAAGTGTAC TAAGGCAGTAC TAAGGCAGTAC	830 TTCATAGGAG TTCATAGGAG TGCACAGGAG TGCACAGGAG	840 	850 CAAGGAAAAG CAAGGAAAAG CAAGGAGAAAG CAAGGAGAAG	860 SACGCGAATGT SACGCGAATGT GCGAGGATGT GCCGAGGATGT GCCGAGGATGT 5'	870 CATGGGCAGTG CATGGGCAGTG CGTGGCCGGTG CGTGGCCGGTG	880 TTTGTTGCAC TTTGTTGCAC TTCTTGGAGC TTCTTGGAGC	890 CCCCACAAGAG CCCCGCAAGAG CGCCGGAAGAG CGCCGGAAGAG	900   IGCA IGCA TGG TGG
Eucgr.F03761.1 Eucgr.F03763.1 Eucgr.L00738.1 Eucgr.L00738.2	910 CTGATCGGACCTCTC CTGATCGGGCCCTCTC GTGGTCGGACCTCTG GTGGTCGGACCTCTG	920 CCGGAGCTCAT CCGGAGCTCAT CCTCAGCTGCT CCTCAGCTGCT	930 CAACGATCAG CAACGATCAG TAGCCCCGAG	940 AACCCGGTGAA AACCCGGTGAA AGTCCCCCGAA AGTCCCCCGAA	950 GTACTCCACC GTACTCCACC ATACAAGGCC ATACAAGGCC	960 CAAGACGTTT CAAGACGTTT CAAGAACGTTT CAAGAAGTTC CAAGAAGTTC	970 GCCGAGTTTC GCCGAGTTTC CAAGGATTACA CAAGGATTACA	980 GATACCGCAA GATACCGCAA TGTACTGCAA TGTACTGCAA	990 ATTCAACAAG ATTCAACAAG ACTCAACAAG ACTCAACAAG	1000 SCTCCCGCAC SCTCCCGCAC SCTTCCCCAC	STAG STAG STAA STAA				

Supplemental Figure 3.1. (Continued)

E	10	20	30	40	50	60	70	80	90	100
Eucgr.D01945.1.egr.23575751/1- GSVIVT01019892001.vvi.17829430	ATGGTGAGCGTTGT	GCTCCTACA	GTCGAGAGCT	IGTCCAGCAG1	IGGCATTCAG	ICGATCCCGCA ICAATCCCCCAA	GGAGTATOTO	COCCCCAA	AGGAGCTCM	CAAGCA
Eucgr.D01945.1.egr.23575751/1- GSVIVT01019892001.vvi.17829430	110 TTGGCGACATCTTC TTGGCAATGTCTTT	120 1 CAGGAGGAGA GAGGAGGAGA	130    AGAAGCATGA	140 SGGCCCTCAGE	TCCCGACCAT	160 III ICGACCTCGAG	170 GACATAGCG GATATTGAG	180 ICTAAAGACCO	190 COTOGTGAGO	200 II GGAGAG GGAGAG
Eucgr.D01945.1.egr.23575751/1- GSVIVT01019892001.vvi.17829430	210 GTGCCACGAGGAGC ATGCCGGGAGGAGT	220 II. TCAGGAAGGC TGAAGAAAGC	230 11 TGCCACCGAC TGCCATGGAG	240 Inggggcgtcat Inggggtgtgat	250 INCACCTORTO	260 CAACCATGGGA	270 TCCCCAACG	200 ACCTGATTGAG	290 CGTGTCAAG	300 1 1 NAGGCT GTTGCT
Eucgr.D01945.1.egr.23575751/1- GSVIVT01019892001.vvi.17829430	310 GGCGAGGTGTTCTT GGAGAGGCCTTTTT	320 11. CAACCTCCCG CAATCTCCCC	330 ATCGAGGAGAA ATCGAGGAAA	340 MGGAGAAGCAT MGGAGAAGTAT	350 GCCAACGACG GCTAATGACG	360     CAGGGCGCCGG CAGGCCTCCGG	370 GAAGATCCAG	380 SGGCTACGGGJ TGGCTATGGCJ	390 AGCAAGCTTGO	400 CAAACA CCAACA
Eucgr.D01945.1.egr.23575751/1- GSVIVT01019892001.vvi.17829430	410 ATGCCAGCGGGCAG ATGCTAGTGGACAG 3'	420 CTCGAGTGGG CTTGAGTGGG	430 AGGACTACTT	440 TTCCACCTCC	450 STTTACCCTG/	460    AGGACAAGCGT NAGACAAGCGC	470 GACTTGTCC/	480	490 AGCACCCGAGO AGACACCAAGO	500 SGATTA CGACTA

Supplemental Figure 3.1. (Continued)

Stage	<b>Biological replicates</b>	ng/µL	260/280	260/230
	1	875,70	2,11	2,30
	2	494,40	2,07	1,93
S1	3	629,80	2,15	1,78
	4	3.173,30	2,12	2,23
	5	418,70	2,10	2,21
	1	1.012,80	2,13	2,25
	2	847,70	2,10	2,01
S2	3	499,30	2,10	1,96
	4	1.039,80	2,13	2,32
	5	1.135,40	2,13	2,30
	1	888,20	2,11	2,20
	2	1.136,90	2,12	2,37
S3	3	890,30	2,13	2,36
	4	1.026,60	2,12	2,29
	5	702,70	2,16	2,21
	1	1.030,30	2,13	1,93
	2	534,20	2,15	2,08
S4	3	1.262,70	2,14	2,35
	4	1.565,40	2,12	2,31
	5	954,20	2,12	1,92



**Supplemental Figure 3.2.** RNA quantification (A) and integrity (B). Agarose gel (1%) for analysis of RNA integrity, approximately 500 µg of each sample was deposited. The numbers indicate the biological replicates for each stage (S1 to S4). S1- buds (day 0), S2- white flowers (day 1), S3- light pink (day 2), S4- dark pink (day 3).

Α
Gene Enzyme		Nucleotides number in partial sequence of <i>T. pulchra</i>	Ammino acids number in partial sequence of <i>T.</i> <i>pulchra</i>	Identity of ammino acids (%) with <i>E.</i> <i>grandis</i>	
PAL	PHENYLALANINE AMMONIA-LYASE	633	206	95	
C4H	CINNAMATE 4-HYDROXYLASE	1.073	357	87	
CHS	CHALCONE SYNTHASE	621	207	88	
FLS	FLAVONOL SYNTHASE	817	272	69	
ANS	ANTHOCYANIDIN SYNTHASE	361	120	83	
RPS	RIBOSOMAL PROTEIN S13	371	123	98	
<u>EF1</u>	ELONGATION FACTOR 1α	694	231	93	



**Supplemental Figure 3.3.** Identity of obtained with *E. grandis* sequences (A). Agarose gel (0.8%) of the cloning sequences (B). The abbreviations indicate: *PHENYLALANINE AMMONIUM LYASE (PAL), CINAMMATE 4-HYDROXYLASE (C4H), CHALCONE SYNTHASE (CHS), FLAVONOL SYNTHASE (FLS)* and *ANTHOCYANIDIN SYNTHASE (ANS), ELONGATION FACTOR 1-α (EF1)* e *RIBOSSOMOAL PROTEIN S13 (RPS)* genes.

#### >T.pulchra\_n\_PAL

#### >T.pulchra\_n\_C4H

#### >T.pulchra\_n\_CHS

**Supplemental Figure 3.4.** Amino acid (aa) and nucleotide (n) partial sequences of genes cloned from *Tibouchina pulchra*. *PHENYLALANINE AMMONIUM LYASE (PAL), CINAMMATE 4-HYDROXYLASE (C4H), CHALCONE SYNTHASE (CHS), FLAVONOL SYNTHASE (FLS), ANTHOCYANIDIN SYNTHASE (ANS), ELONGATION FACTOR 1-\alpha (<i>EF1*) e *RIBOSSOMOAL PROTEIN S13 (RPS*) genes, S1- buds (day 0), S2- white flowers (day 1), S3- light pink (day 2), S4- dark pink (day 3).

#### >T.pulchra\_n\_FLS

#### >T.pulchra\_n\_ANS

#### >T.pulchra\_n\_EF1

AACATCGTGGTCATCGGCCATGTCGACTCCGGAAAGTCGACCACCACTGGGCACTTGATCTACAAGCTCGGTGGGATCGACAA GCGTGTTATCGAGAGATTCGAGAAGGAAGCTGCTGAGATGAACAAGAGGTCTTTCAAGTATGCTTGGGTGCTCGACAAGCTGA AGGCCGAGCGCGGAGCGTGGTATCACCATTGATATTGCCCTGTGGAAGTTCGAGACCACCAAGTACTACTGCACCGTCATCGAC GCTCCCGGACATCGCGACTTTATCAAGAACATGATCACGGGAACTTCCCAGGCTGACTGTGCTGTCCTTATCATTGACTCCAC CACTGGTGGTTTCGAAGCTGGTATCTCCAAGGATGGACAGACCCGTGAGCACGCTCTTCTGGCTTTCACCCTTGGTGTCAGGC AAATGATTTGCTGCTGCCAACAAGATGGATGCCACCACCCCCAAGTACTCGAAGGCTAGGTACGACGAAATTGTCAAGGAAGTC TCTTCATATATGAAGAAGGTCGGATACAACCCCGAGAAGATCCCGTTTGTCCCCATTTCCGGATTCGAGGGTGACAATATGAT TGACAGATCCACCAACCTTGACTGGTACAAGGGCCCTACCTGTTGGATGCTCTTGACATGATTCAGAGGCCCAAGAGCCCCTC GGACAAGCCCCTCCGTCTTCCCCTTCAGAC

#### >T.pulchra\_n\_RPS

#### Supplemental Figure 4. (Continued)

## > T.pulchra\_aa\_PAL

GLASIVLFEAIMEHILEGSEIIKAAQKLHEMDPLQKPKQDRYALRTSPQWLGPQIEVIRAATKMIEREINSVNDNPLIDVSRN KALHGGNFQGTPIGVSMDNTRLAIASIGKLMFAQFPELVNDFYNNGLPSNLSGGRNPSLDYGFKGAEIAMASYCSELQFLANP VTNHVQSAEQHNQDVNSLGLISSRKTAEAVDVLKLMSSPS

## > T.pulchra\_aa\_C4H

RKMRRIMTVPFFTNKVVQQYRFGWEDEATKVVEDVKKNPESATNGIVLRRRLQLMMYNNMFRIMFDRRFESEEDPLFIRLRTL NGERSRLAQSFEYNYGDFIPILRPFLRGYLKVCKEVKEKRLKLFKDYFVDERKKLAGTKKVENDGLKCAIDHILDAQQKGEIN EDNVLYIVENINVAAIETTLWSIEWGIAELVNHPEIQRKLRDELDTVLGPGVQITEPDTYRLPYLQAVIKETLRLRMAIPLLV PHMNLHDAKLGGHDIPAESKILVNAWYLANNPSLWKNPEEFRPERFMEEEAKVEANGNDFRYLPFGVGRRSCPGIILALPILG ITIGRLVQNFEMLPPPGQSKLARYQ

## > T.pulchra\_aa\_CHS

RLMMYQQGCFAGGTVLRLAKDLAENNKGARVLVVCSEITAVTFRGPSESHLDSLVGQALFGDGAAAIIMGSDPVPGVERPMFE LVSAAQTILPDSDGAIDGHLREVGLTFHLLKDVPGLISKNIEKSLVEAFTPIGISDWNSIFWIAHPGGPAILDQVEEKMGLKP EKMRATRQVLSDYGNMSSACVLFILDEMRTNSAQNGMKTTG

## > T.pulchra\_aa\_FLS

VASSFISKGVIPAEYIRPEQEQPSVTTYHGPVPTIPTVDLGDADHEGLVRAVADASREWGMFQVFNHGIPAEVIAELQRVGKE FFELPAEEKEKYARPPGAQSLEGYGTRLQKNLDGKKSWVDHLFHKIWPPSSVDYKYWPANTPAYREANEEYAKHIRKVADKLF GCLSVGLGLEEDAMKKAVGGDDLLYNLKINYYPPCPRPDLALGVLPHTDLSALTLLVPNEIPGLQVFKDGKWIDAEYIPNALI VHIGDQVEILSNGEYRSVLHRTT

## > T.pulchra\_aa\_ANS

ILSIPKEYIRPQEELRSIGDVFEEEKKHEGPQVPTIDLQDIDSEDPVVRETCREELKKAATQWGVMHLVNHGIPNELIERVKK AGDEFFTLPVEEKEKYANDQGSGKIQGYGSKLANNAS

### > T.pulchra\_aa\_RPS

KGISAXALPYKRTSPSWLKISSQDVEENICKFAKKGLTPSQIGVILRDSHGIAQVRSVTGSKILRILKAHGLAPEIPEDLYHL IKKAVSIRKHLERNRKDKDSKFRLILVESRIHRLARYYKK

### > T.pulchra\_aa\_EF1

NIVVIGHVDSGKSTTTGHLIYKLGGIDKRVIERFEKEAAEMNKRSFKYAWVLDKLKAERERGITIDIALWKFETTKYYCTVID APGHRDFIKNMITGTSQADCAVLIIDSTTGGFEAGISKDGQTREHALLAFTLGVRQMICCCNKMDATTPKYSKARYDEIVKEV SSYMKKVGYNPEKIPFVPISGFEGDNMIDRSTNLDWYKGPTLLDALDMIQSPRGPRTSPSVFPFR

## Supplemental Figure 4. (Continued)



**Supplemental Figure 3.5.** *Tibouchina pulchra* flowers need light to turn from white to pink color. Plants were maintained indoor under low light irradiance. After 24 h (A to B) white flowers did not homogenously turn to pink (D) and fell down to following day (B to C). E. Normal pink flower.

Supplemental	Table 3.1.	. Standard	curves	parameters
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Supplemental Table 3.1. Standard curves parameters.			
Compound	R <sup>2</sup>	equation	
<i>p</i> -coumaric	0.99	y=12955x	
kaempferol	0.97	y=49285x	
cyanidin	0.99	y=364963x	

# **Supplemental Table 3.2.** Primers used for gene cloning and RT-qPCR.

Gene	Description	Forward	Melting temperature	Reverse	Melting temperature	Amplicon	Use
PAL	flavonoid biosynthesis	MTYGTSAAYRGCACBGSWGTYGGGTC	65.3°C	CSRTGCCTSAGGTCDAYBGCYTG	63.4°C	813	Cloning
C4H	flavonoid biosynthesis	CTGGCTCCAGGTSGGCRAYGACCTC	65.4°C	GKCCWCCYTTCTCRSTSGTGTCGAGC	64.3°C	1277	Cloning
CSH	flavonoid biosynthesis	GGYCTSMGCCCCWCCGTCAAG	65.0°C	CGCCCCAYTCSARCCCCTC	63.5°C	661	Cloning
FLS	flavonoid biosynthesis	GGARGTGGARAGARTKCARGC	58.1°C	GACATYCKCGYCTTKTCCTTGTTCAC	56.2°C	863	Cloning
ANS	flavonoid biosynthesis	GAGTYGAGAGCTTGTCSAGCAGTGG	62.7°C	GTAGTCCTCCCACTCRAGCTGSCC	64.7°C	385	Cloning
EF1α	reference gene	GGGTAARGARAAGGTTCACATC	54.9°C	CCRATACCACCAATCTTGASAC	53.7°C	740	Cloning
RPS13	reference gene	GTCGCATGCACAGYCGMGG	61.9°C	CCARACRGGWGGKAGCTTCTTGG	58.1°C	412	Cloning
PAL	flavonoid biosynthesis	GCACGAGATGGACCCGTTGC	61.3°C	GTGCCTTGGAAATTCCCGCC	59.6°C	158	RT-qPCR
C4H	flavonoid biosynthesis	GGCGAGATCAACGAGGACAACGTCC	63.2°C	GAGGTTCATGTGGGGCACGAGG	62.4°C	229	RT-qPCR
CSH	flavonoid biosynthesis	GAACAACAAGGGGGCTCGTGTCC	62.4°C	CCGAGTCAGGCAGGATGGTCTG	61.6°C	168	RT-qPCR
FLS	flavonoid biosynthesis	GCTGACCACGAGGGATTGGTGCG	64.5°C	CTTCTGGAGCCGAGTCCCGTAACC	64.5°C	166	RT-qPCR
ANS	flavonoid biosynthesis	CGGTTGTGCGTGAGACCTGCC	63.3°C	GCCTTGGATCTTCCCCGACCCC	64.3°C	150	RT-qPCR
EF1α	reference gene	GAGGAGCGTCACGGGAAGCAAG	62.3°C	GTAGTAGCGAGCCAGCCTGTGG	61.8°C	142	RT-qPCR
RPS13	reference gene	GGATGGACAGACCCGTGAGCACG	63.8°C	GGGACAAACGGGATCTTCTCGGGG	63.3°C	155	RT-qPCR

## **Final Considerations**

Floral color change along flower development is a phenomenom controlled by serveral factors. This work proposed to gain insights about the chemical and genetic mechanism that underneath *Tibouchina pulchra* (Melastomataceae) flower color change.

Among the thrirty detected compounds, five were described for the first time in Melastomataceae, fourteen in *Tibouchina* and sixteen in *T. pulchra*. Moreover, an inedited substance was described: kaempferol-(2''-O-methyl)-4'-O- $\alpha$ -D-glucopyranoside. Five key genes of the flavonoid biosynthetic pathway were successfully cloned, and their expression profiled. Our results exhibited a clear relation among color transition, pigmentation, copigmentation, gene expression, and metal content. The flower color change in *T. pulchra* is mostly the consequence of an increment in *ANS* expression, which in turn, results in the accumulation of petunidin and malvidin found in pink stages (S3 and S4). Only one copigment (kaempferol-*p*-coumaroylhexoside - **30**) appeared in the pink stages, while all the others were present in the white ones remaining all along flower development. Interestingly, Fe<sup>3+</sup> ion increased in dark pink stage (S4), suggesting its involvement in color saturation.

Futher analyzes are necessary to fully elucidate the mechanism of color change in *T. pulchra*. Regarding gene expression, the identification of the transcriptional factors that control flavonols/anthocyanins accumulation would bring light on the differential expression pattern displayed by the distinct analyzed genes. Additionally, the genetic analysis of the late reactions, such as methylations, glycosilations and acylations of flavonoid structures has been lagged behind, probably because of the ubiquity of these reactions, which difficults the identification of the specific enzyme encoding genes (*OMT*, *FGT* and *ACT*). Due to the demonstrated post-transcriptional regulation of some flavonoid biosynthetic genes, the study of the enzymatic activity would increase the knowledge about this phenomenon. Finally, the effect of Fe<sup>3+</sup> ion on flower color needs to be better understood by means addressing the subcellular localization, the formation of metal-flavonoid complexes and/or its influence on vacuolar pH.

Concluding, the generated information contributed, for the first time, to the understanding of the flower color change phenomena in a Brazilian native species and constitute a repository data for future studies with Melastomataceae.